

In-situ diffraction

beatriz.moreno@lightsource.ca

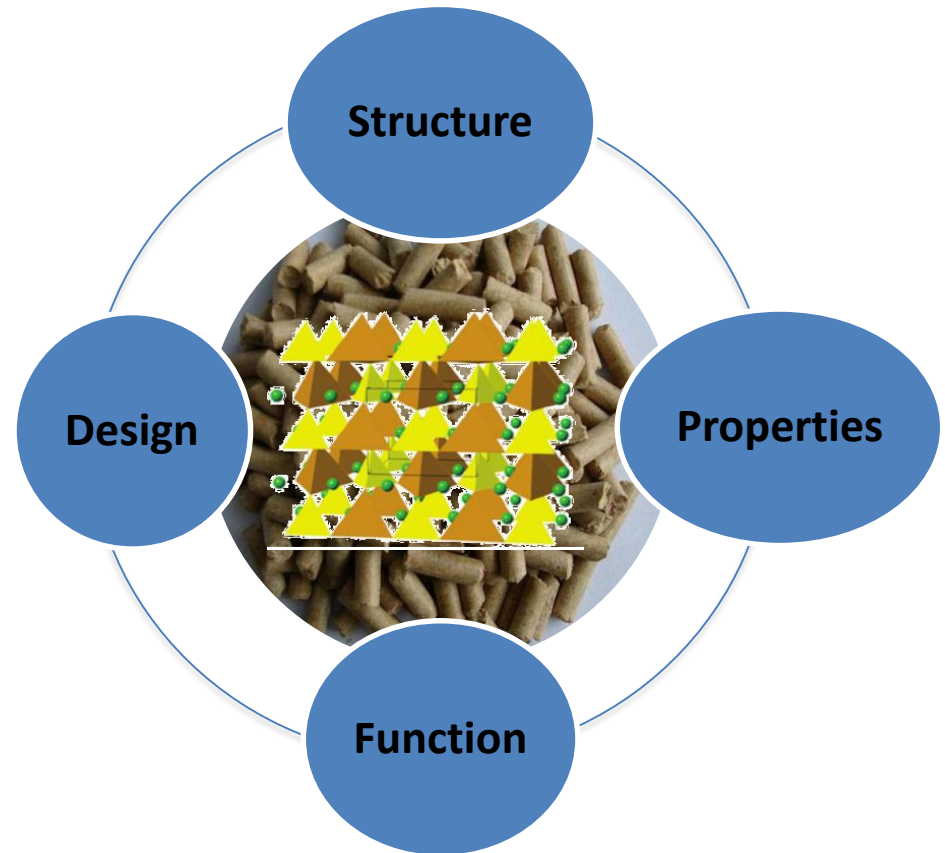


Motivation for in-situ x-ray diffraction

Studying materials under working conditions:

1. temperature
2. atmosphere
3. pressure
4. stress/strain
5. voltage/current
6. Light
7. ...

Provides information on the chemical and physical properties of materials and devices under realistic processing conditions



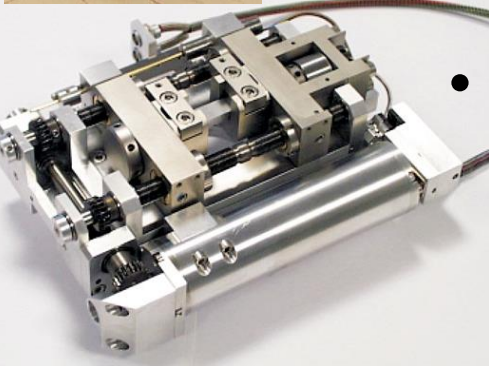
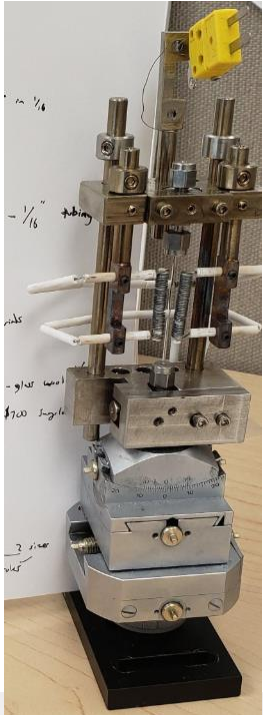
Applications: Microelectronics, batteries and fuel cells, catalysis, solar cells, materials under extreme conditions, etc.



Sample environments

Sample environments seek to mimic the operation conditions of the materials or devices being tested

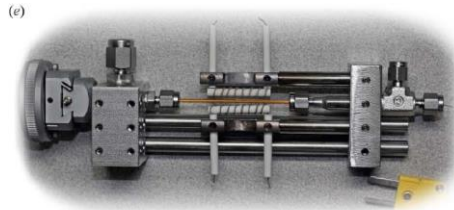
- Furnaces with controlled atmosphere and temperature
- Tensile rigs
- Temperature/humidity light chambers
- in situ cycling / temperature control for battery studies
- Customized setups



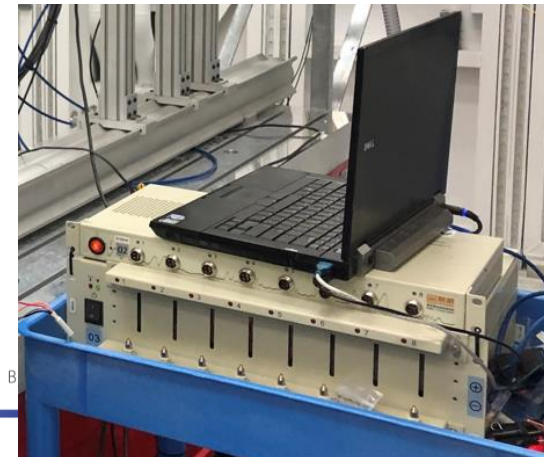
The Anton Paar domed heating stage (DHS1100) can be used to heat flat plate samples from ambient temperature to 1100 °C in a variety of inert gas atmospheres or under vacuum.



A Stoe capillary furnace can heat capillaries up to 1770K. Quartz capillaries can be used up to 1370 K, above which sapphire capillaries must be used. Users must supply their own sapphire capillaries.

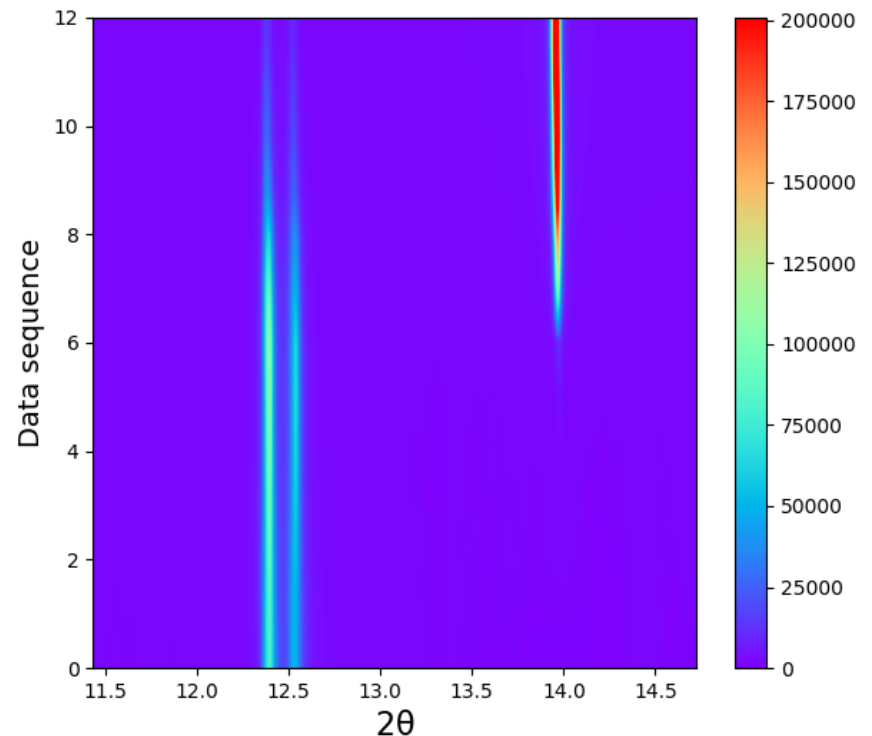


THE B



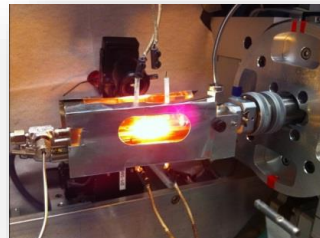
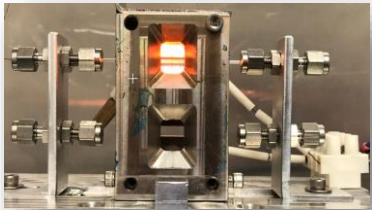
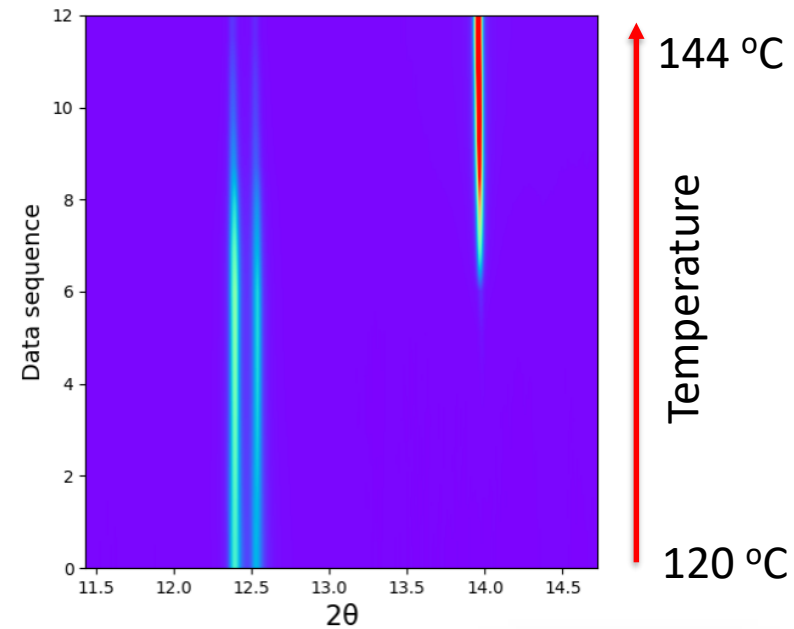
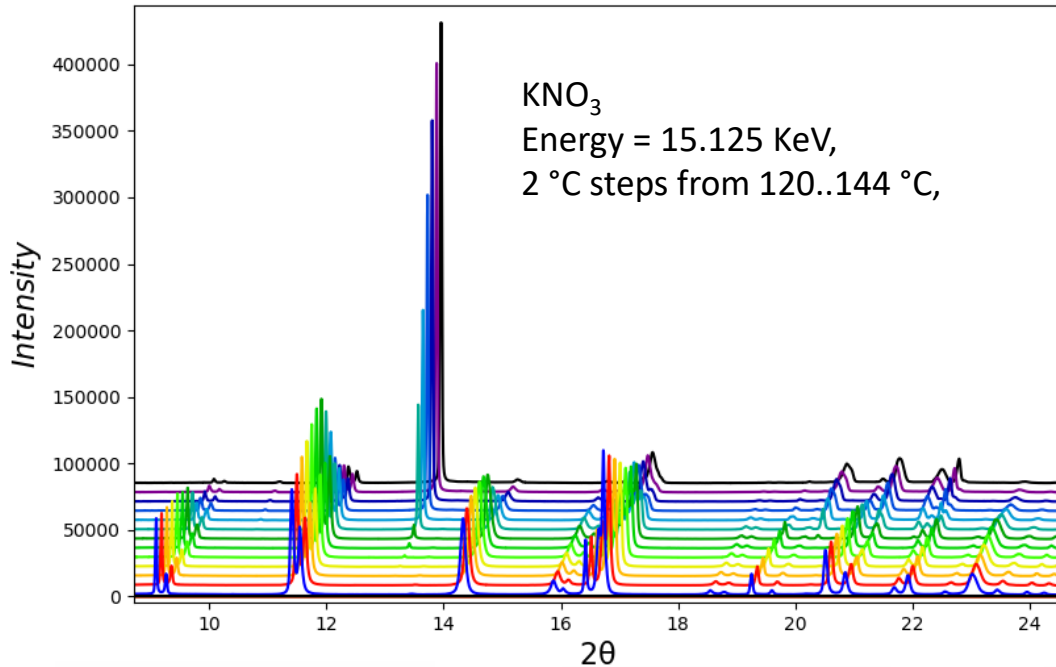
Contents

1. Phase transitions
2. Microelectronics
3. Batteries
4. Solar cells
5. Mechanical rigs
6. Catalysts
7. Corrosion
8. High pressure



Applications – Structural phase transitions

Orthorhombic α - KNO_3 \rightarrow trigonal β - KNO_3



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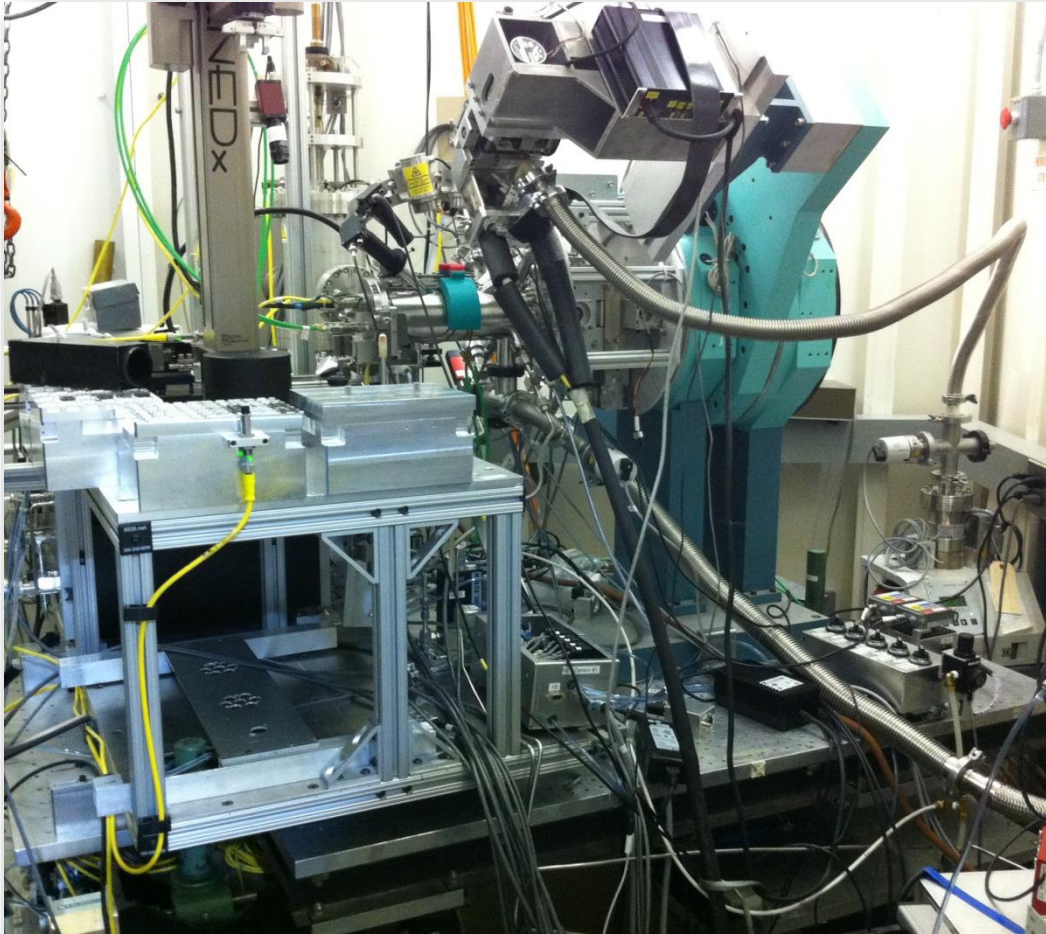
Applications to Microelectronics



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IBM end-station



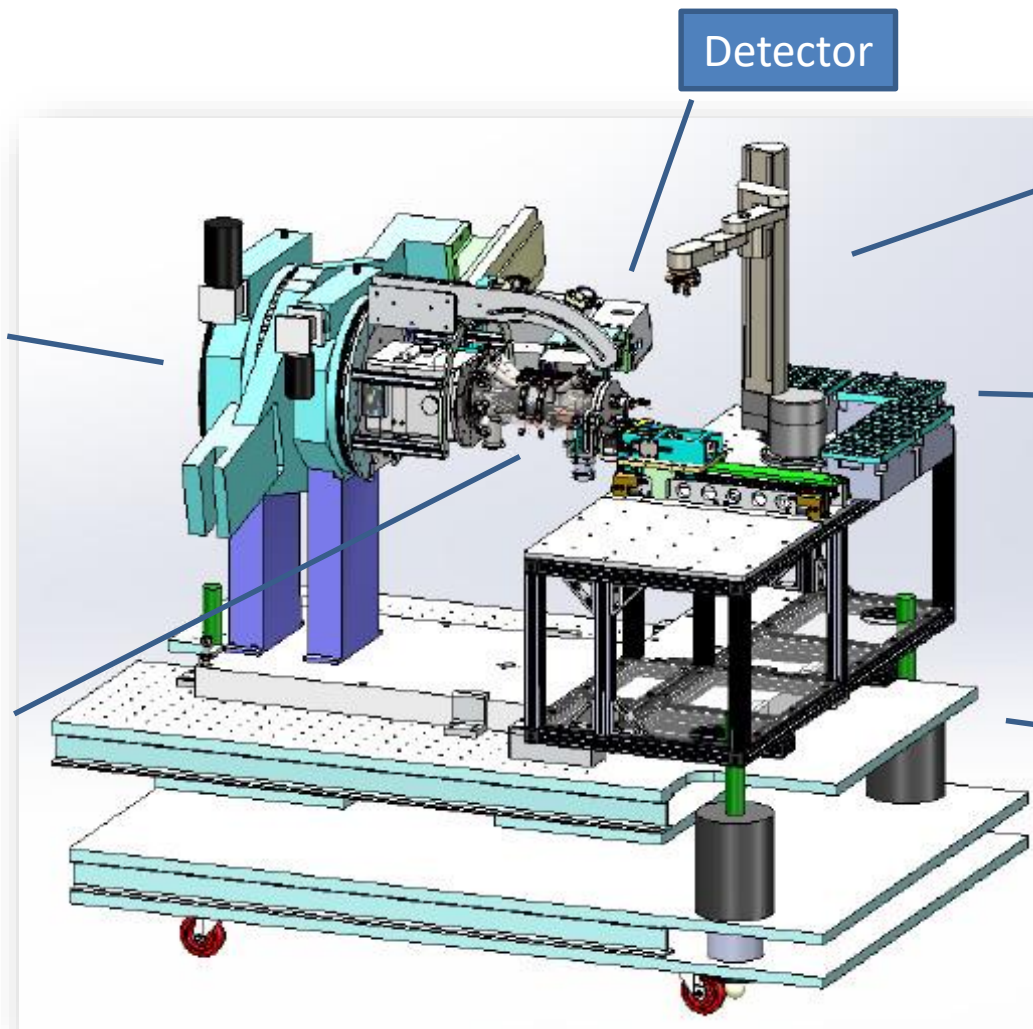
- **In-situ characterization**
 - thin films
 - microelectronic materials
 - semiconductors
- **Rapid thermal annealing (up to ~ 1100°C)**

Techniques

- ✓ X-Ray diffraction
- ✓ Four point probe to measure film resistivity
- ✓ Optical light scattering to measure surface morphology



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Huber
diffractometer

Detector

Robot

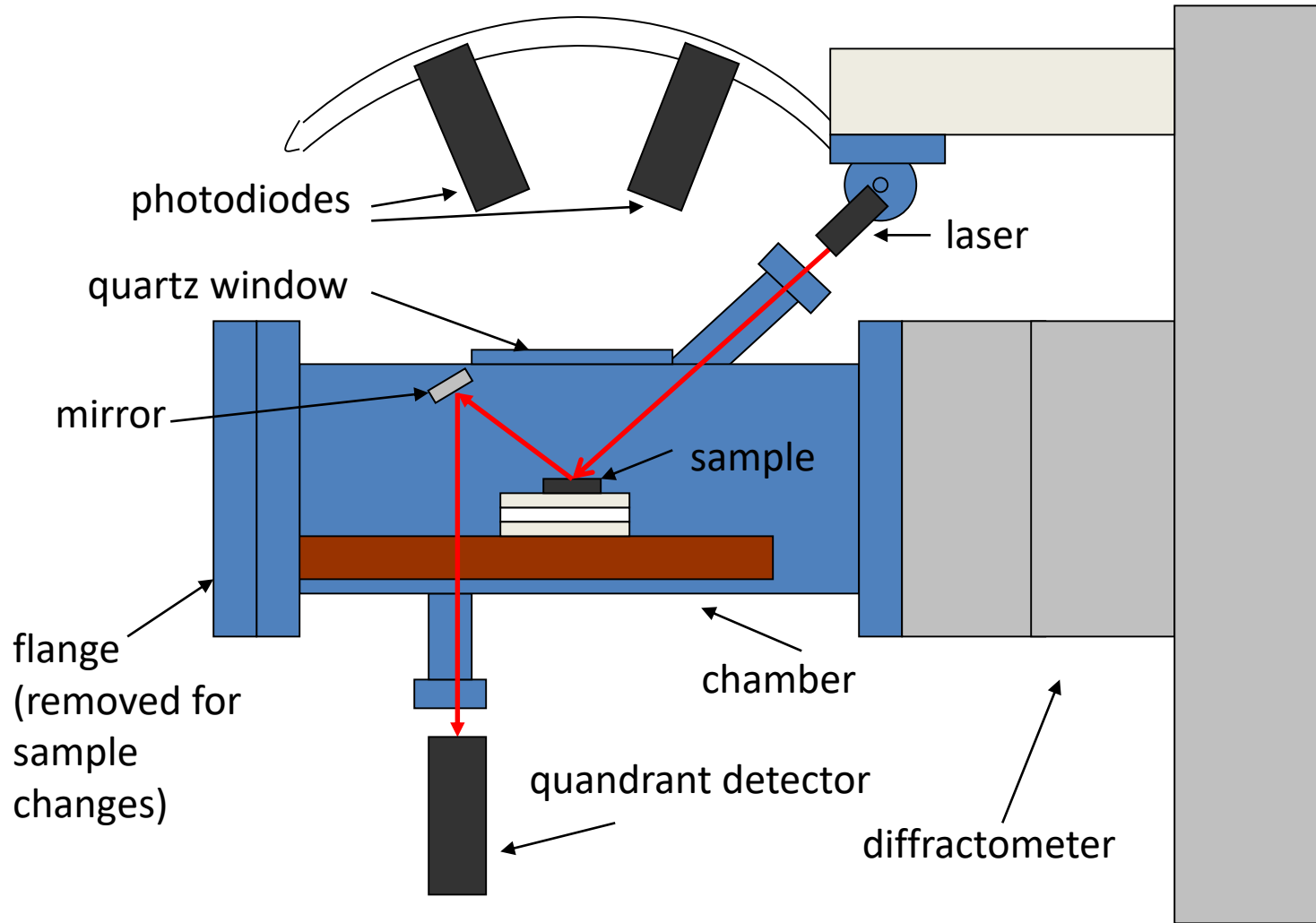
Sample
magazines

Experiment
chamber

Support
table



Experiment chamber

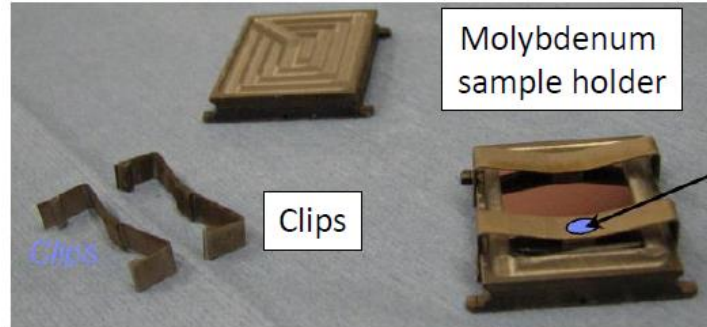
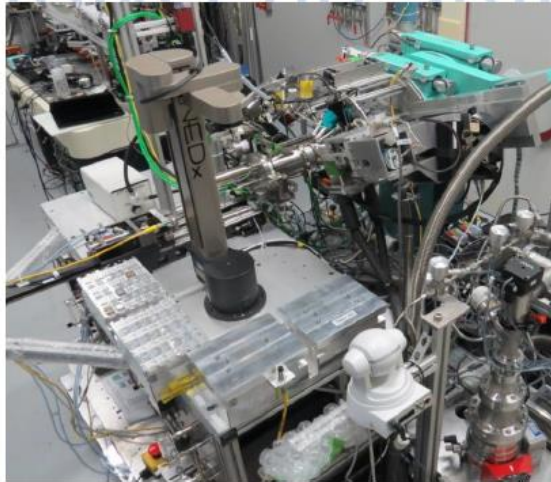


Sample size and format

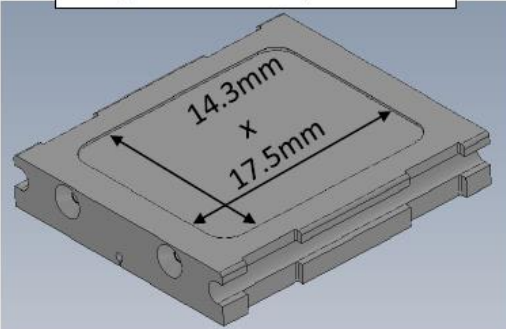
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Thin films studies

Combined diffraction, resistivity and roughness measurements under ultra high purity N₂ or He. Temperature up to 1100 °C.

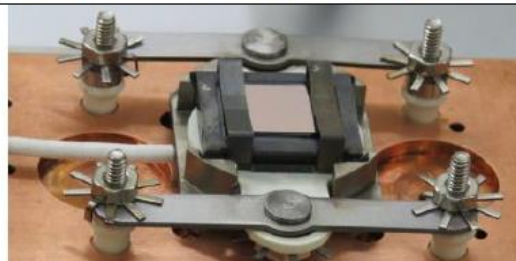


Molybdenum sample holder

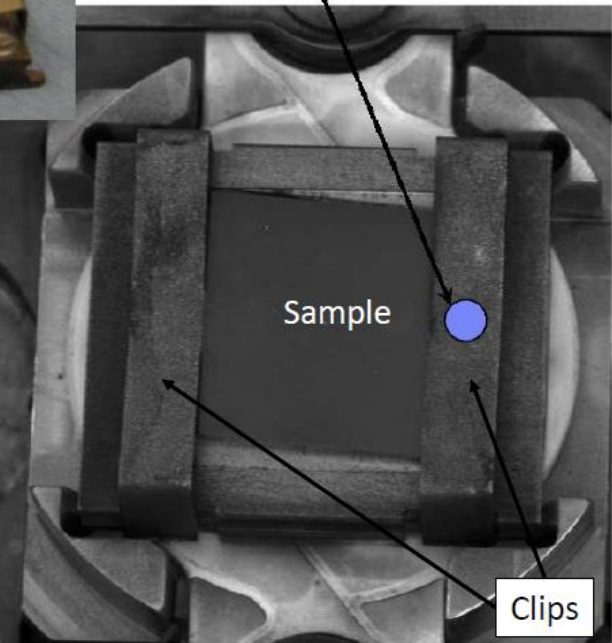


Sample size should be approximately 12mm x 15 mm

Sample holder on top of the heater



Maximum temperature: 1100°C



IBM End-Station – Canadian Light



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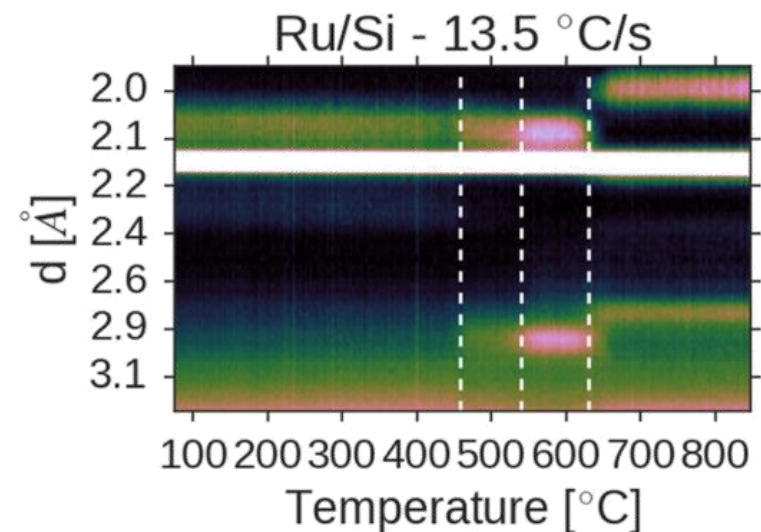
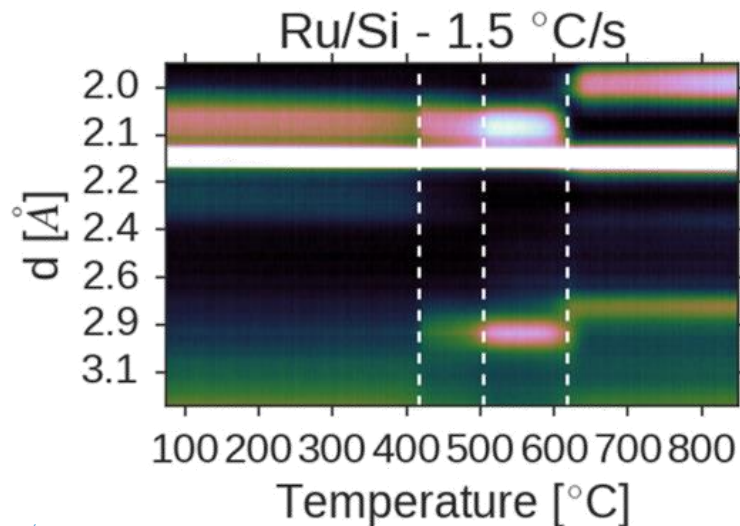
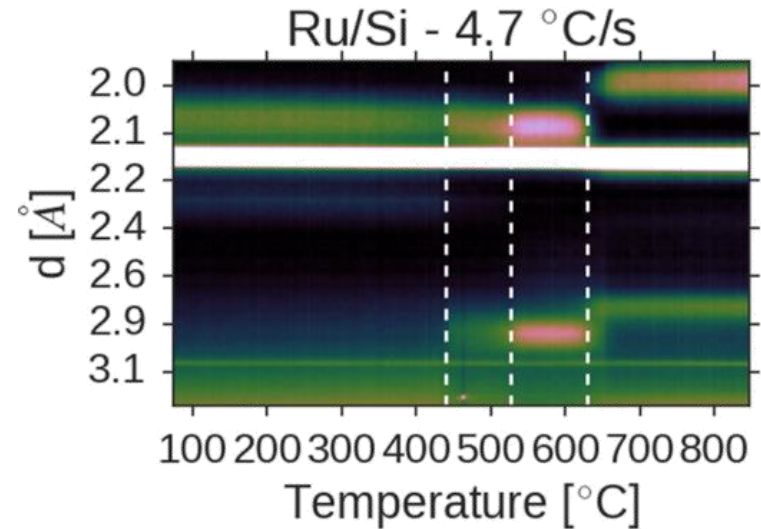
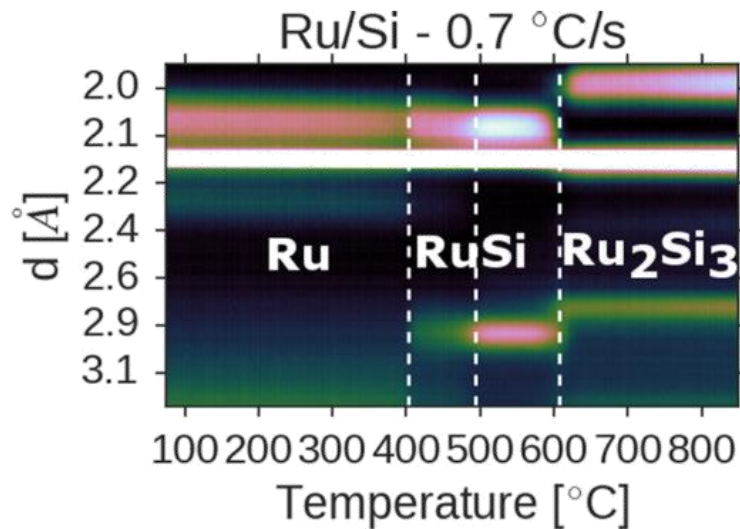
IBM end-station in action



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Atomic layer deposited ultrathin metal nitride barrier layers for ruthenium interconnect applications.



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- ✓ Quick investigation of different growth conditions
- ✓ Better nucleation processes
- ✓ Improved film stability at higher temperatures
- ✓ Lower interface roughness

Microelectronic Engineering
83, 2042-2054 (2006)

C. Lavoie et al.

Effects of additive elements on the phase formation and morphological stability of nickel monosilicide films

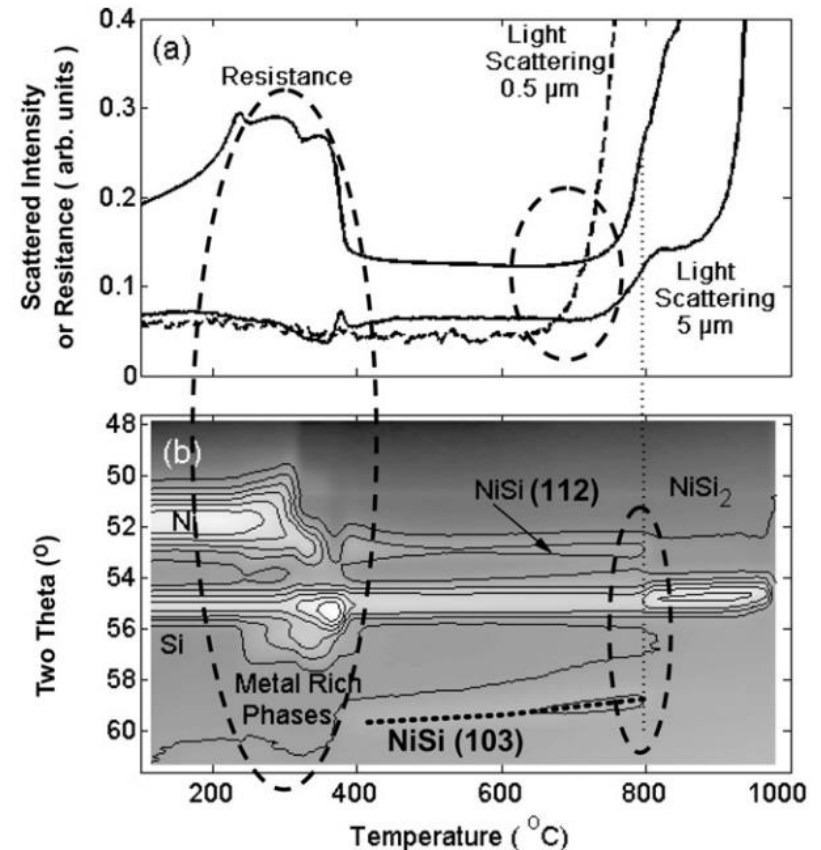
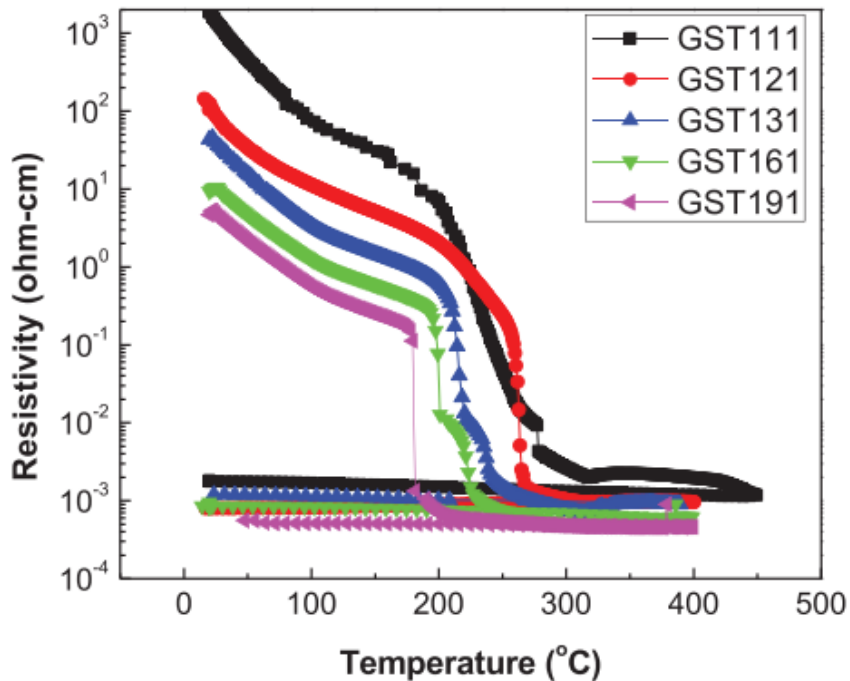


Fig. 4. (a) Elastic light scattering at 0.5 μm and 5 μm length scales and resistance measurements together with (b) X-ray diffraction measurements performed *in situ* during annealing in purified He of a 15 nm Ni layer deposited on a 100 nm poly-Si film (3 °C/s). The three ellipses also refer to the challenges discussed using the phase diagram in Fig. 1.

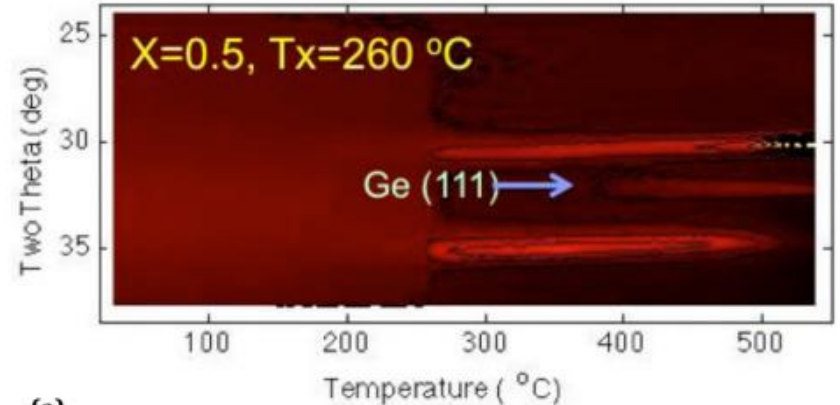


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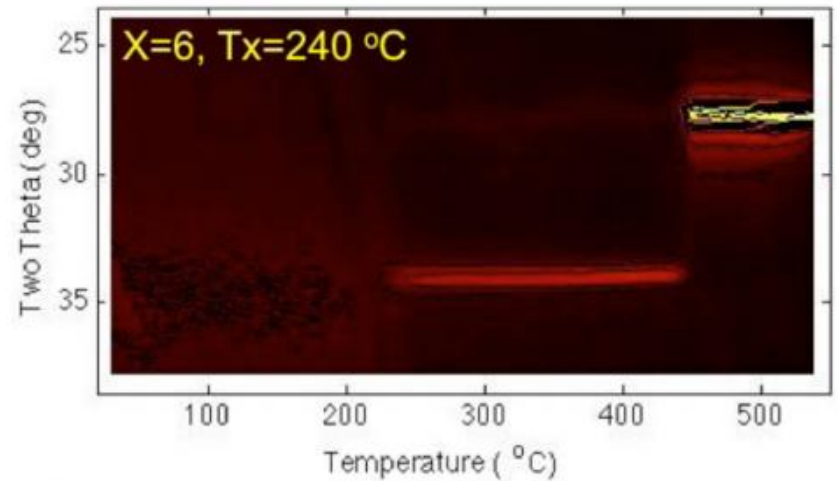


Resistivity as a function of temperature for Ge₁Sb_xTe₁ films for $x > 1$ during a heating ramp to 300–450 °C at 5 °C/min and subsequent cooling back to room temperature.

Crystallization properties of materials along the pseudo-binary line between GeTe and Sb



(a)

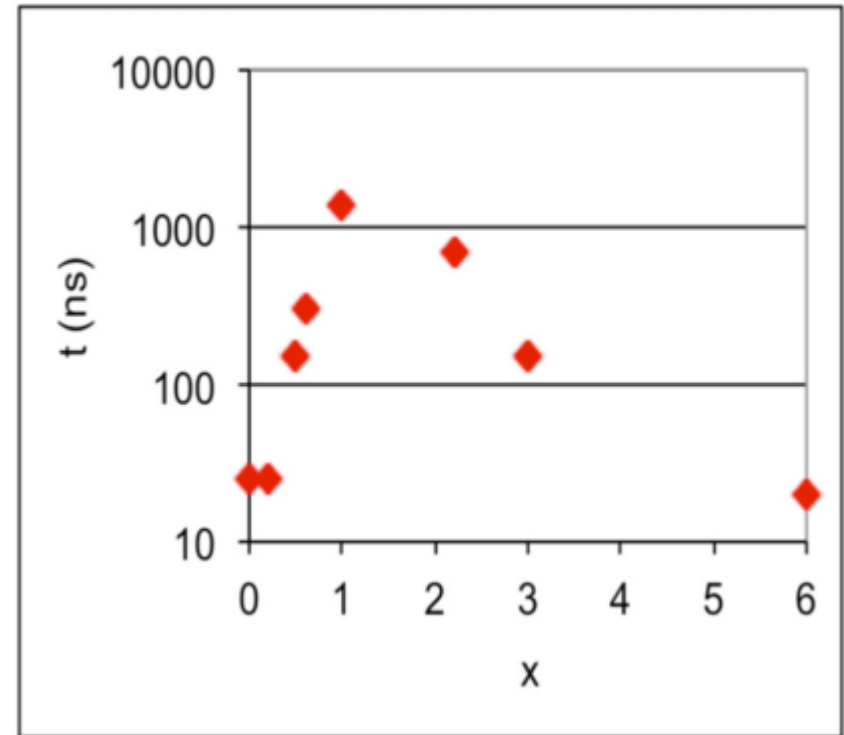
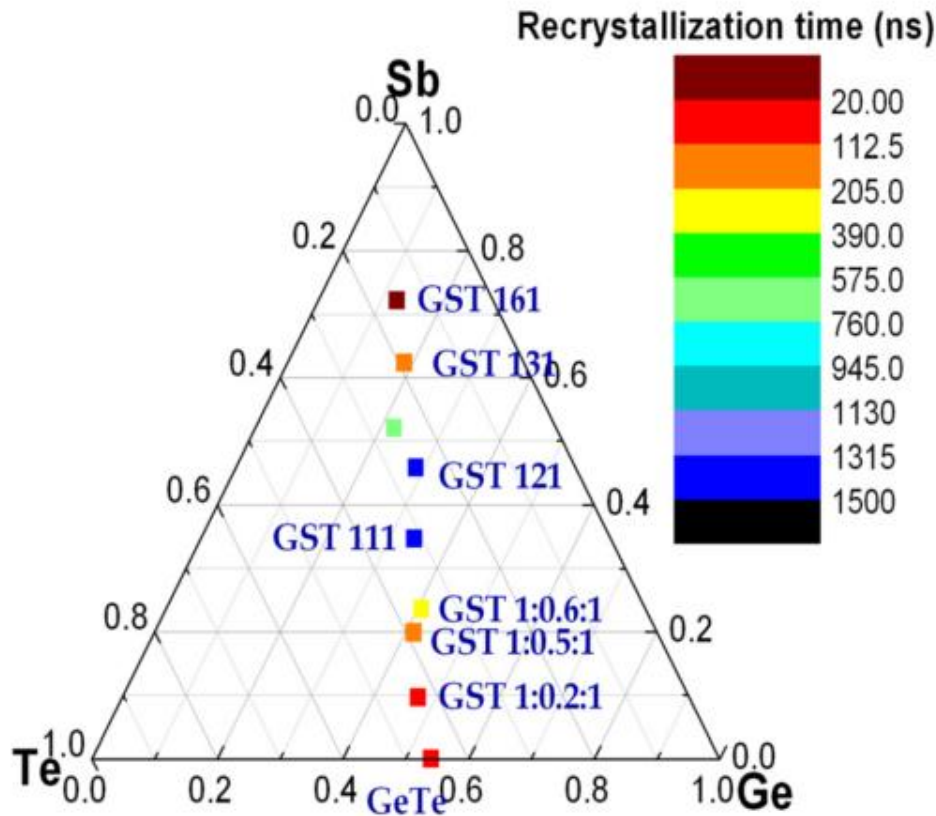


(b)

FIG. 2. XRD peak intensity as a function of temperature T during heating at 3 °C/s to 550 °C of a Ge₁Sb_xTe₁ film with (a) $x=0.5$ and (b) $x=6$, respectively.

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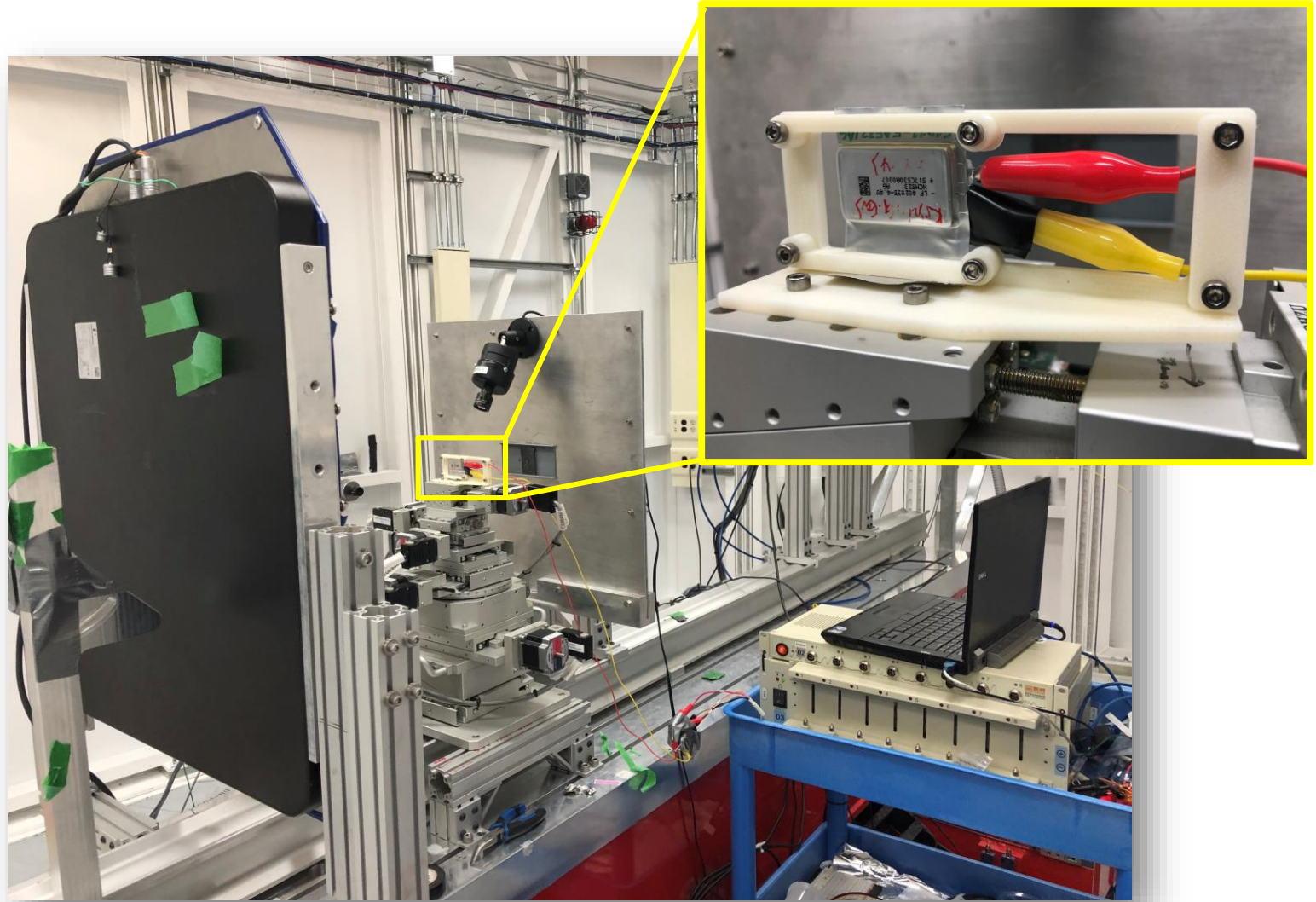
Crystallization properties of materials along the pseudo-binary line between GeTe and Sb



Battery experiments

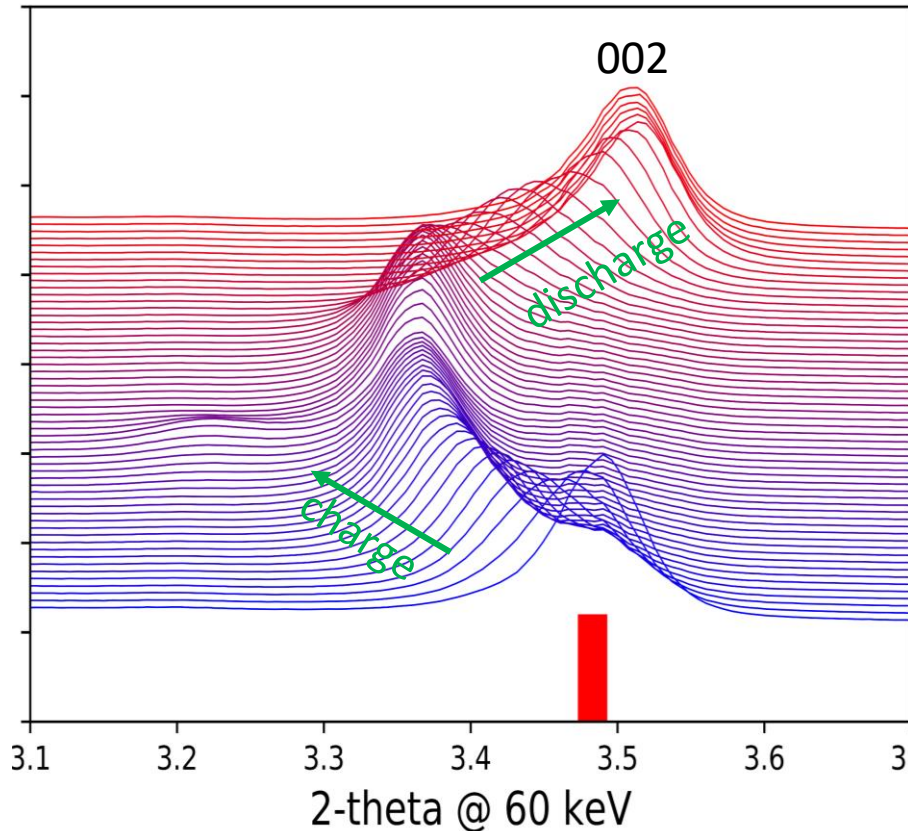


In-situ battery research

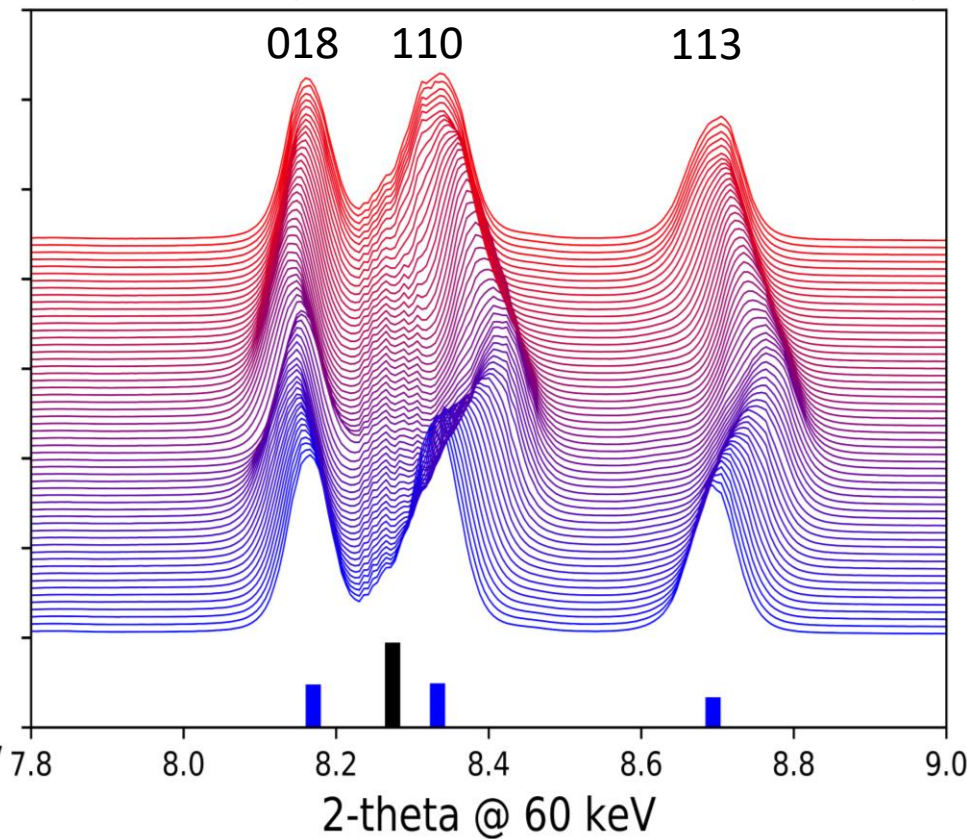


In-situ battery research

Anode (Graphite)

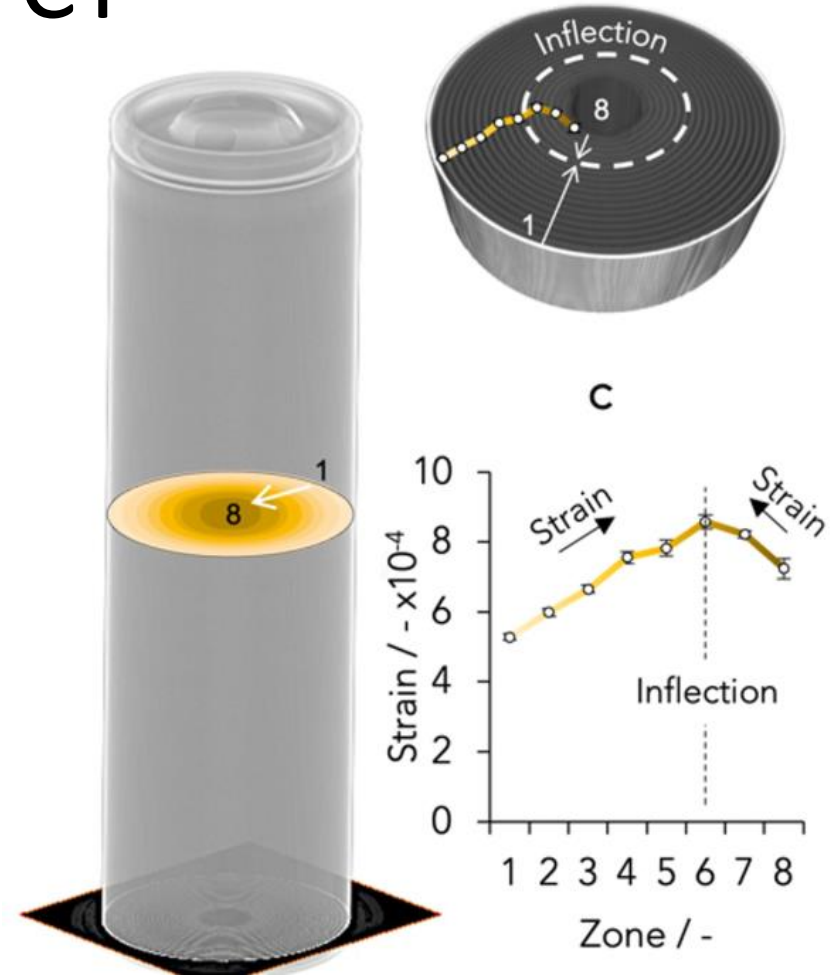
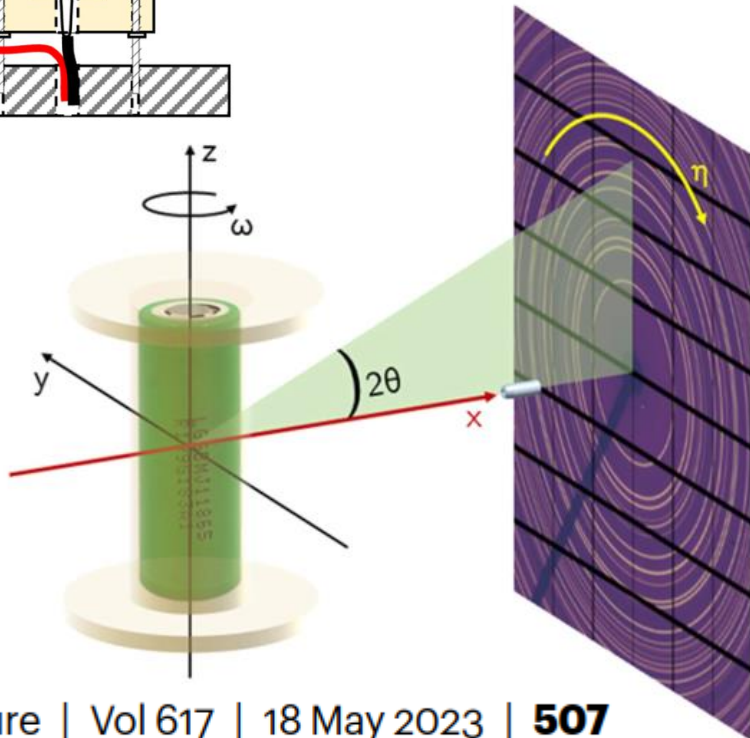
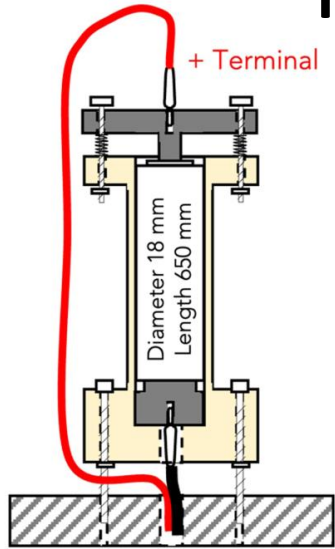


Cathode (Transition Metal Oxide)



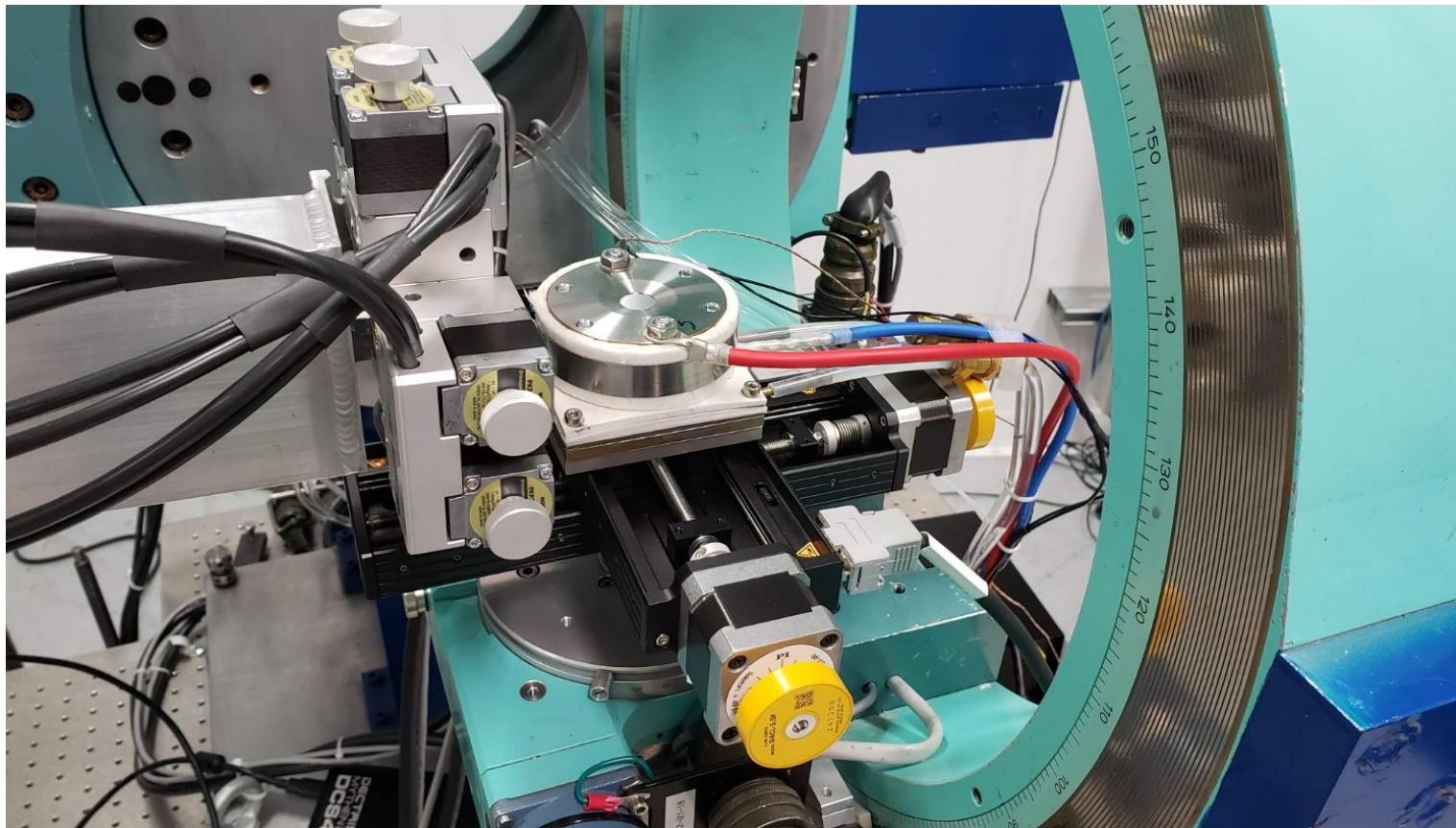
Mapping internal temperatures during high rate battery applications

XRD - CT



In-situ battery research

High Temperature Compatible Conflat Cell with Adjustable Stack Pressure for In-Situ and Operando X-Ray Studies of Lithium-Ion Battery Materials



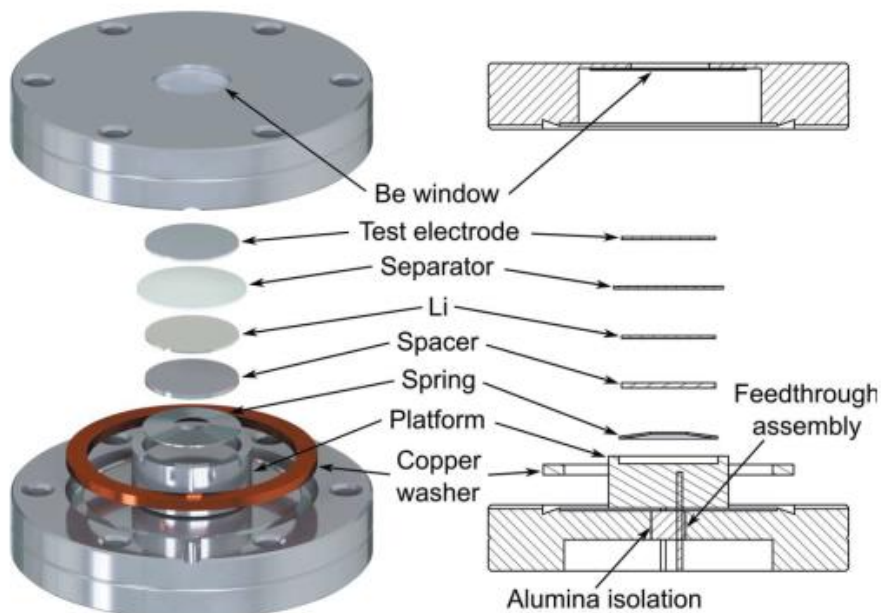
Michael Fleischauer - NRC / University of Alberta



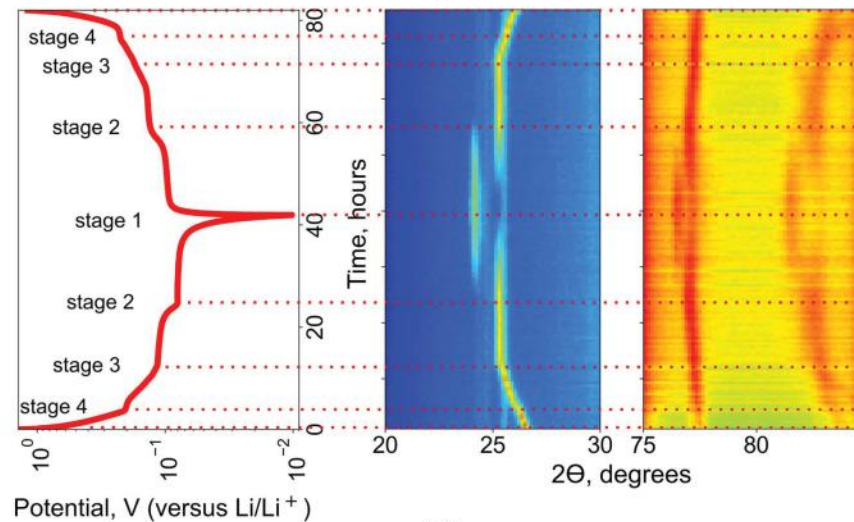
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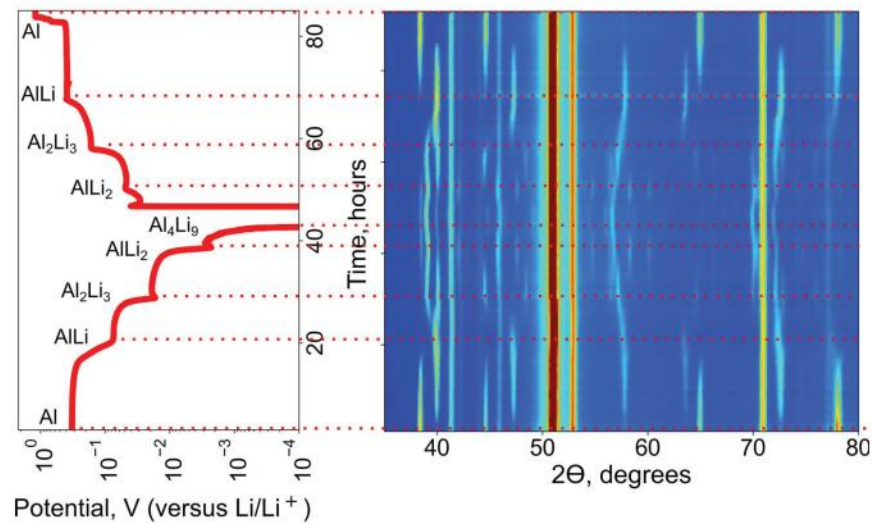
In-situ battery research



(a)

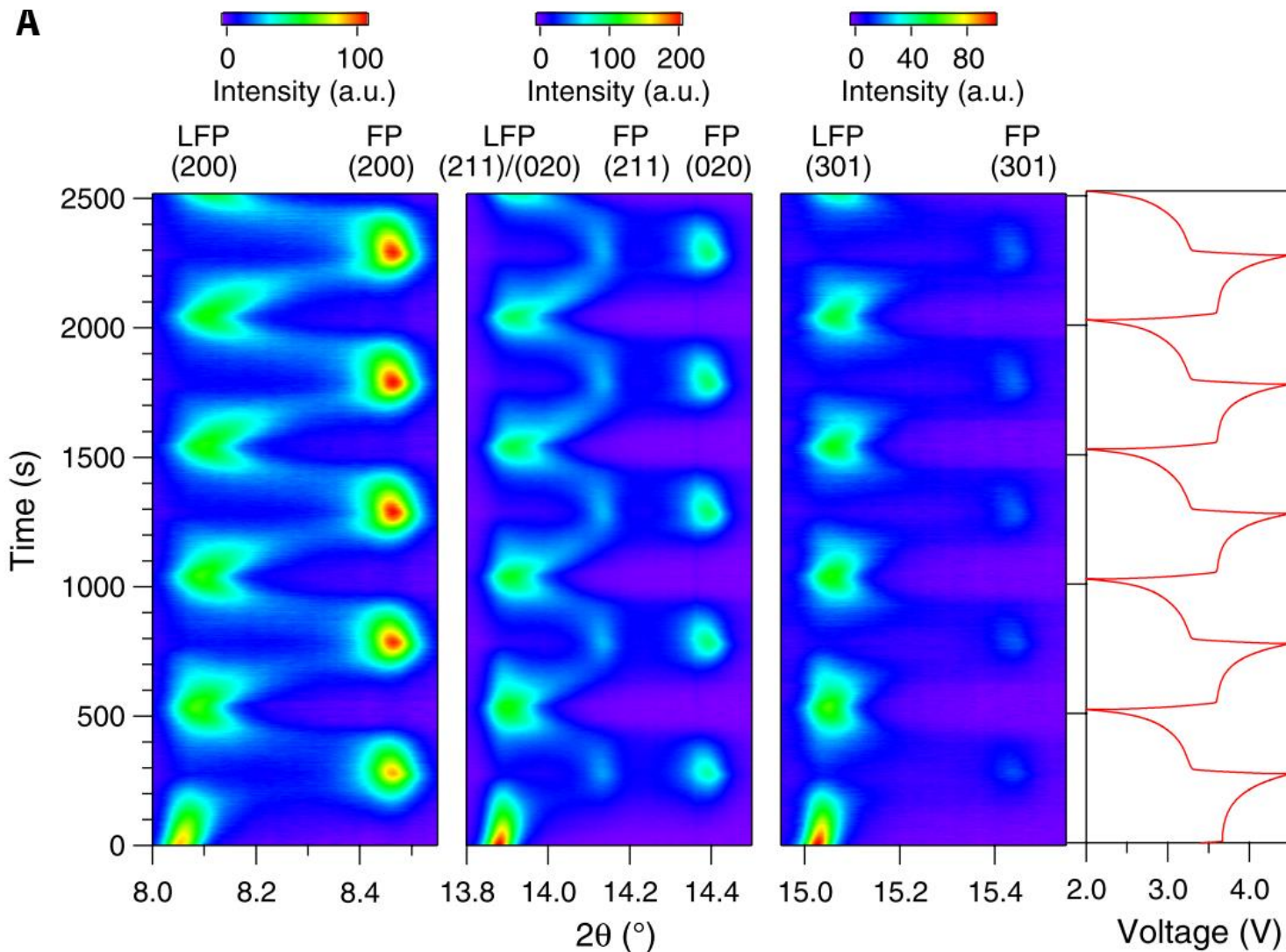


(a)



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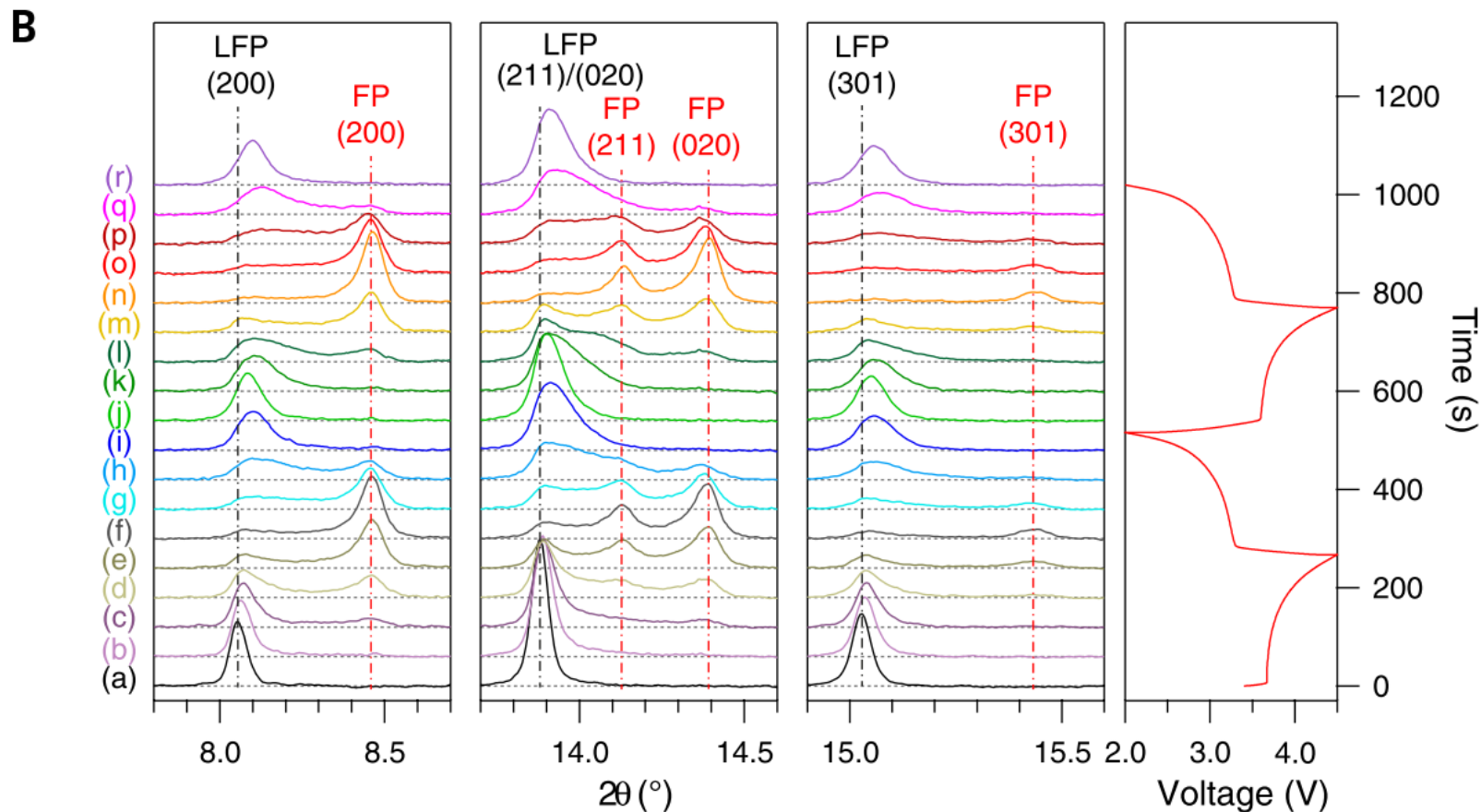
Capturing metastable structures during high-rate cycling of LiFePO_4 nanoparticle electrodes

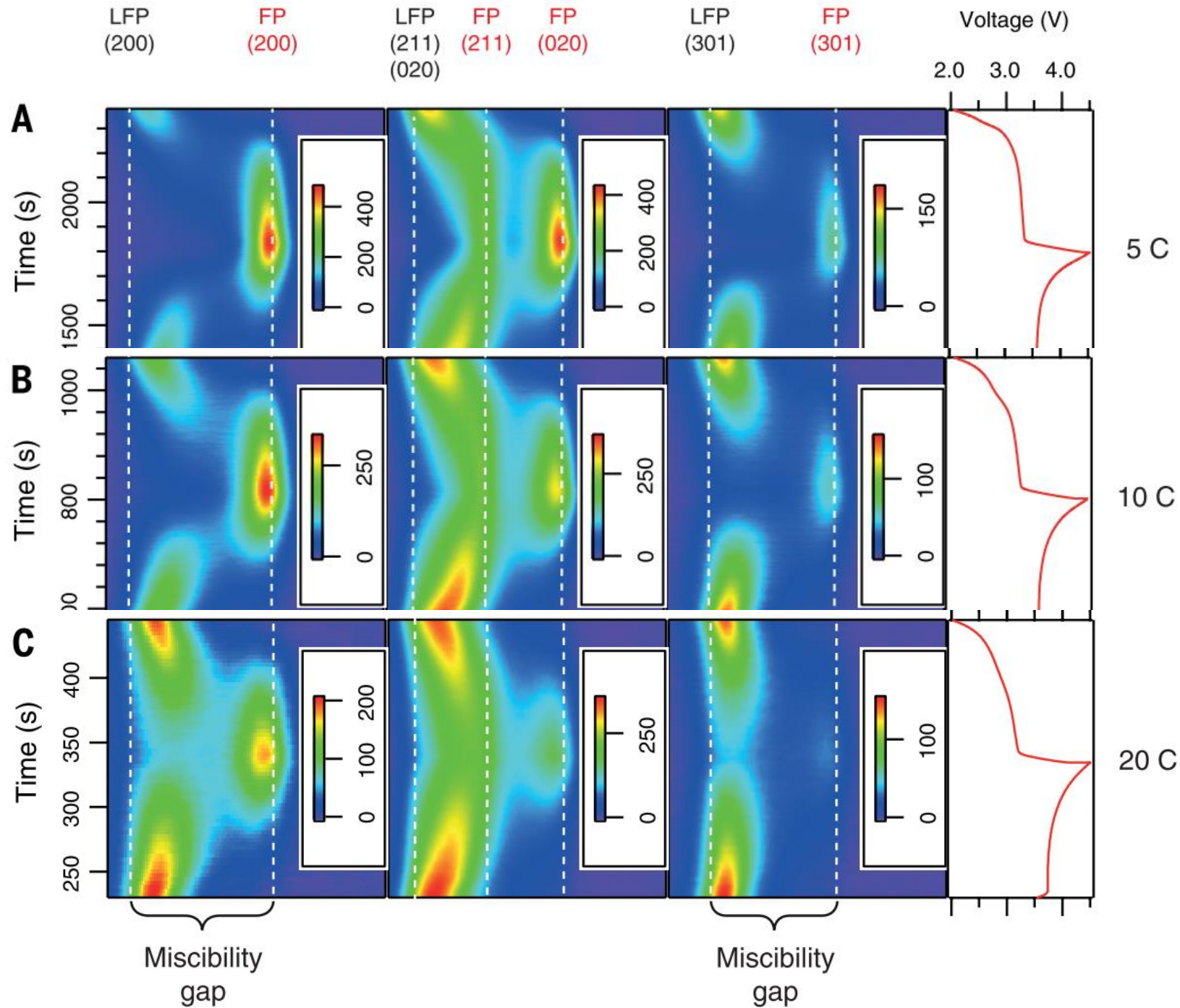


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le rayonnement
synchrotron

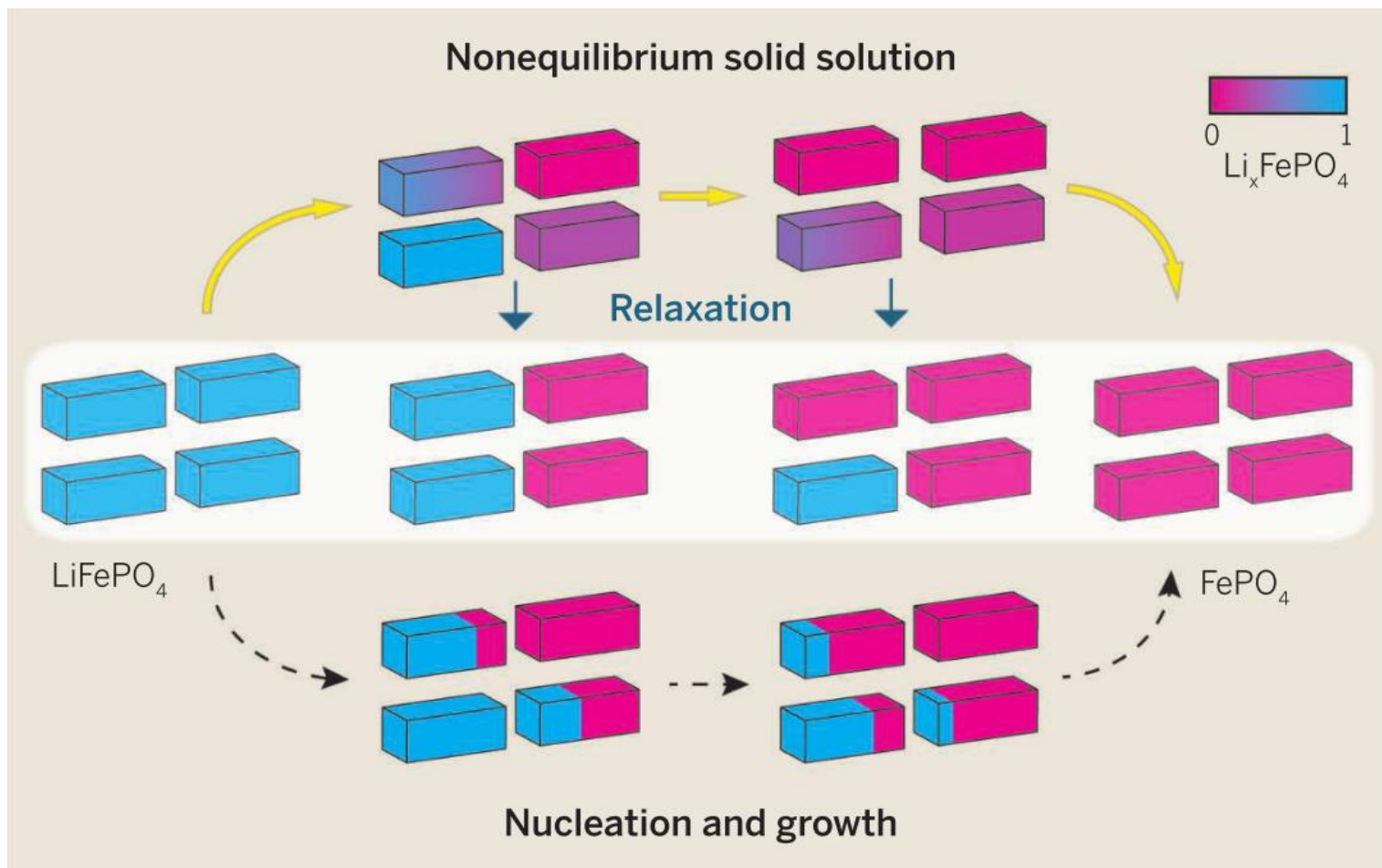
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Capturing metastable structures during high-rate cycling of LiFePO_4 nanoparticle electrodes



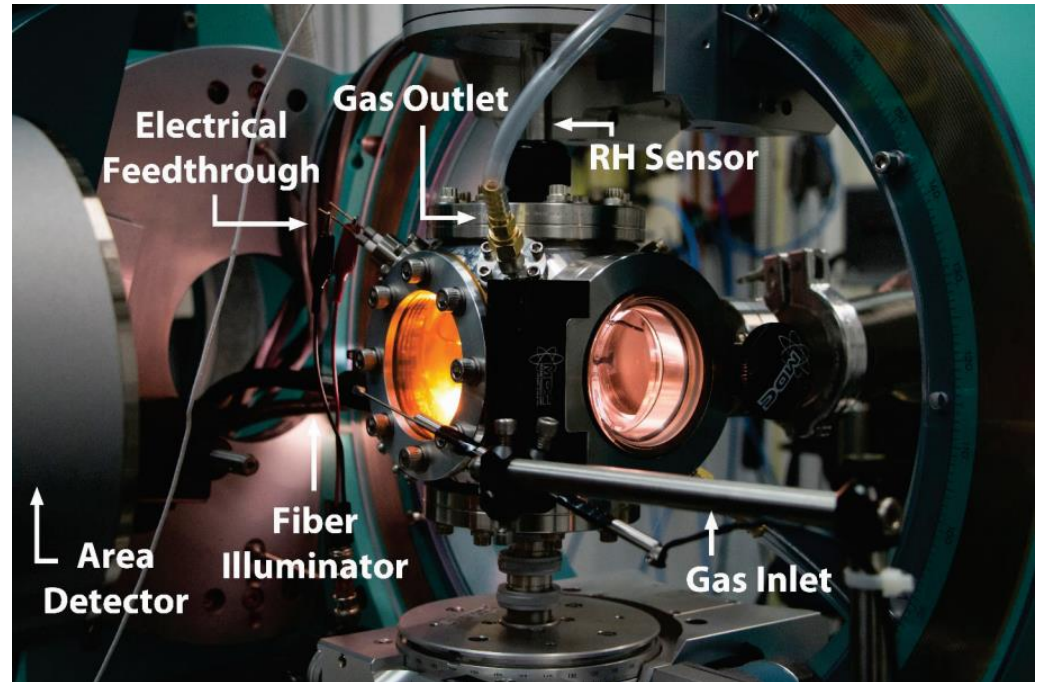
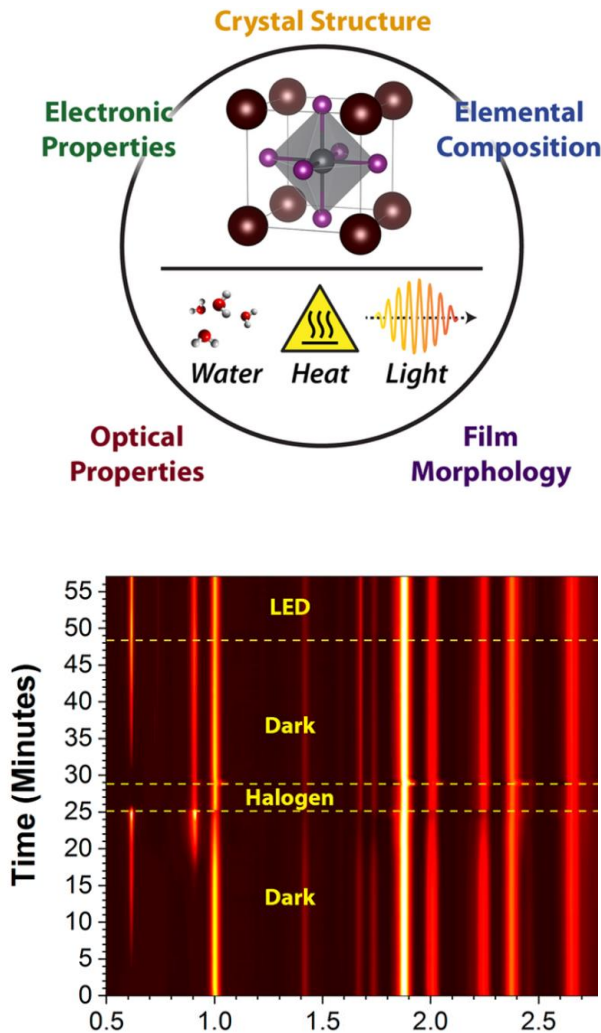


Capturing metastable structures during high-rate cycling of LiFePO_4 nanoparticle electrodes



Solar cell research

In situ studies of the degradation mechanisms of perovskite solar cells



Elucidating the Failure Mechanisms of Perovskite Solar Cells in Humid Environments Using In Situ GIWAXS

Tim Kelly - USASK

Corrosion studies

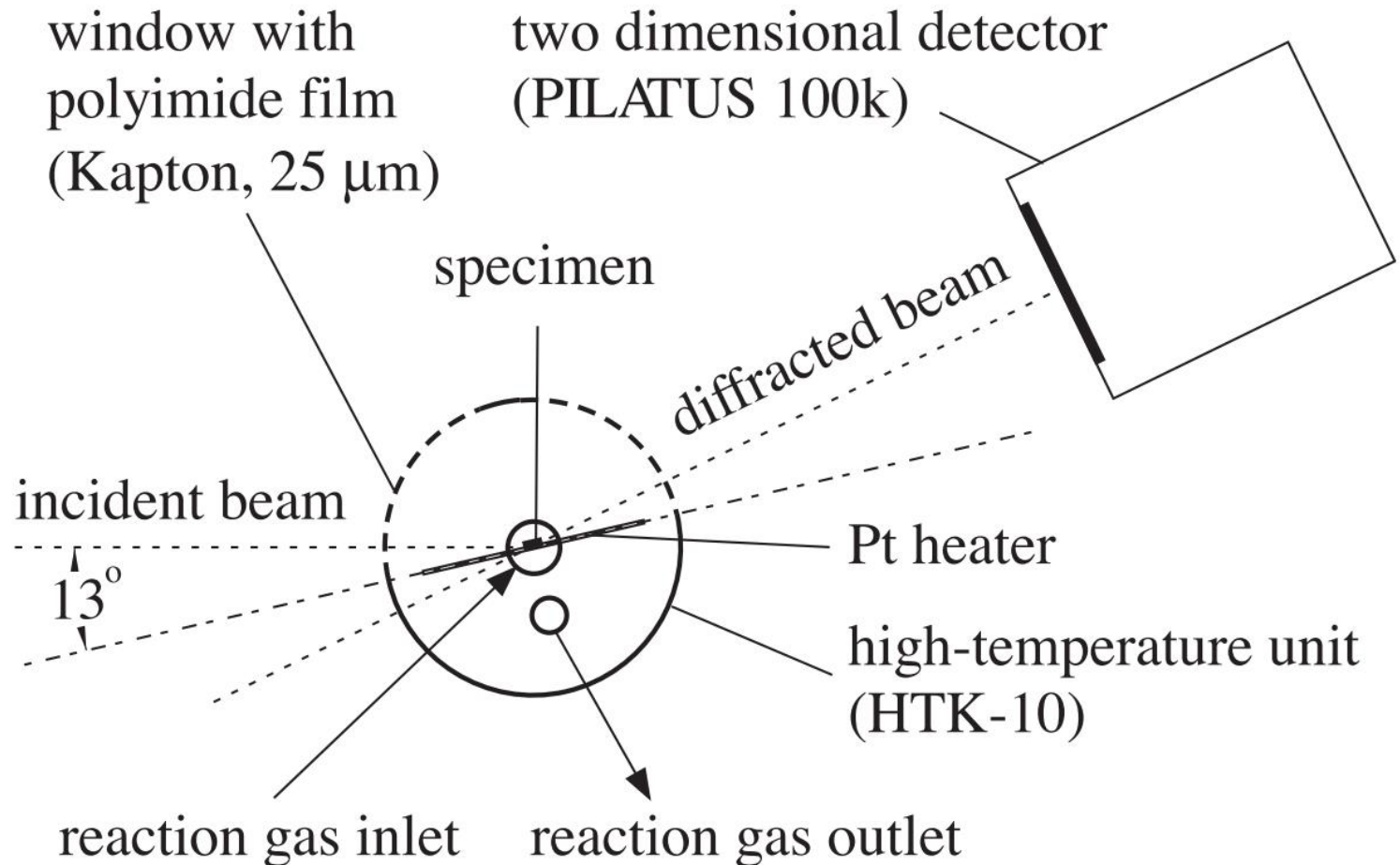
Corrosion can lead to failure involving personal injuries, fatalities, unscheduled shutdowns and environmental contamination



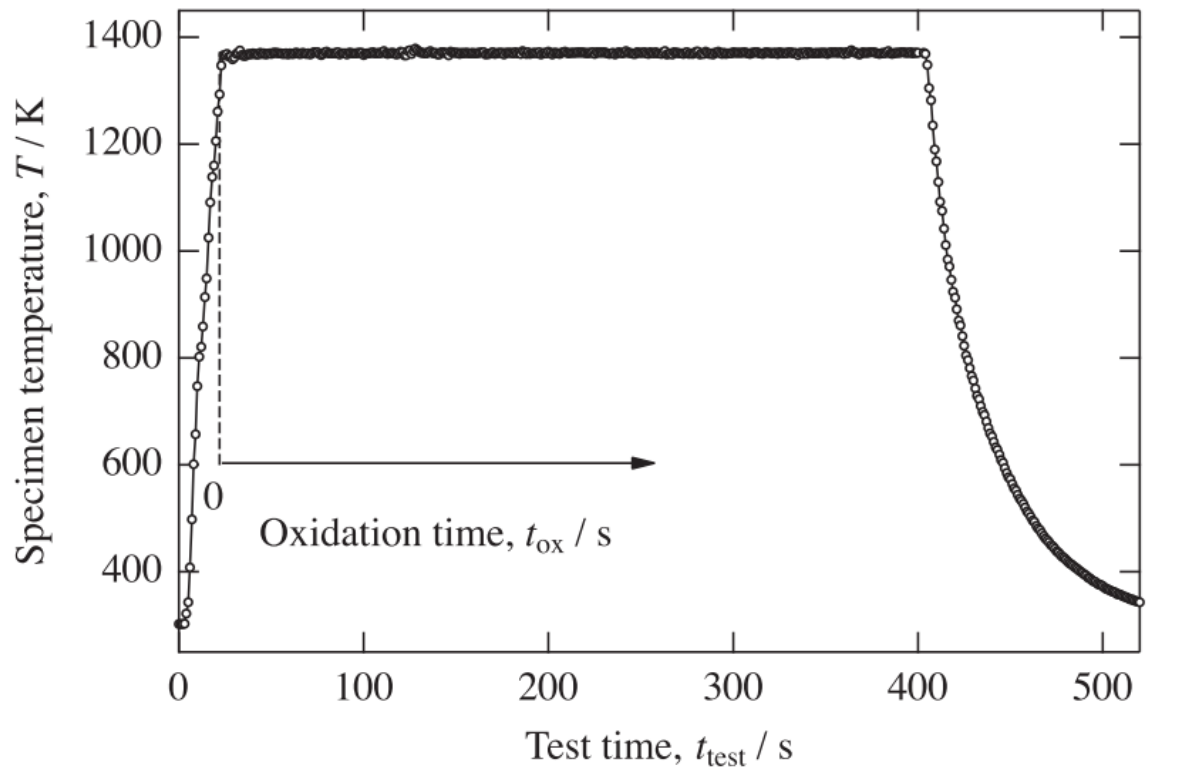
Several studies over the past 30 years have shown that the annual direct cost of corrosion to an industrial economy is approximately 3.1% of the country's Gross National Product (GNP). In the United States, this amounts to over \$276 B per year. For Canada, \$60 B per year.



In situ studies of breakaway oxidation in type 430 stainless steel



In situ studies of breakaway oxidation in type 430 stainless steel



These specimens were heated at a rate of 50 K/s and kept at 1373 K for ~ 400 s;

They were then cooled in the high-temperature unit.

Gas mixture:

- 17 vol.% O₂
- 20 vol.% H₂O
- N₂ gas

Rate of 8.3 cm³/s

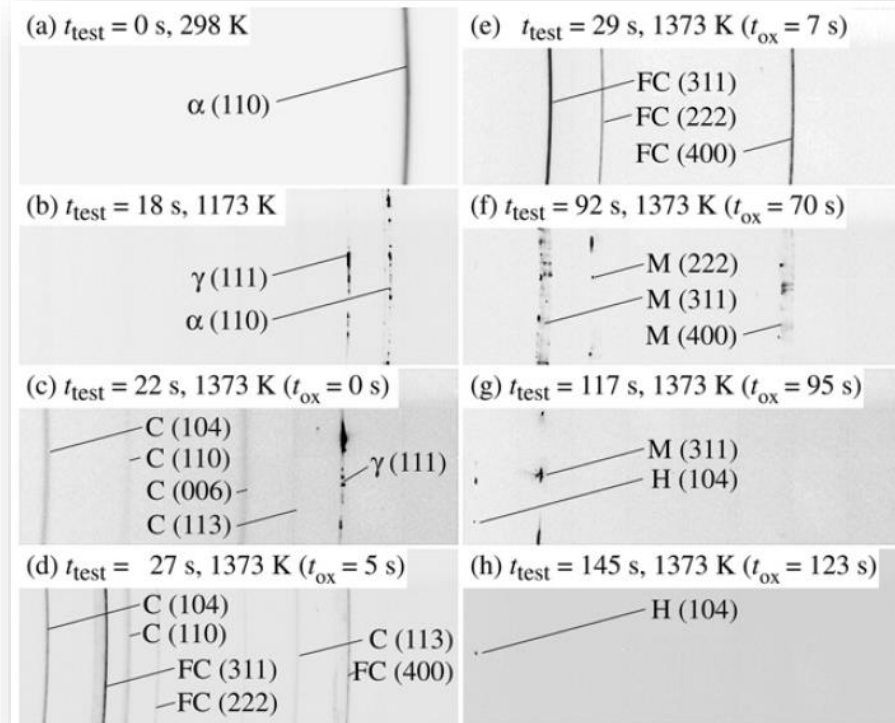
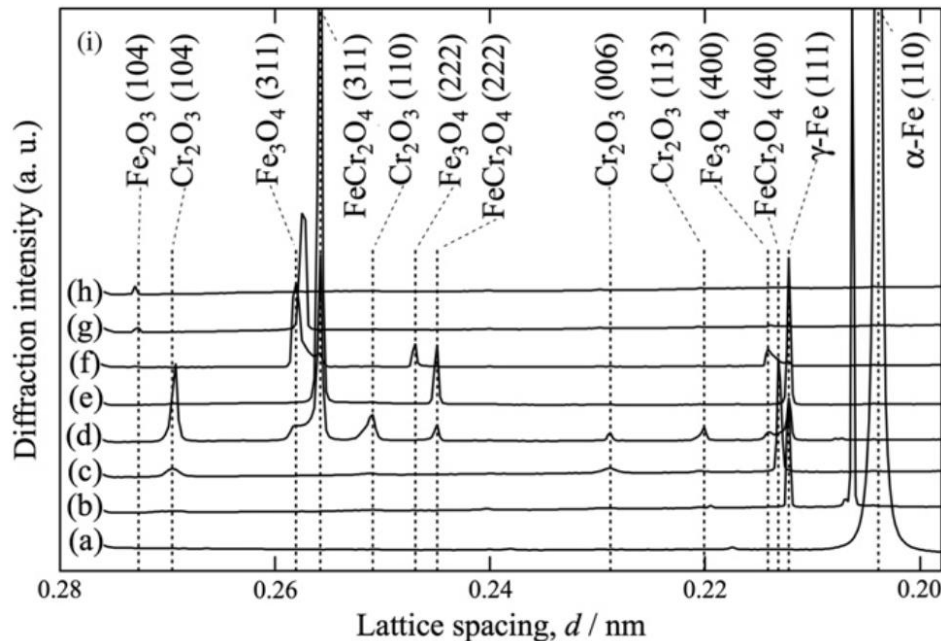


In situ studies of breakaway oxidation in type 430 stainless steel

Commercial type 430 stainless steel was used.

The composition of the steel was:

- 0.054 mass% C
- 0.55 mass% Si
- 0.09 mass% Mn
- 0.004 mass% S
- 0.13 mass% Ni
- **16.1 mass% Cr**
- and Fe

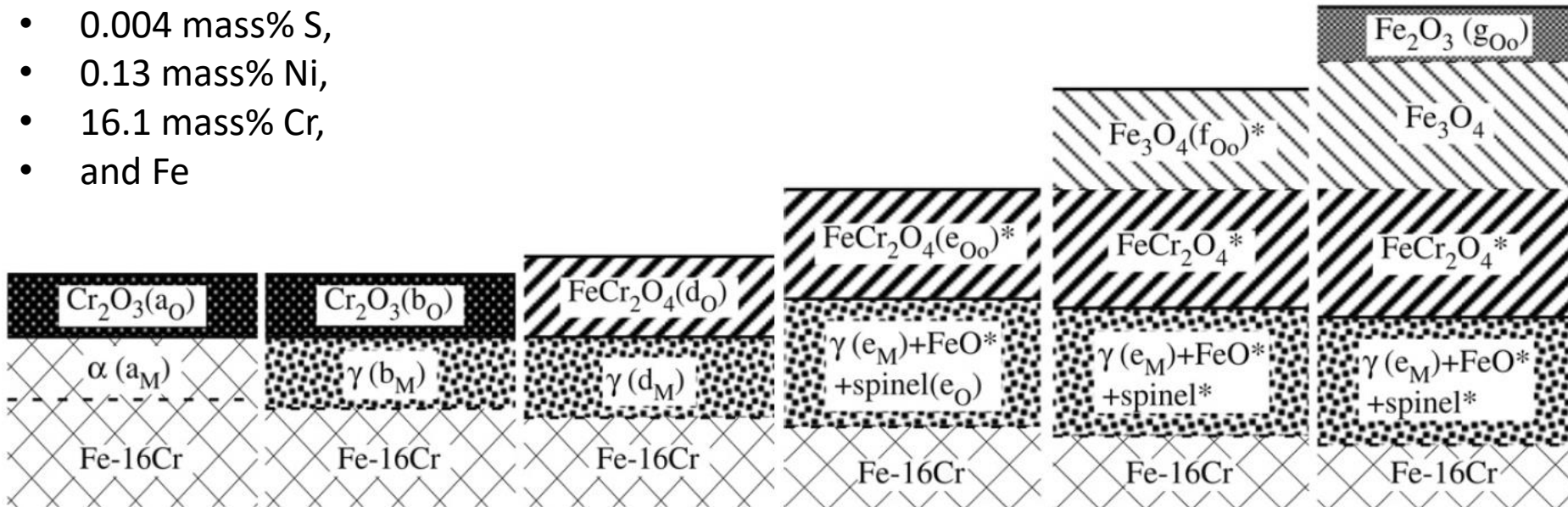


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- and Fe



In situ mechanical testing



Gleeble 3S50

Gleeble®Synchrotron

Allows the material of interest to be subject to a wide range of thermo-mechanical conditions

Applications:

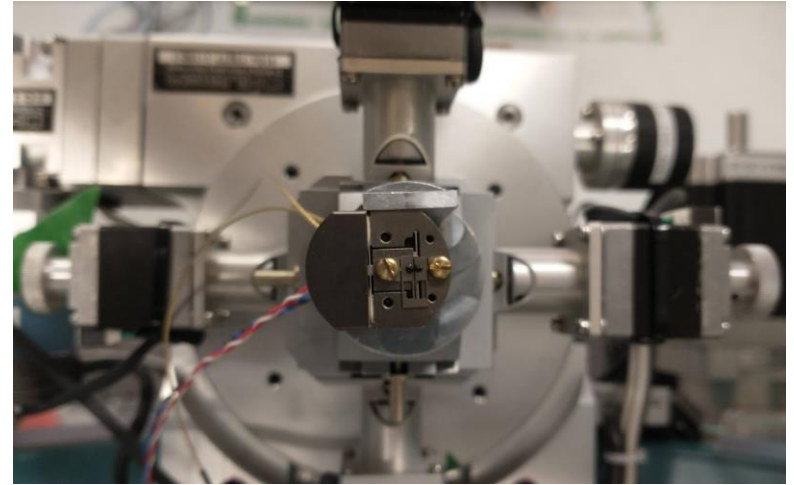
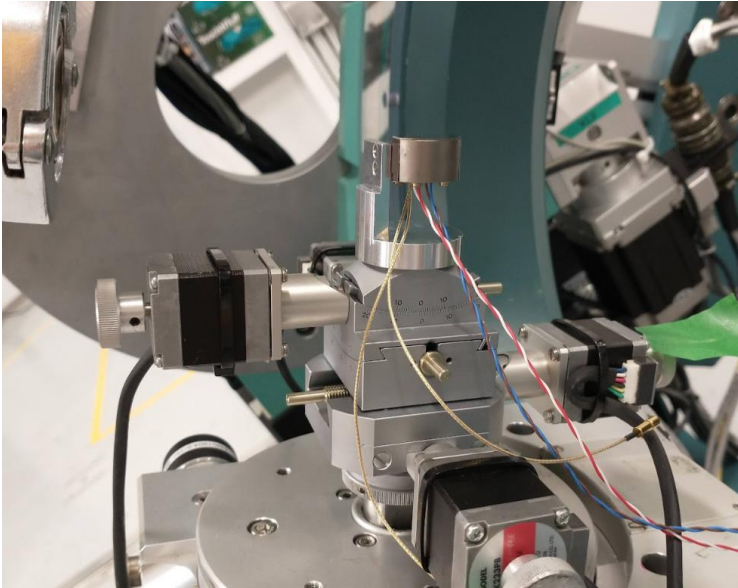
- ✓ Phase transformations
- ✓ Residual stress evolution
- ✓ Corrosion
- ✓ Oxidation



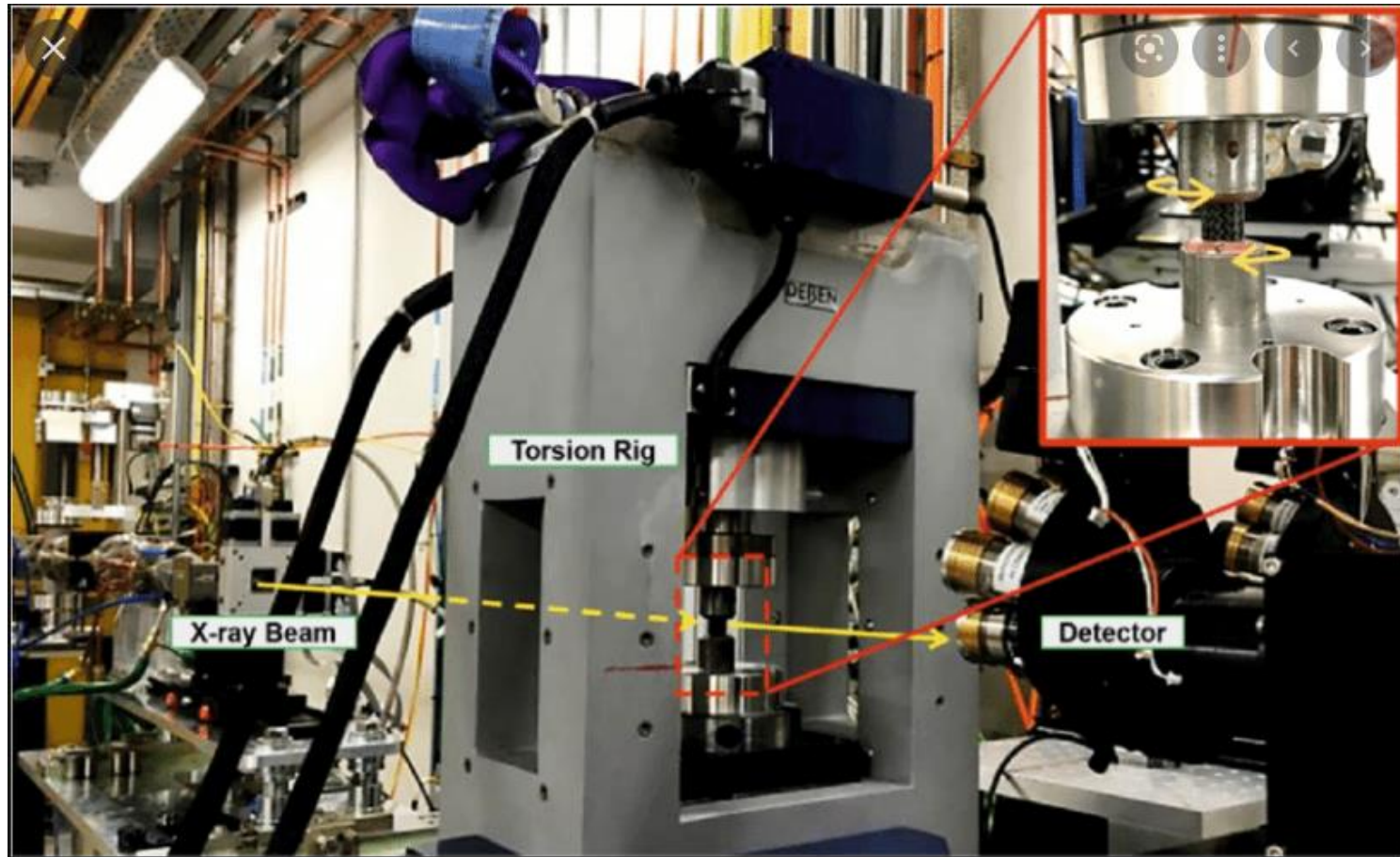
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Source de synchrotron

<https://www.gleeble.com/products/specialty-systems/gleeble-synchrotron.html>

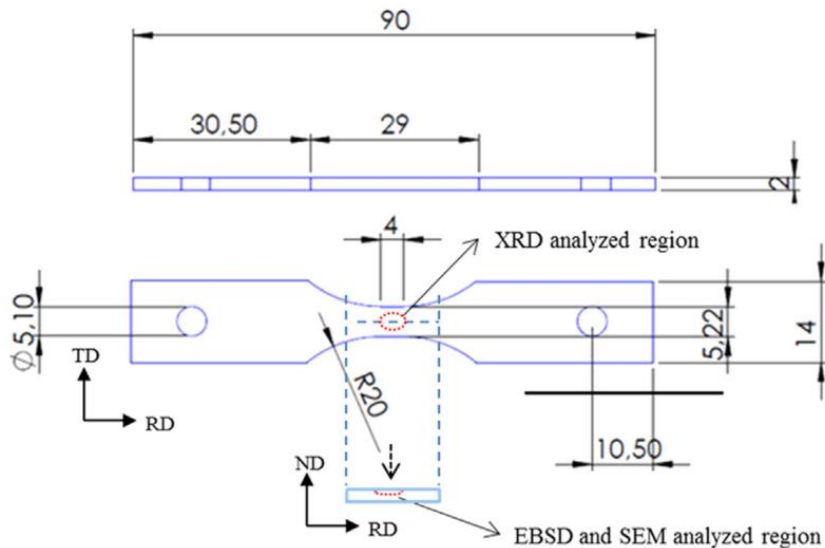
Razorbill strain cell



Deben 20kN stress rig at Diamond



In-situ synchrotron x-ray evaluation of strain-induced martensite in AISI 201 austenitic stainless steel during tensile testing



Goal:

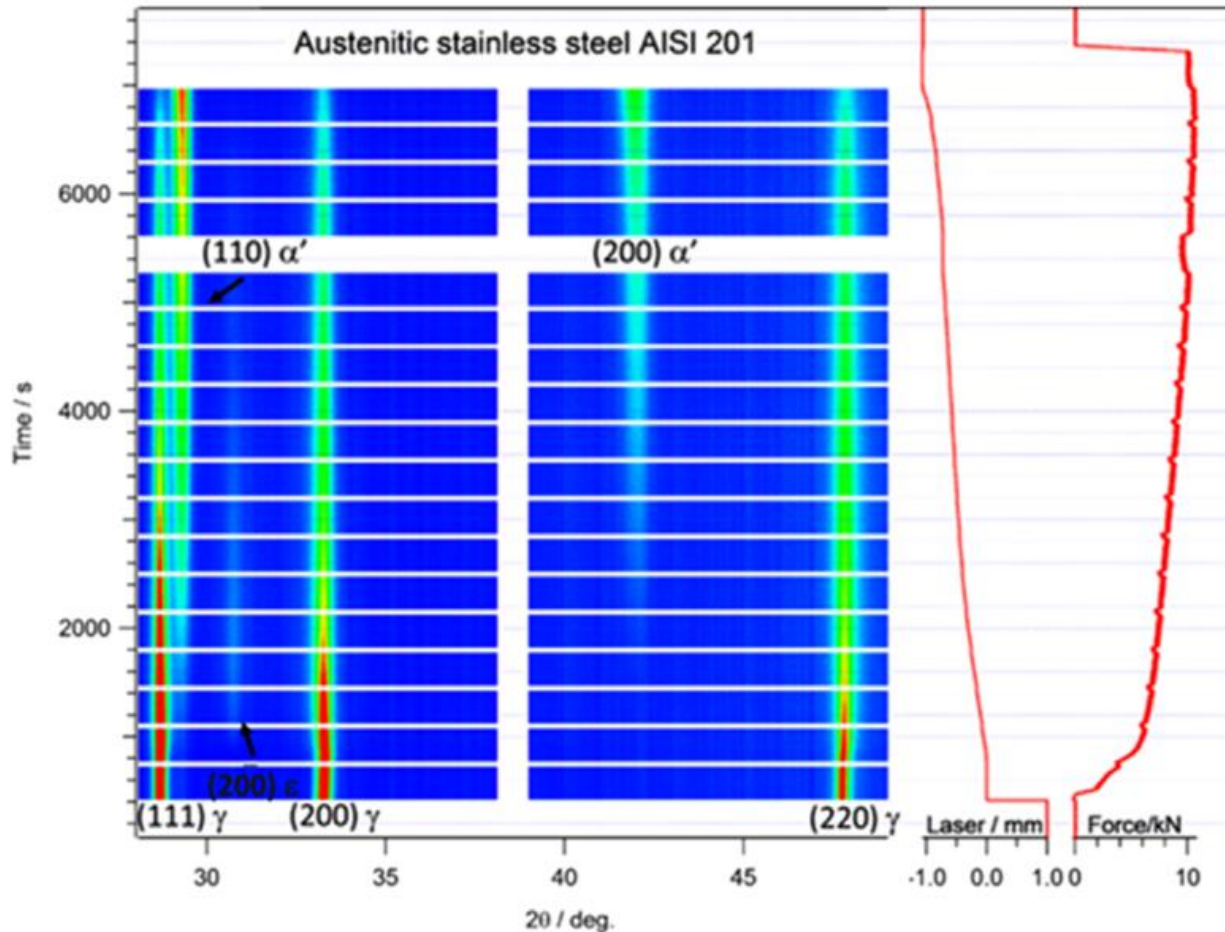
- To understand the changes in microstructure of AISI 201 by applying a tensile stress using in-situ XRD.
- To track phase transformation and strain partitioning among the phases

Chemical composition of the austenitic stainless steel AISI 201 used in this work. Values expressed in wt%.

C	Mn	Si	P	S	Cr	Ni	Mo	Al
0.0237	7.018	0.382	0.037	0.0014	17.06	4.07	0.0429	0.0047
Cu	Co	V	Nb	Ti	Sn	W	N	O
0.0717	0.0616	0.0408	0.0038	0.0041	0.0064	0.0147	0.1640	0.0029



In-situ synchrotron x-ray evaluation of strain-induced martensite in AISI 201 austenitic stainless steel during tensile testing

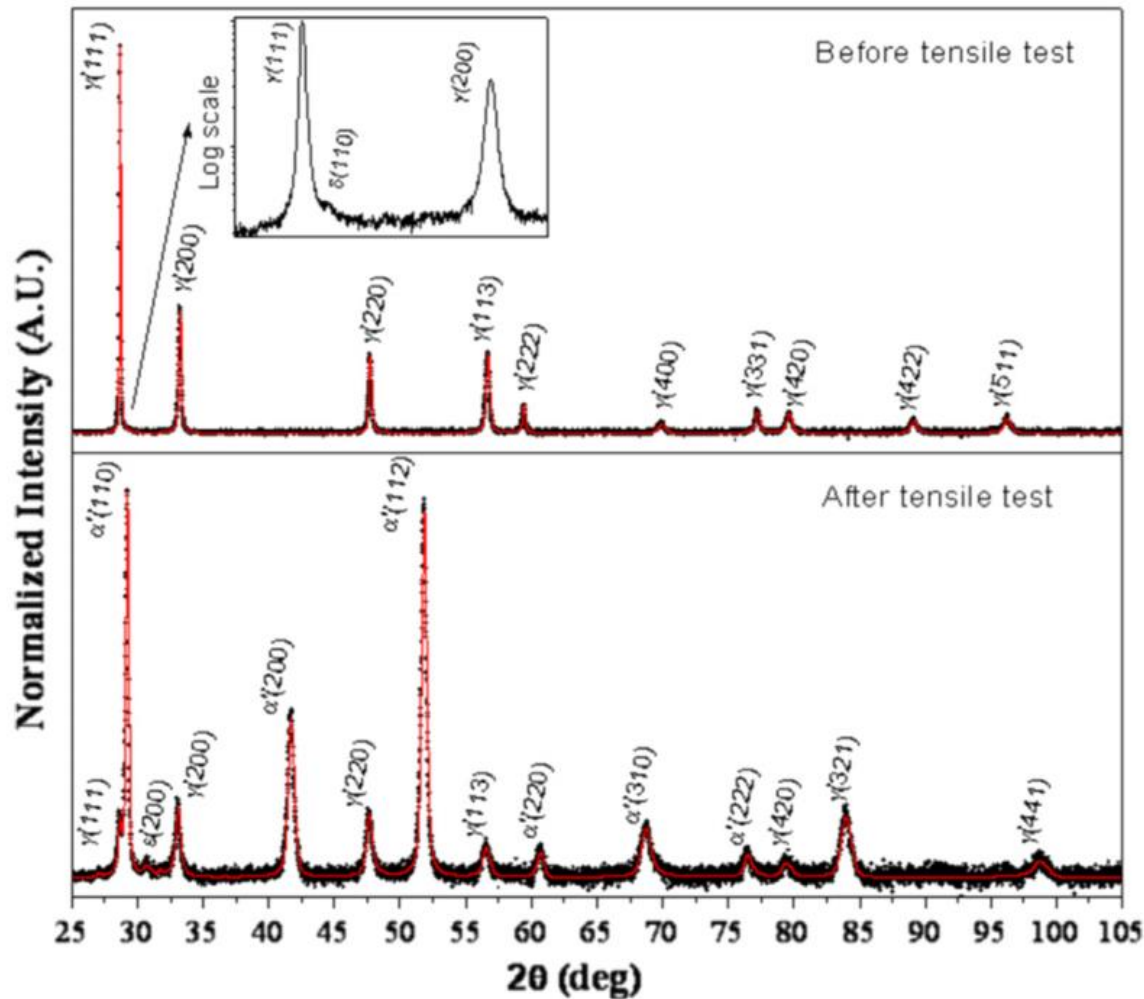


Metastable austenite (γ) decomposes into

- hcp- ϵ -martensite and
- bcc- α' -martensite

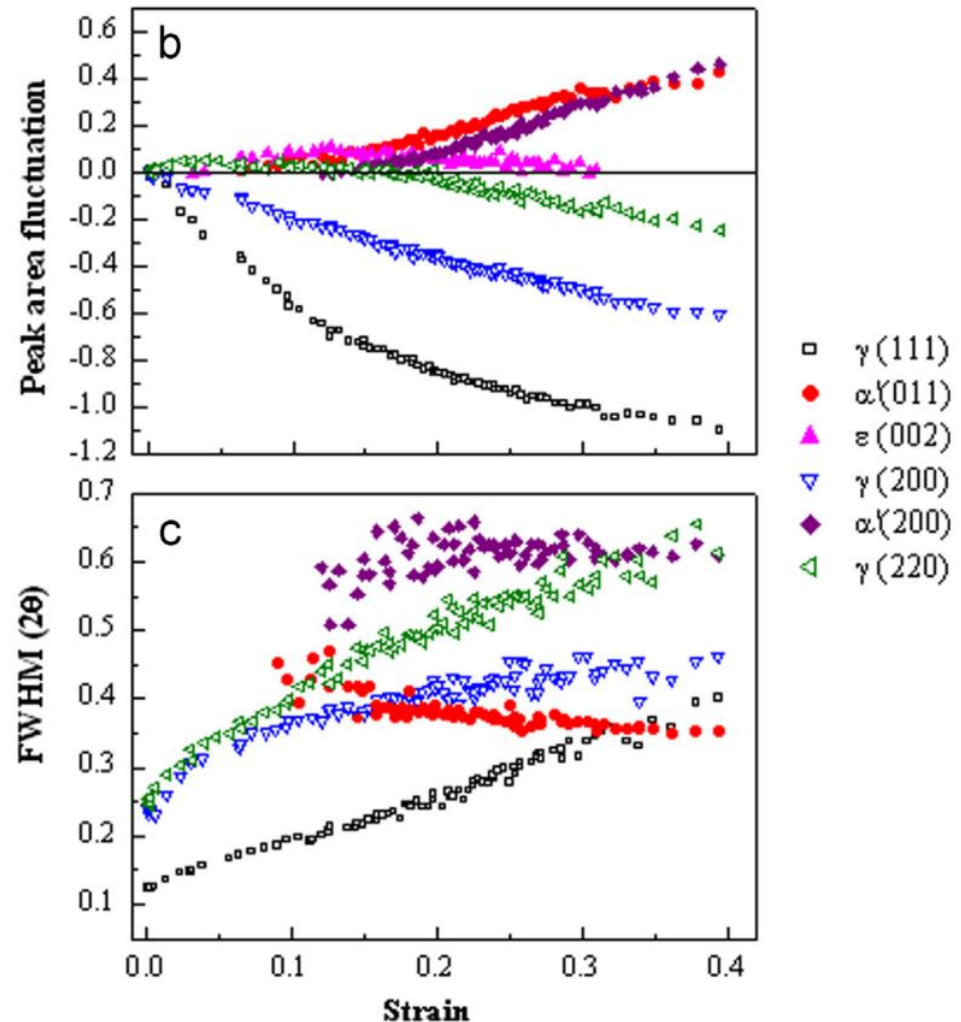
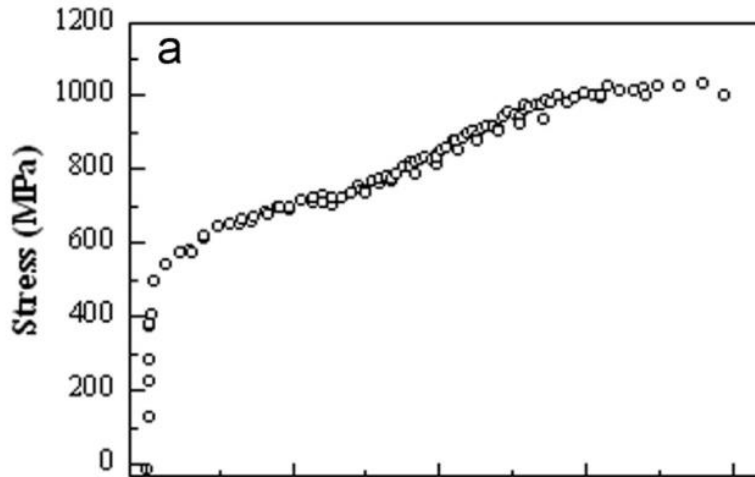


In-situ synchrotron x-ray evaluation of strain-induced martensite in AISI 201 austenitic stainless steel during tensile testing



- Metastable austenite (γ) decomposes into
- hcp- ϵ -martensite and
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In-situ synchrotron x-ray evaluation of strain-induced martensite in AISI 201 austenitic stainless steel during tensile testing

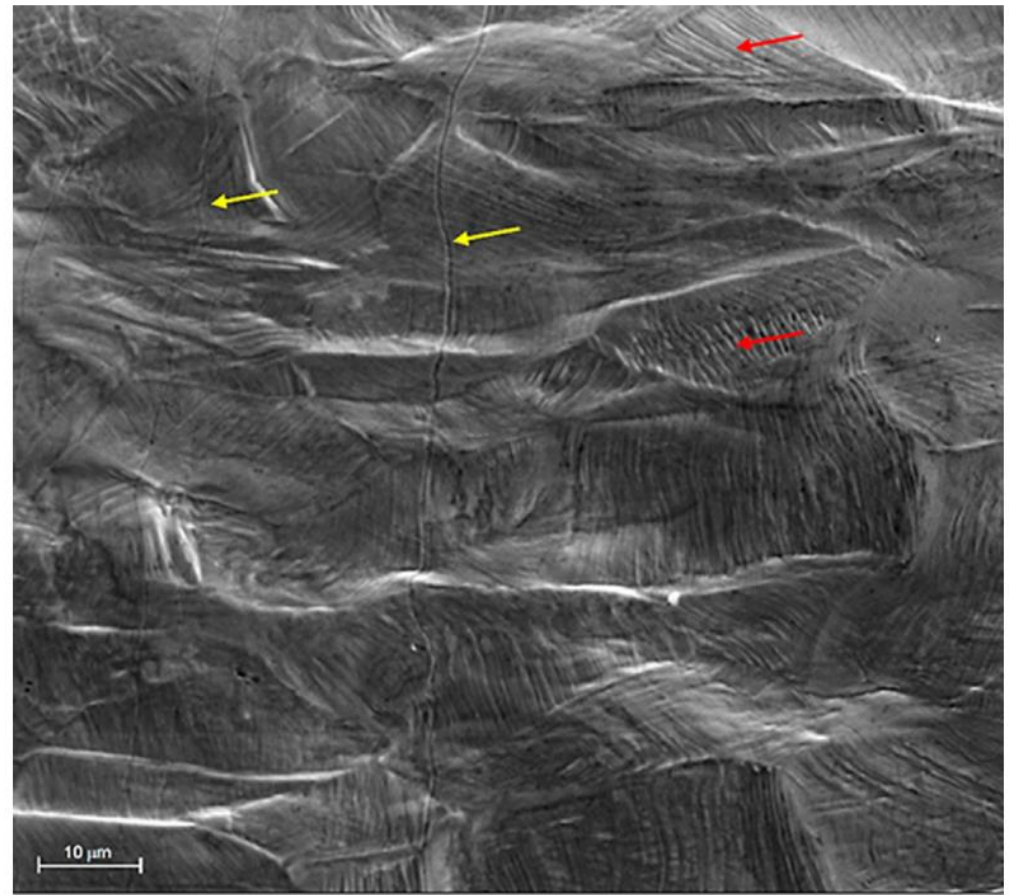
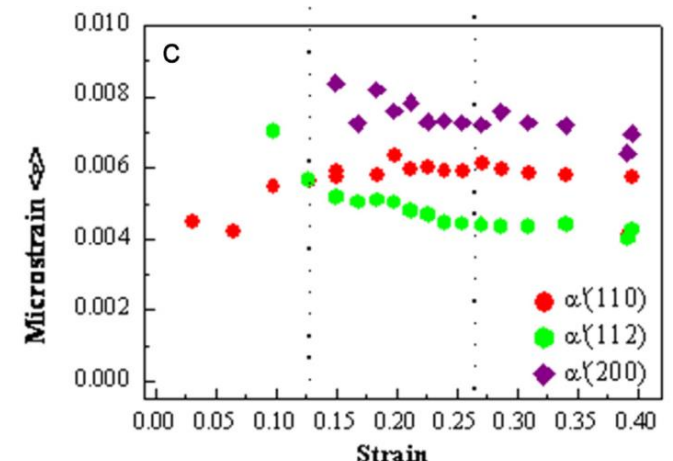
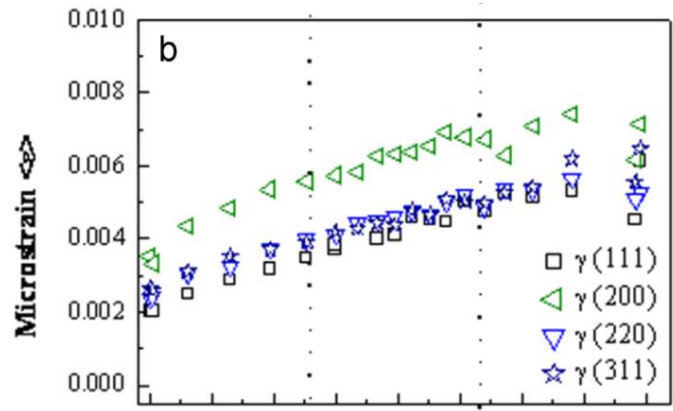
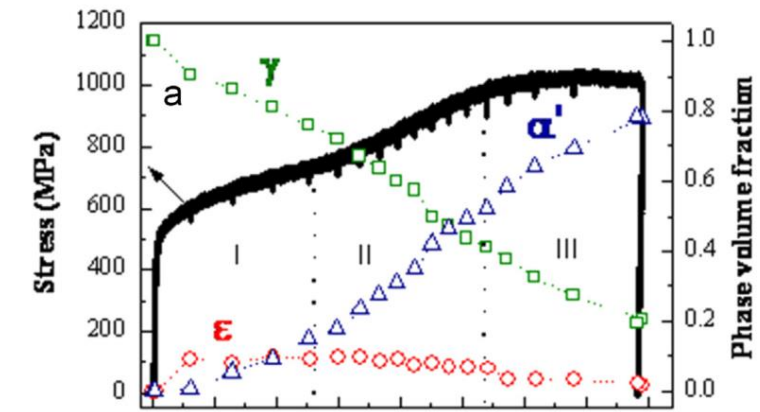


Data analysis:

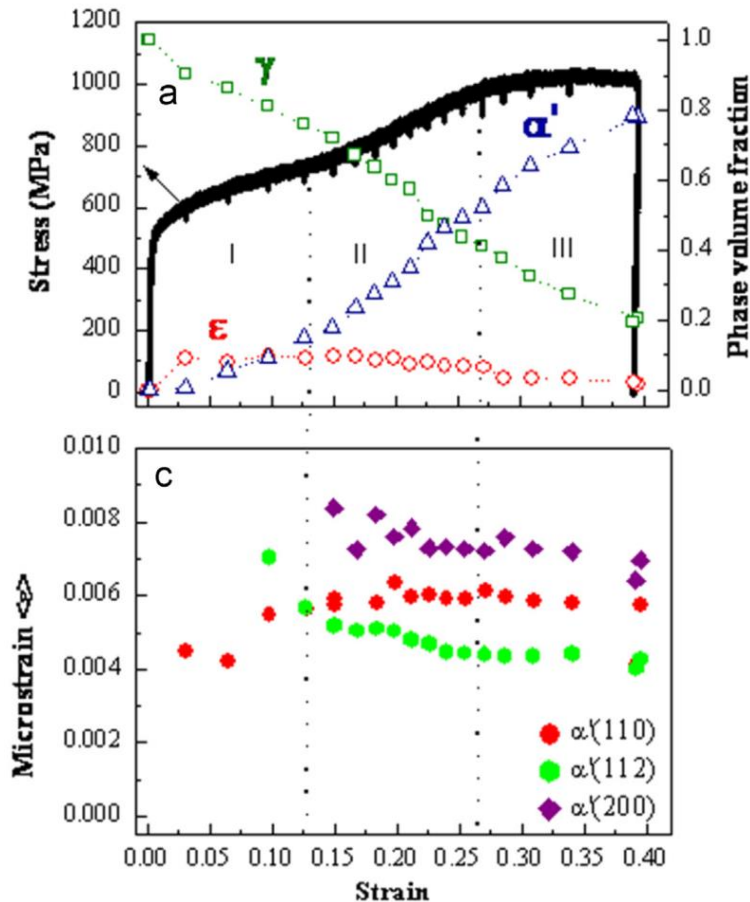
- ✓ Phase identification
- ✓ Integrated intensity
- ✓ Peak position
- ✓ FWHM



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In-situ synchrotron x-ray evaluation of strain-induced martensite in AISI 201 austenitic stainless steel during tensile testing

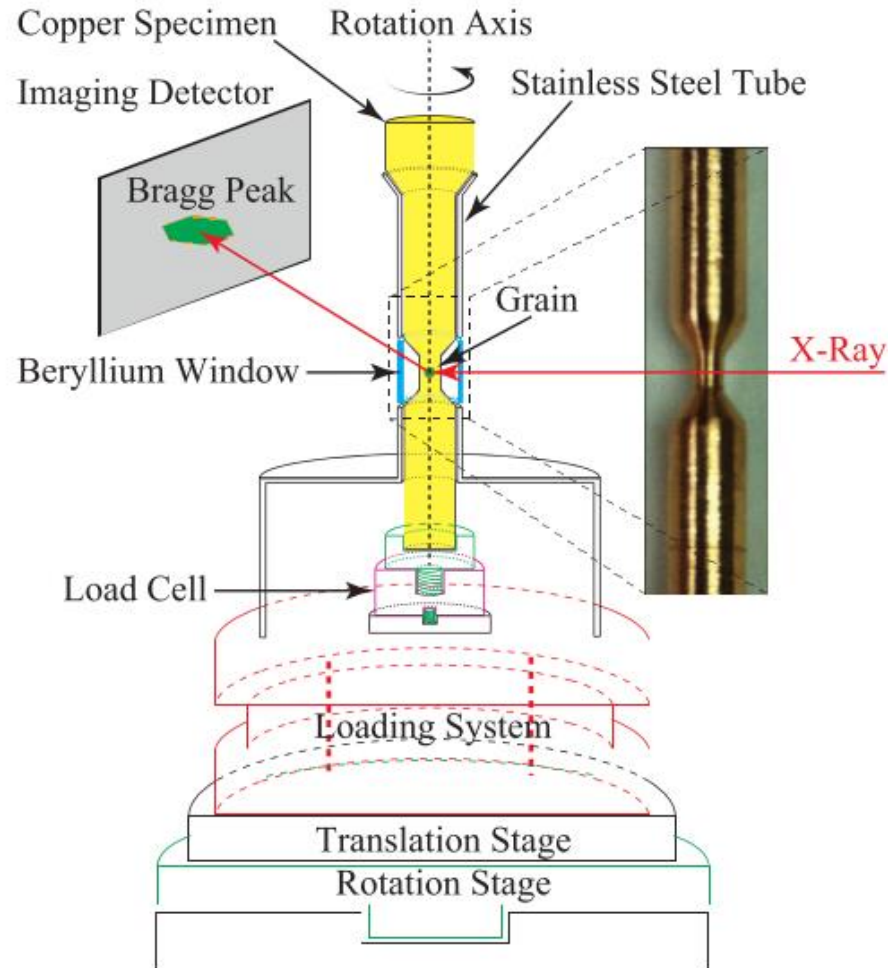


- ✓ The strain induced transformation of metastable gamma austenite (fcc structure) was followed in real time.
- ✓ ϵ -martensite is the first phase to appear followed by α' – martensite.
- ✓ Got information about the phase volume fractions and microstrain.
- ✓ FWHM of peaks is related to macroscopic mechanical properties. FWHM remains constant in the elastic regime and increases at the yield strength with the onset of plastic flow.



In-situ observation of bulk 3D grain evolution during plastic deformation in polycrystalline Cu

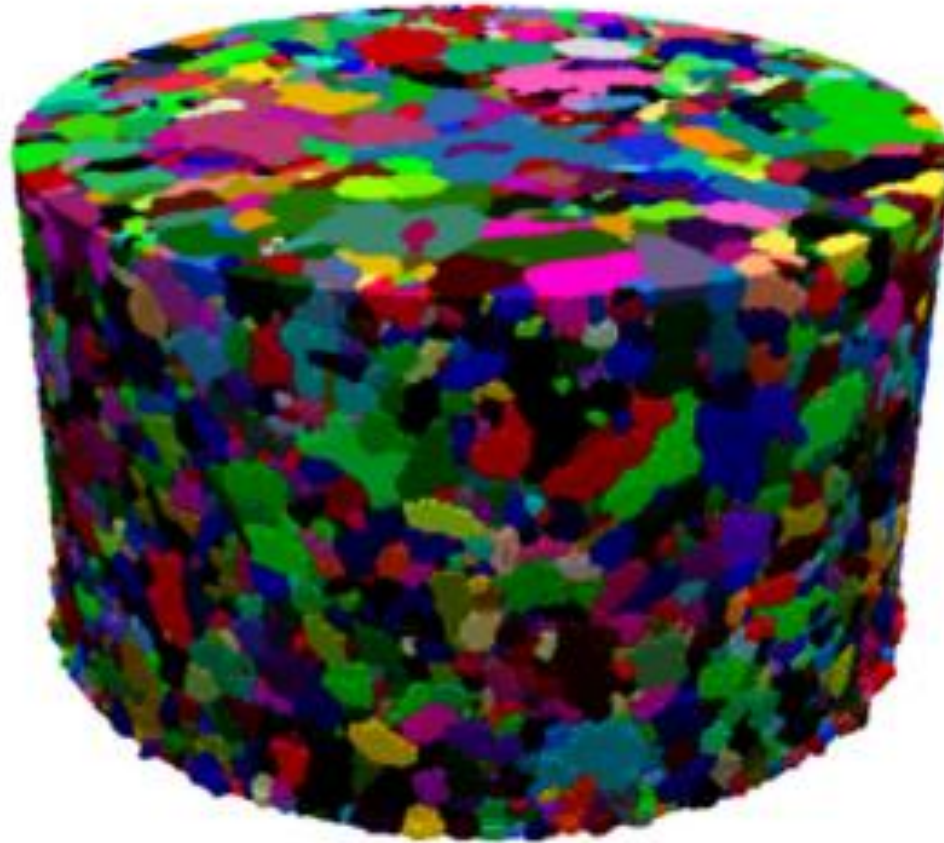
Reeju Pokharel^{a,c,*}, Jonathan Lind^{a,b}, Shiu Fai Li^b, Peter Kenesei^d, Ricardo A. Lebensohn^c, Robert M. Suter^a, Anthony D. Rollett^a



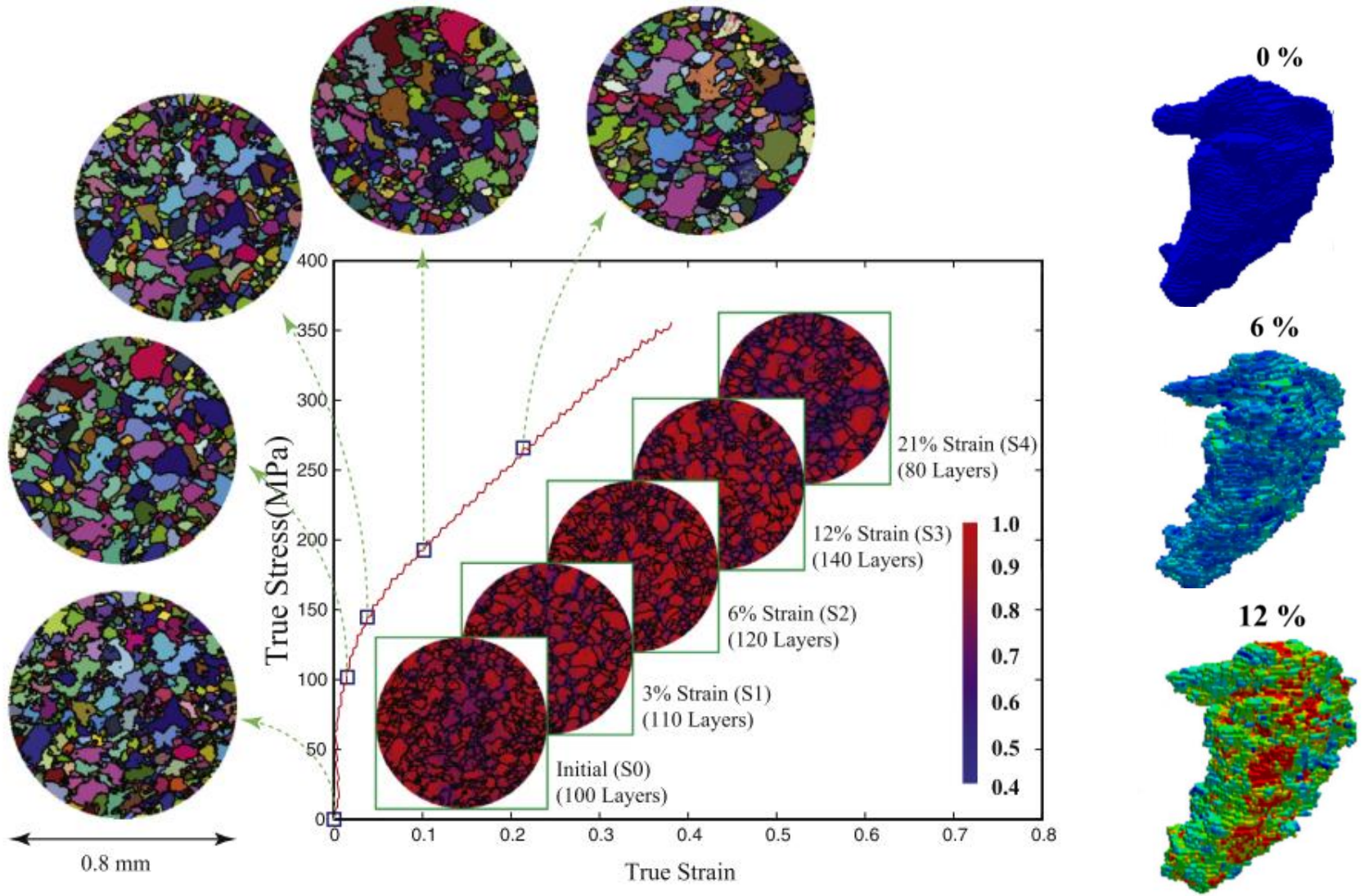
99.995% pure polycrystalline copper during tensile loading



In-situ observation of bulk 3D grain evolution during plastic deformation in polycrystalline Cu



In-situ observation of bulk 3D grain evolution during plastic deformation in polycrystalline Cu



Catalysts



Oil refinery catalytic reactor



Catalysis

Catalysts play an important part in many chemical processes. More than 85% of chemicals come from catalytic reactions.

Catalysts:

- increase the rate of reaction
- are not consumed by the reaction
- are only needed in very small amounts

In-situ and operando catalytic experiments include

- Temperature and pressure control
- Structural characterization of intermediate compounds
- Gases in/out, flow control, analysis of gases out of the catalytic reaction

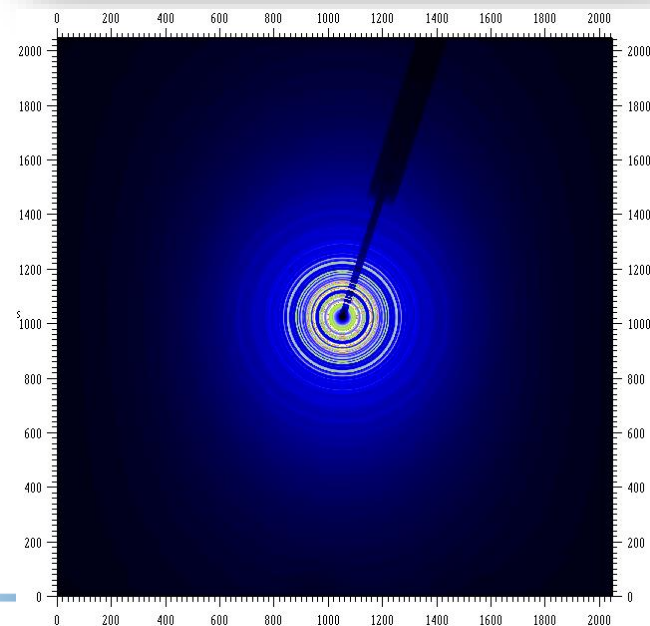
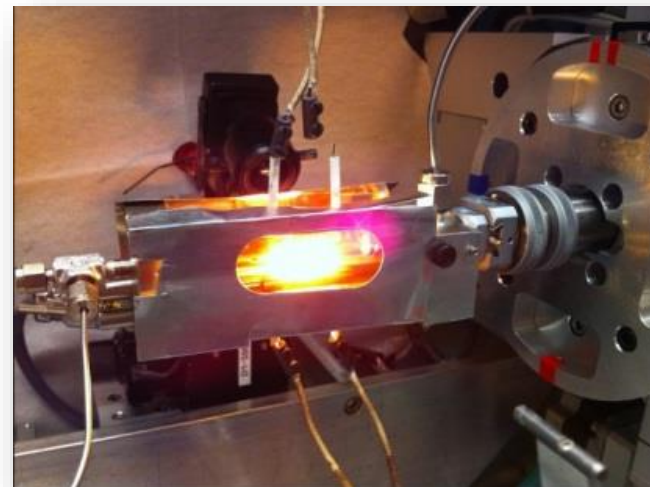
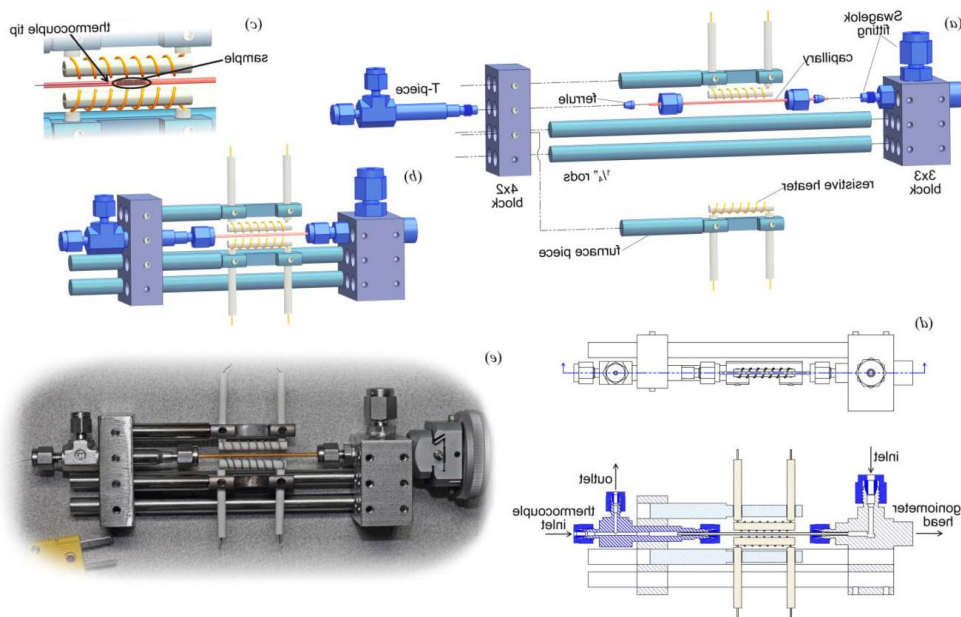


In-situ catalytic experiments

A versatile sample-environment cell for non-ambient X-ray scattering experiments

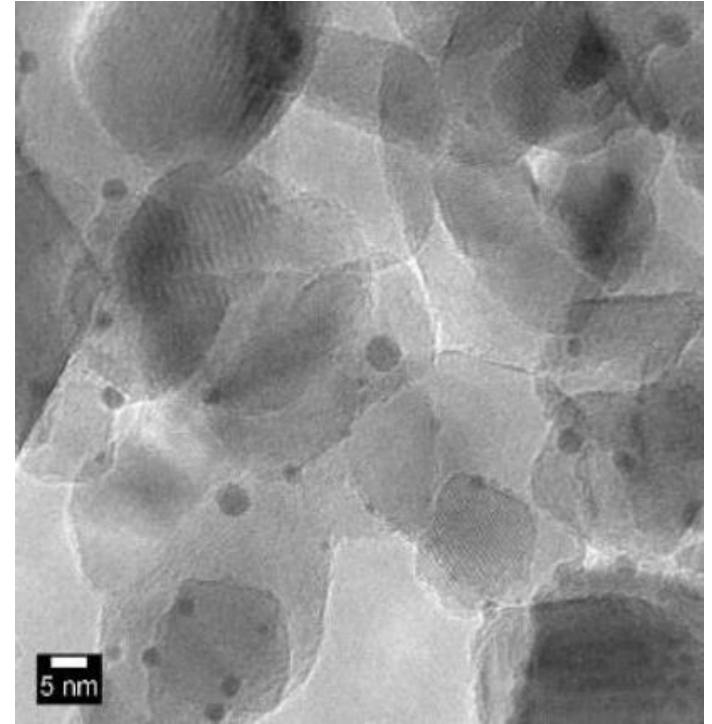
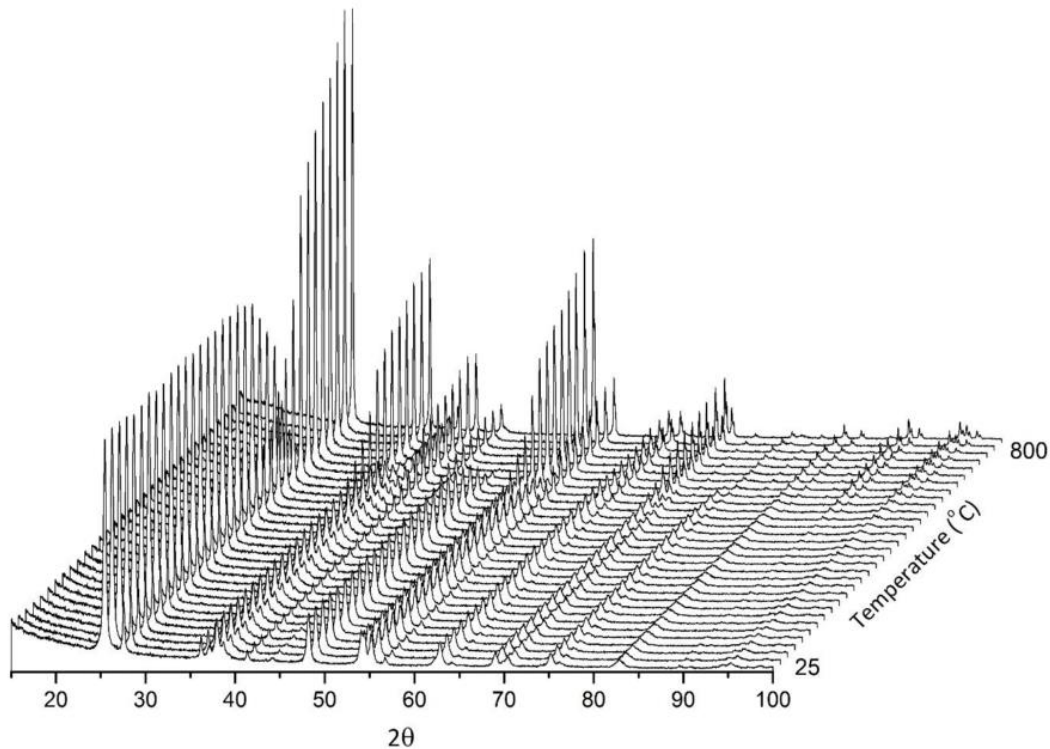
Peter J. Chupas,^{a*} Karena W. Chapman,^a Charles Kurtz,^a Jonathan C. Hanson,^b
Peter L. Lee^a and Clare P. Grey^c

J. Appl. Cryst. (2008). **41**, 822–824



In-situ catalysis

Commercial catalyst Aurolite[®] (1% Au-P25)



Phase change from anatase to the thermodynamically stable rutile phase



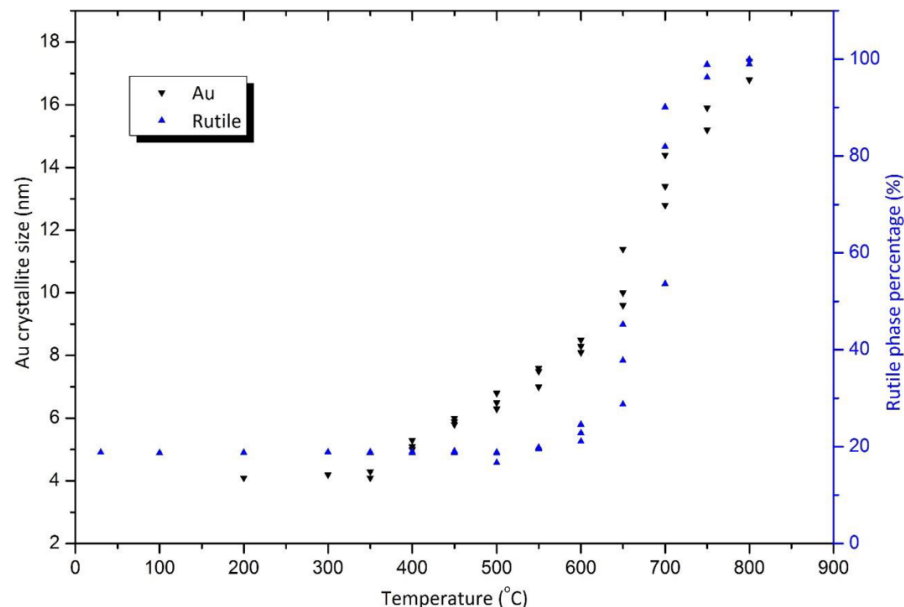
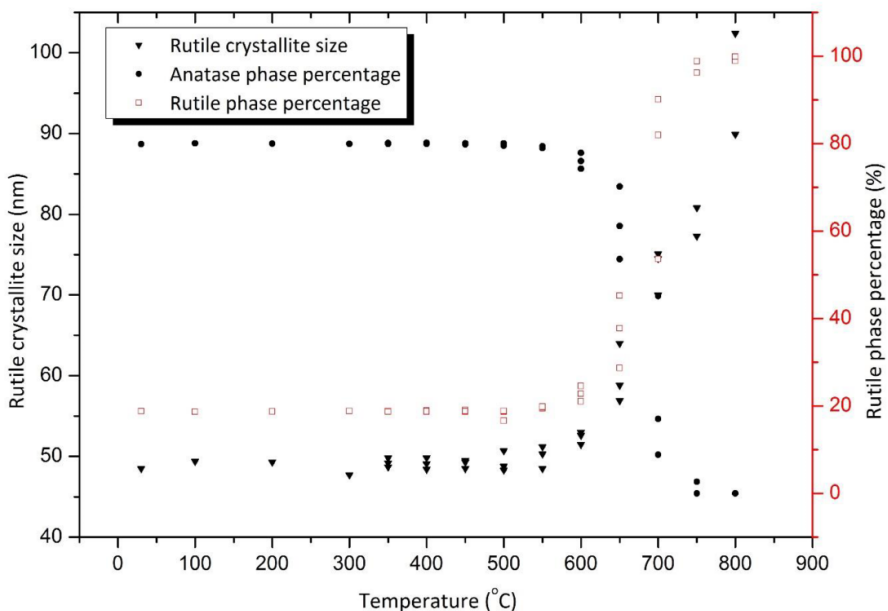
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Centre canadien de rayonnement synchrotron

Achieving nano-gold stability through rational design†
D. Barret et al. 2016



In-situ catalysis

Commercial catalyst Aurolite[®] (1% Au-P25)



Structural instability of the support is a major factor in Au-nanoparticle growth
→ catalytic activity decreases

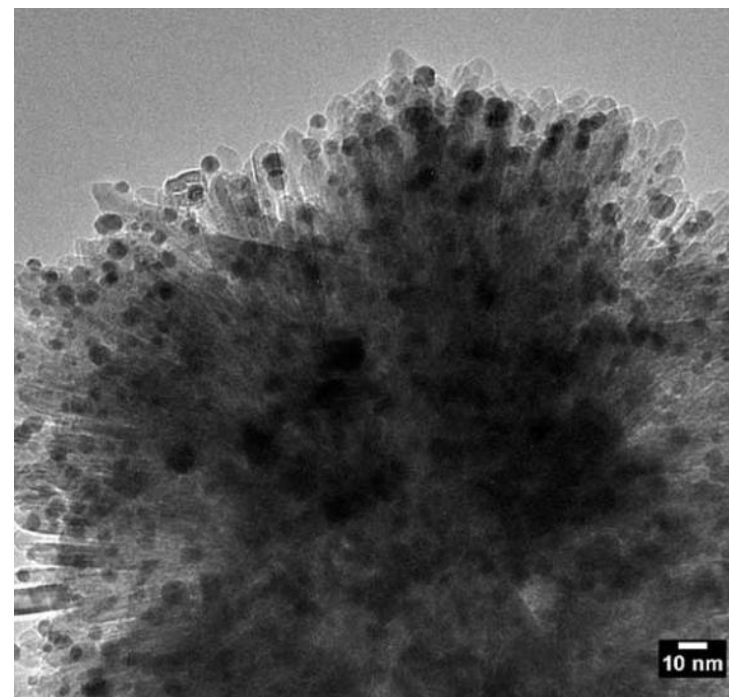
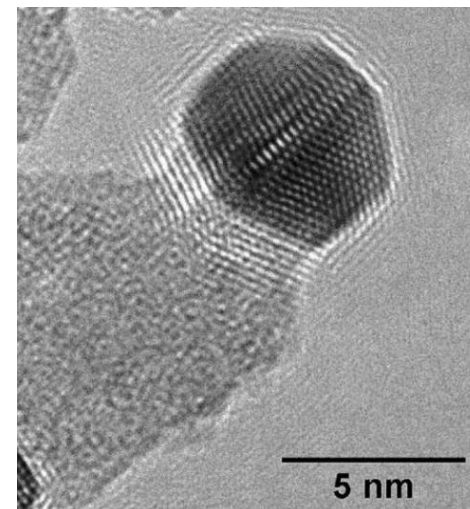
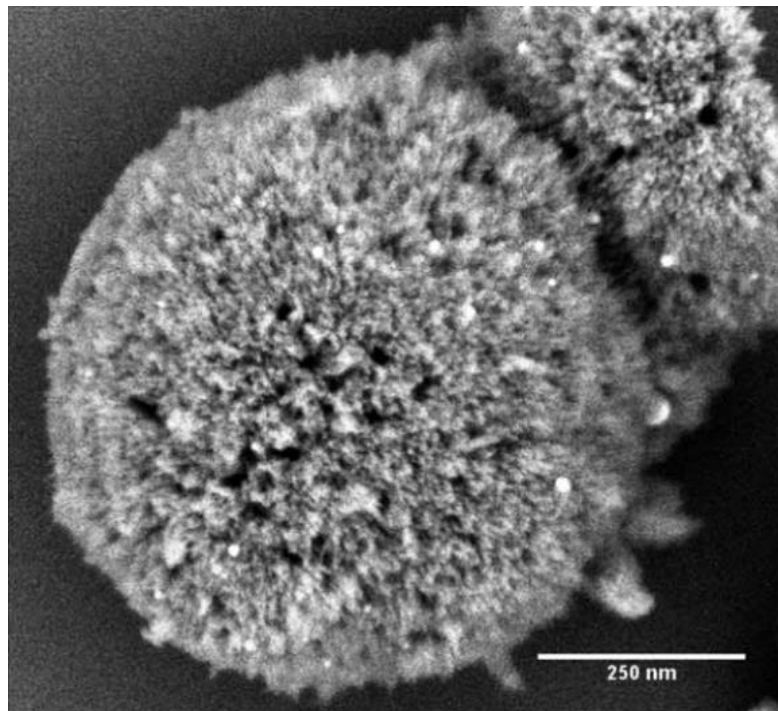


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In-situ catalysis



Achieving nano-gold stability through rational design†

D. Barret et al.

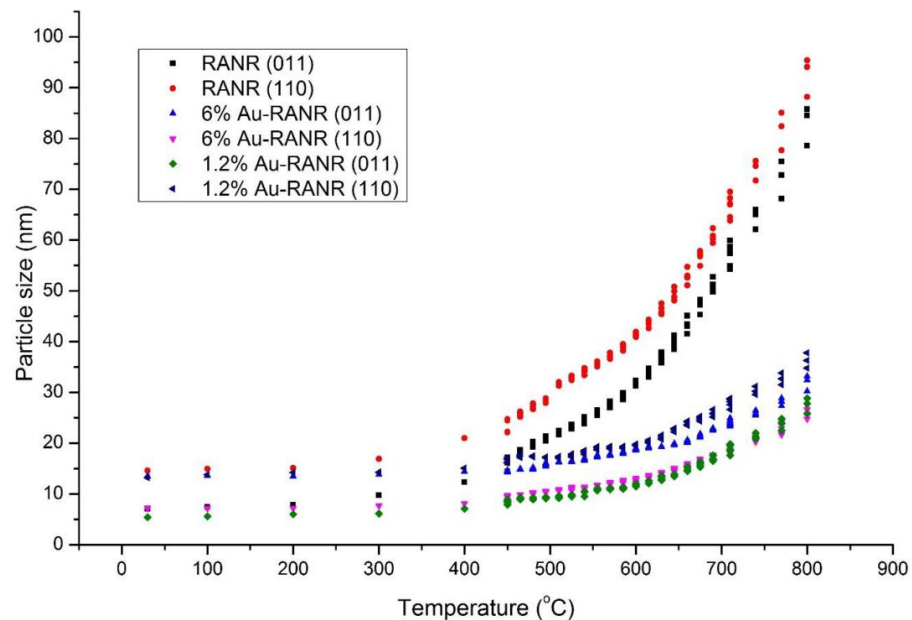
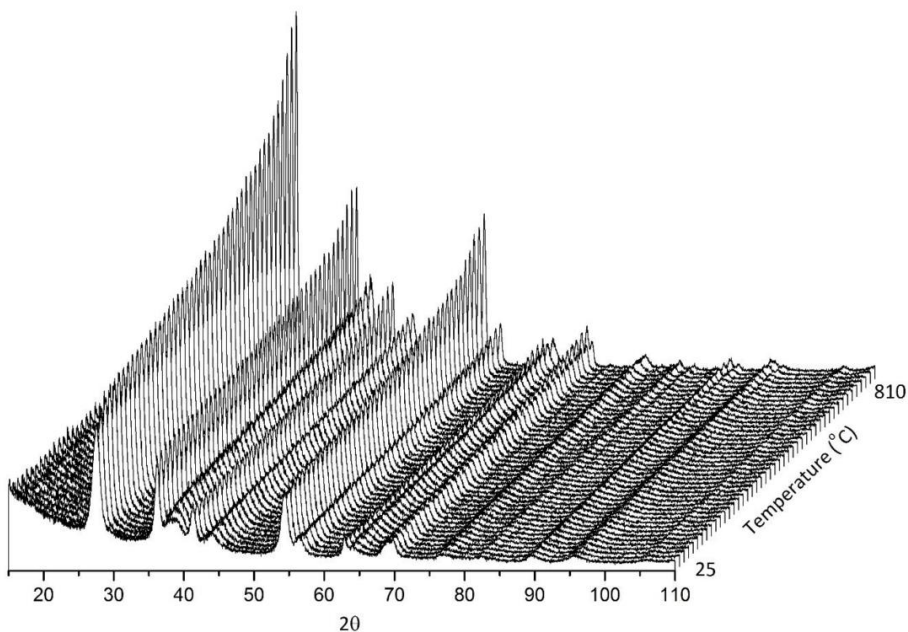


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In-situ catalysis

The presence of Au resulted in a stabilizing effect with regards to the growth of the support structure.



Achieving nano-gold stability through rational design†

D. Barret et al.



In-situ catalysis

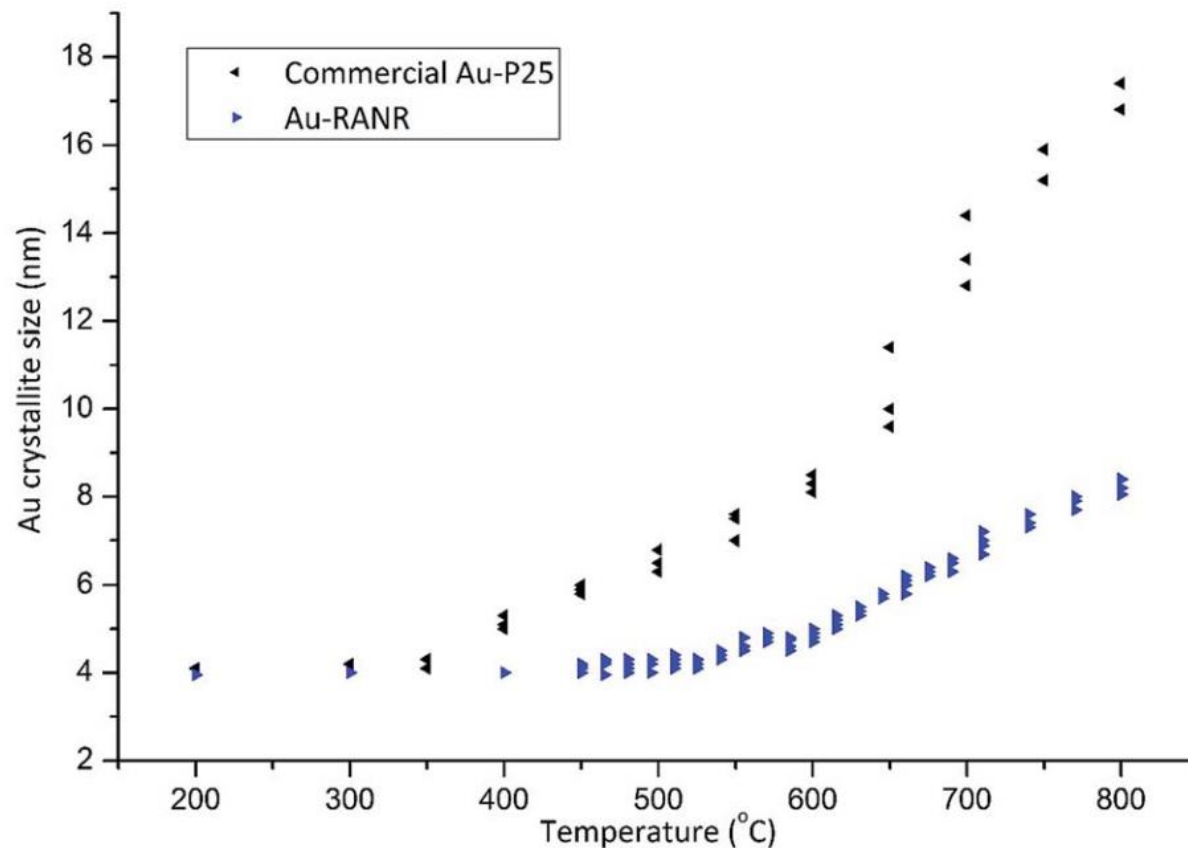


Fig. 3 Au crystallite sizes of the commercial Au-TiO₂ and Au-RANR catalysts determined from Rietveld refinement from *in situ* PXRD.



In-situ catalysis

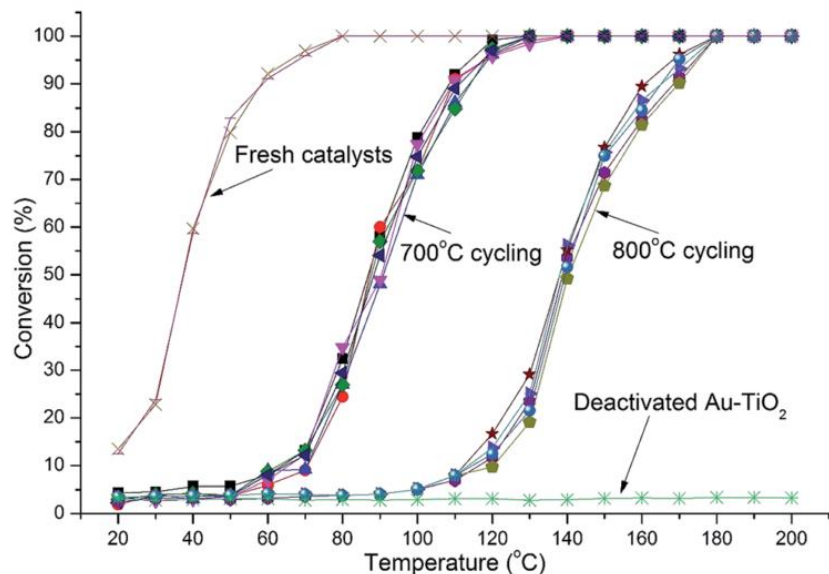


Fig. 4 Light-off curves for catalysts with 1.2% Au-RANR and commercial Au-TiO₂ after multiple 700 and 800 °C heating cycles (10 cycles in total).

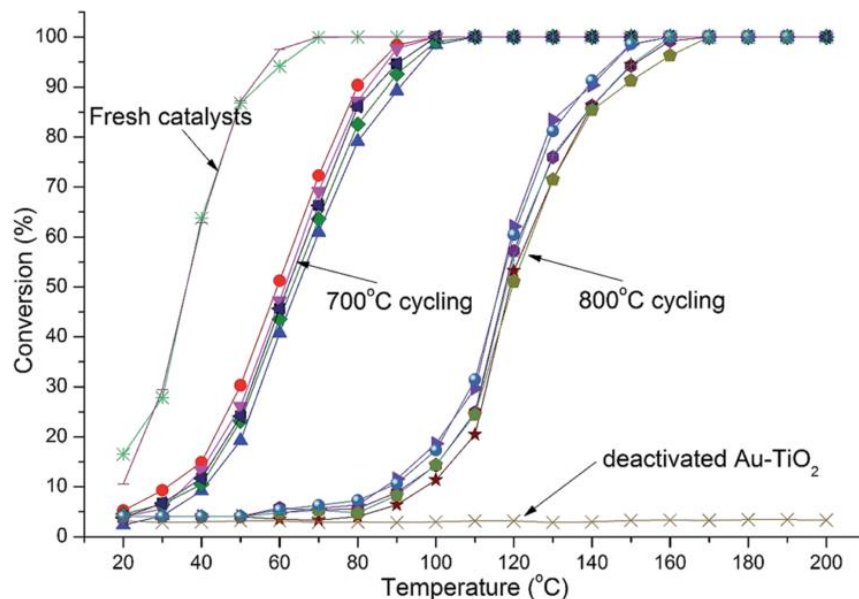


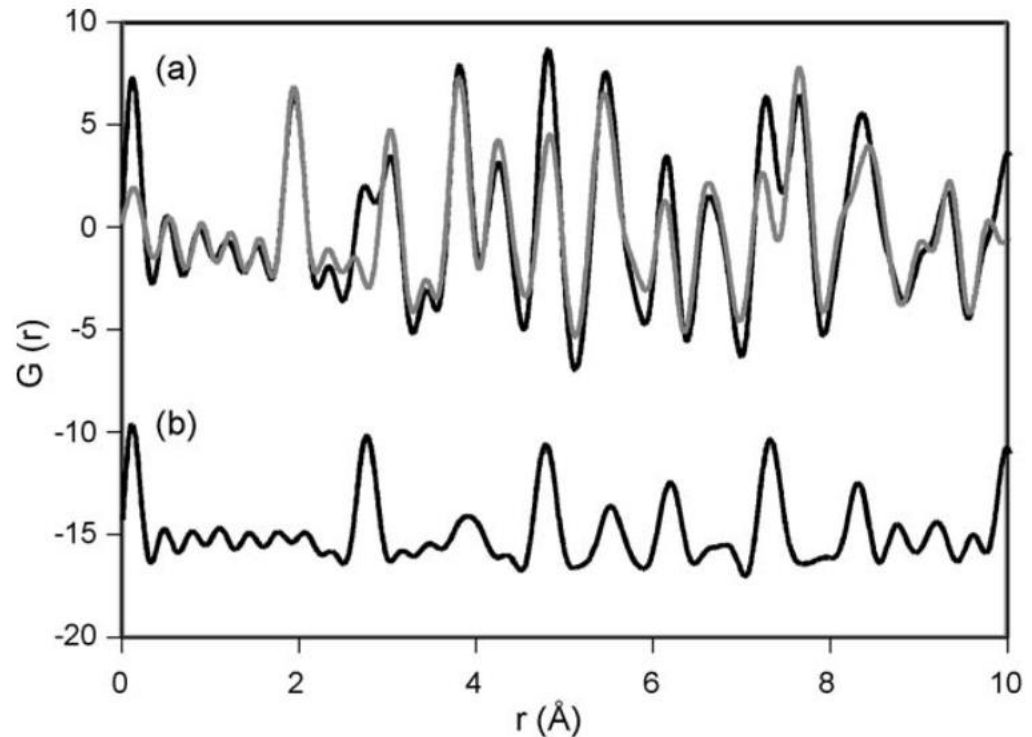
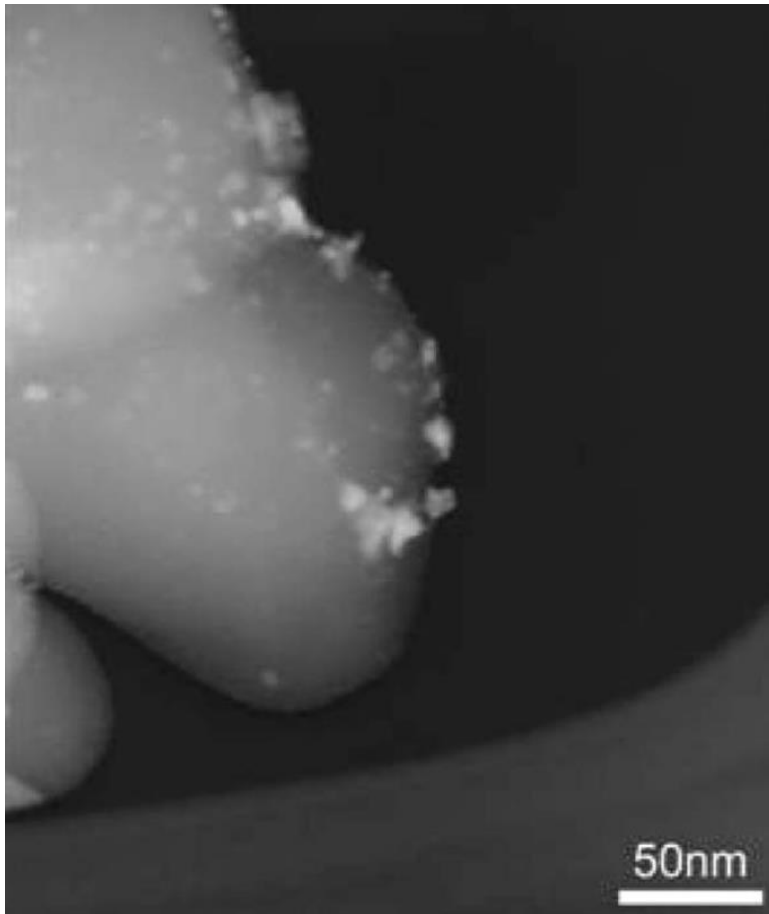
Fig. 5 Light-off curves for catalysts with 5% Au-RANR and commercial Au-TiO₂ catalyst after multiple 700 and 800 °C heating cycles (10 cycles in total).

- ✓ Thermodynamically stable support material
- ✓ Improved morphology
- ✓ Au nanoparticles sit isolated on the rod tips -> reduced mobility and coalescence
- ✓ Remarkable catalytic stability tested with CO oxidation



Application of high-energy X-rays and Pair-Distribution-Function analysis to nano-scale structural studies in catalysis

Peter J. Chupas^{a,*}, Karena W. Chapman^a, Hailong Chen^b, Clare P. Grey^b

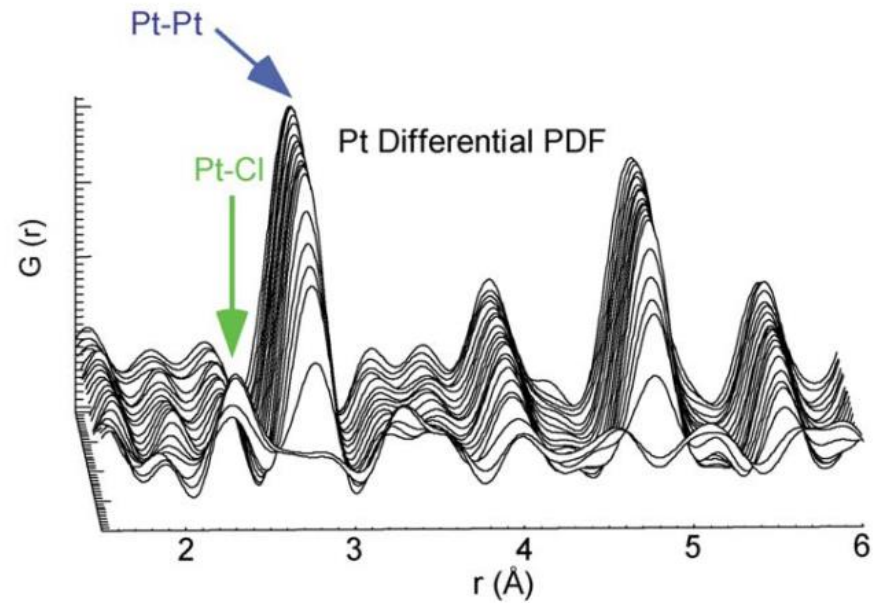
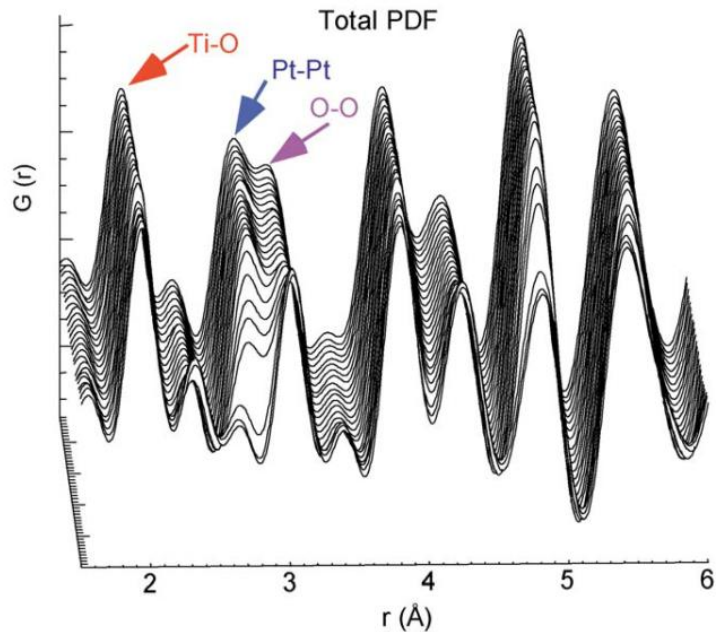


- (a) Grey line: $G(r)$ for TiO_2 support
Black line: $G(r)$ for 2.5% Pt on TiO_2 calcined under H_2 flow
- (b) Differential PDF



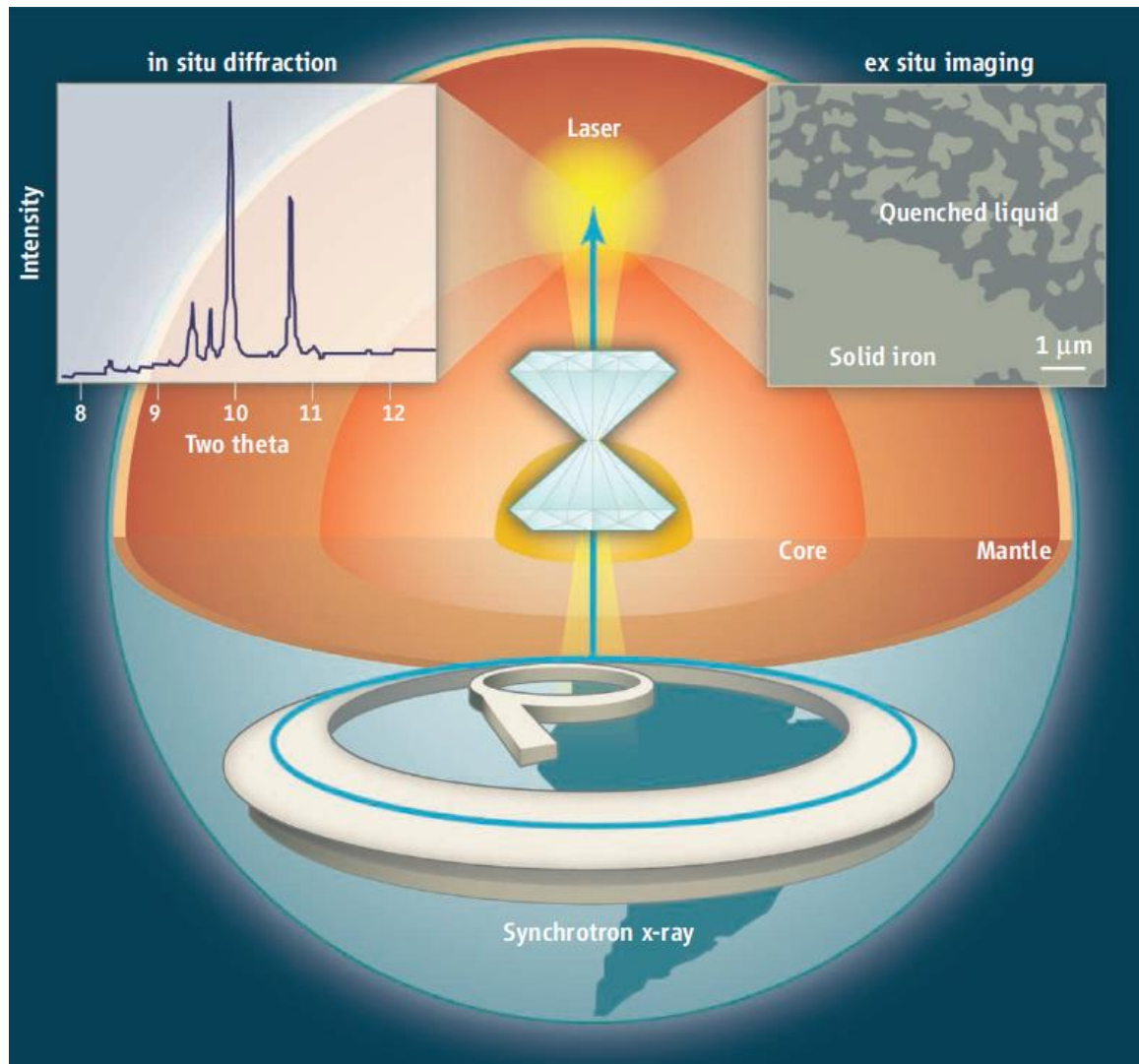
Application of high-energy X-rays and Pair-Distribution-Function analysis to nano-scale structural studies in catalysis

Peter J. Chupas^{a,*}, Karena W. Chapman^a, Hailong Chen^b, Clare P. Grey^b

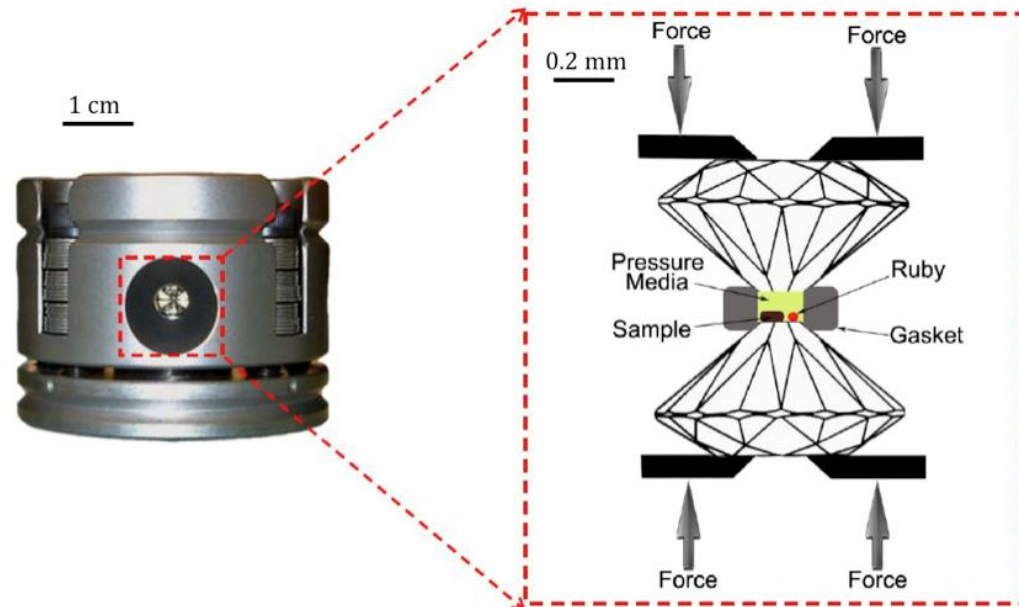
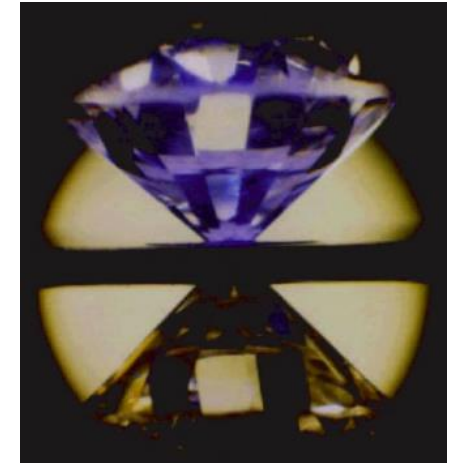
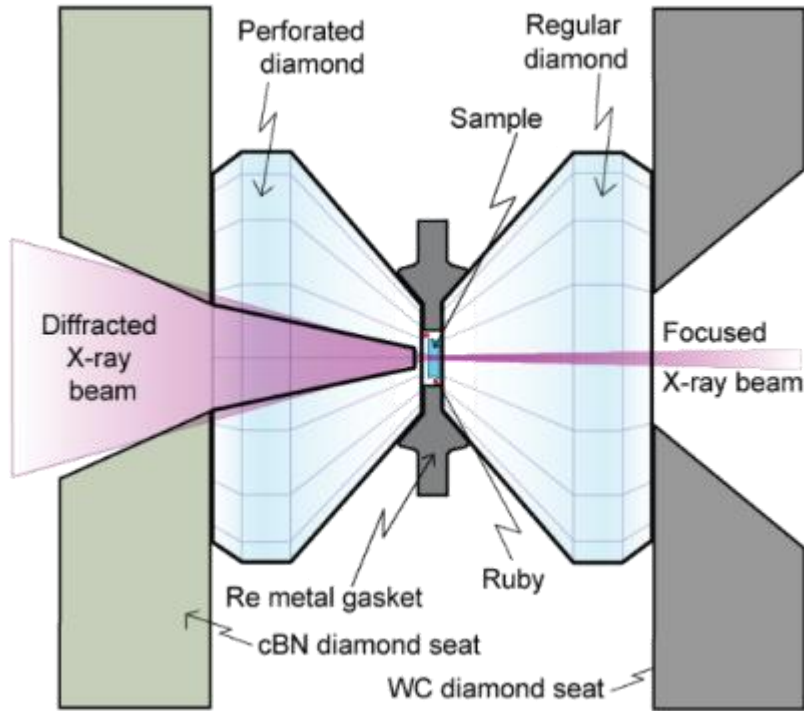


1. Pt nano-particles examined with atomic resolution
2. In-situ transformation of Pt^{4+} in PtCl_6^{2-}
3. Pt^{4+} reduced in situ with H_2 forming metallic fcc Pt nano particles
4. Observation of the Pt-O bond yields insight about catalyst interaction with TiO_2 support
5. Initial nano-particles are $\sim 1\text{nm}$, while by 200°C they are larger and more crystalline
6. Suggests agglomeration of smaller particles to form larger particles

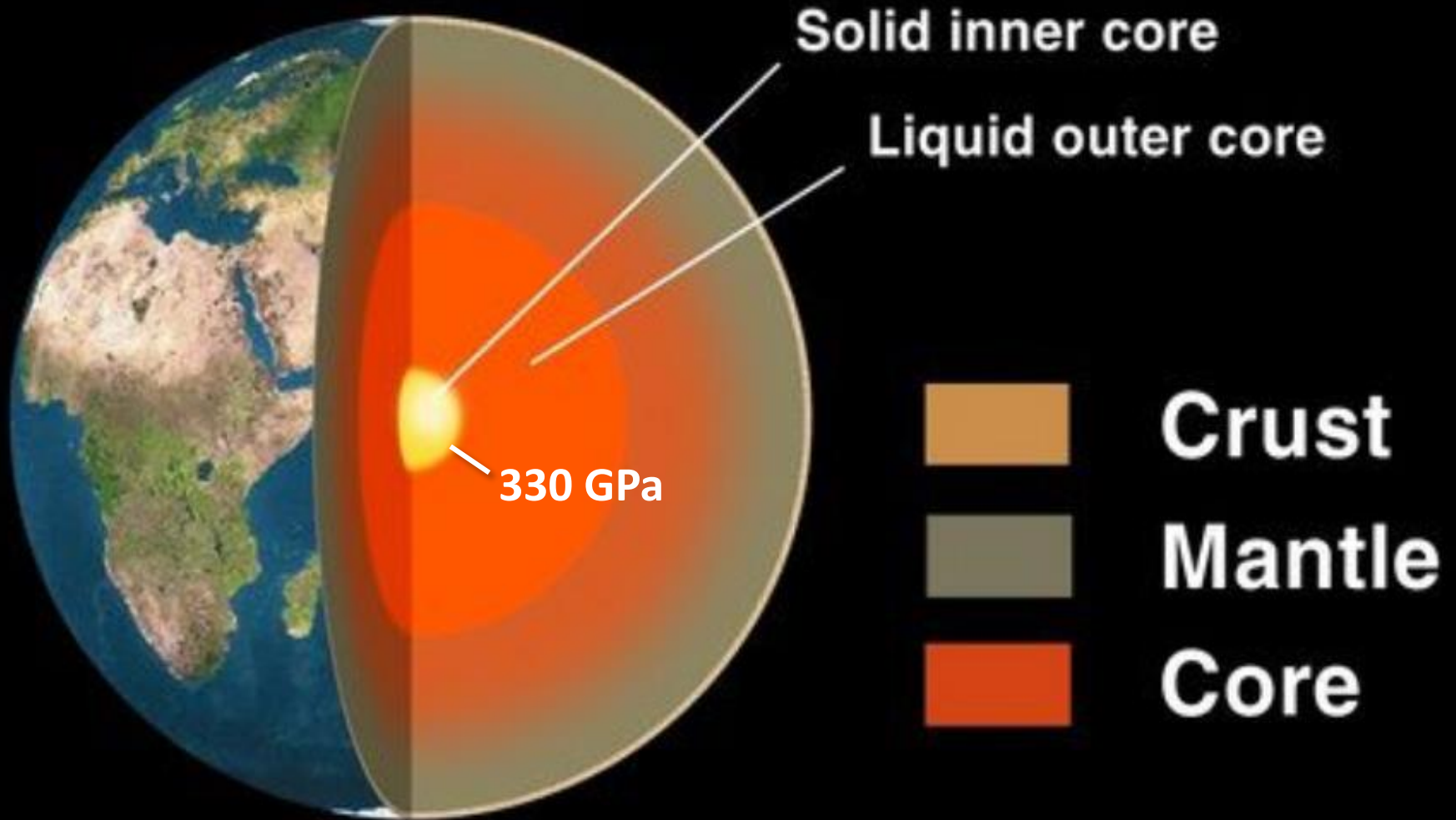
High pressure studies



High pressure studies



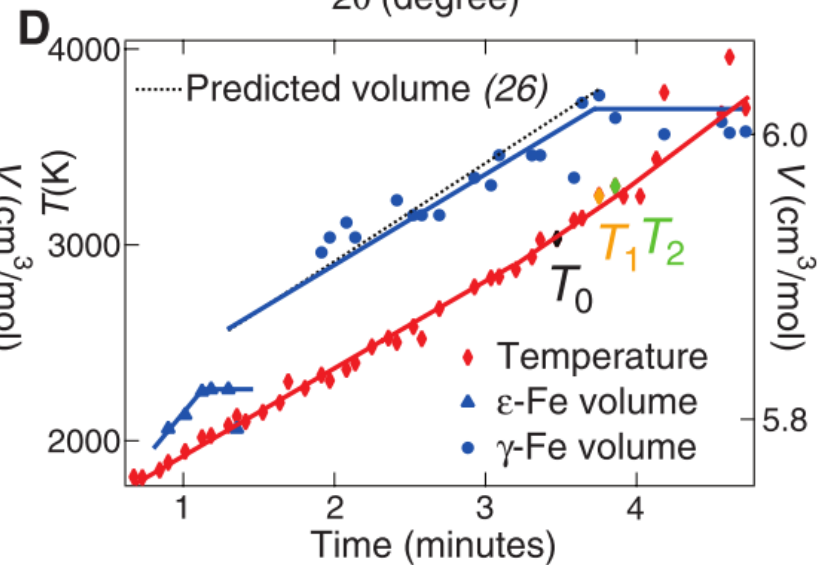
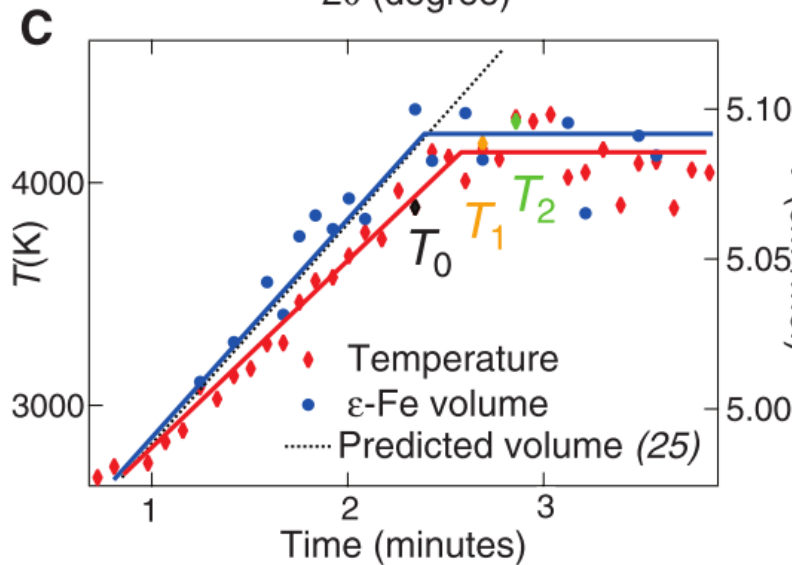
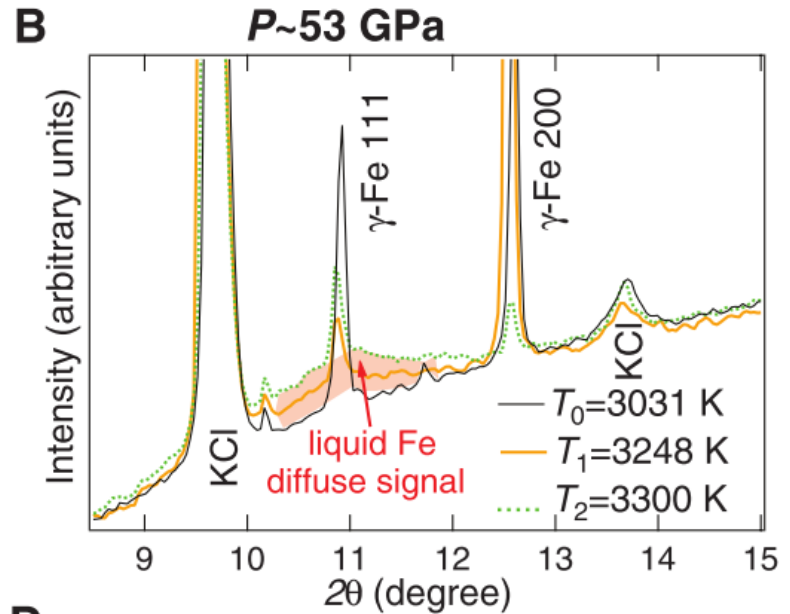
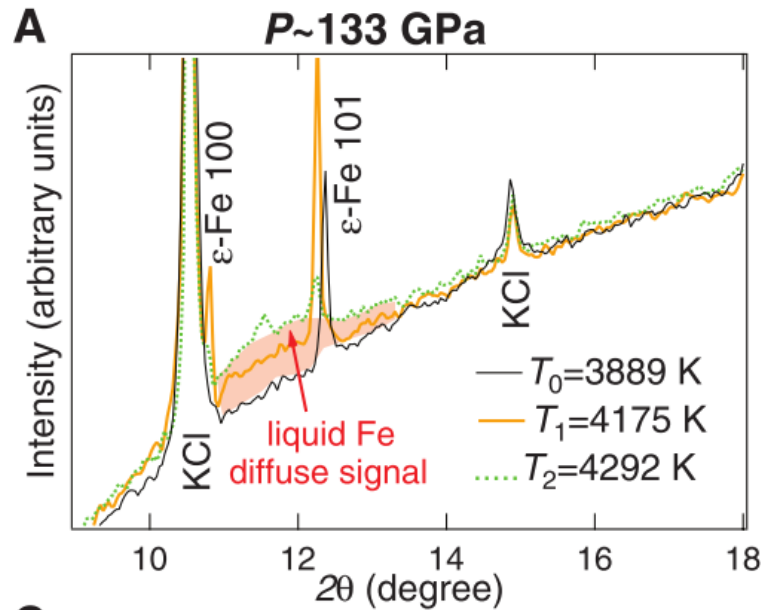
Earth's core is structured in a solid inner core, mainly composed of iron, and a liquid outer core.



How does iron behave at these extreme temperatures and pressures?

Melting of Iron at Earth's Inner core Boundary based on Fast X-ray Diffraction

Static laser-heated diamond anvil cell experiments up to 200 GPa



4000 K = 3727 °Celsius

1 GPa = 9869 atm

Science **340**, 464, 2013. Anzellini et al.



Melting of Iron at Earth's Inner core Boundary based on Fast X-ray Diffraction

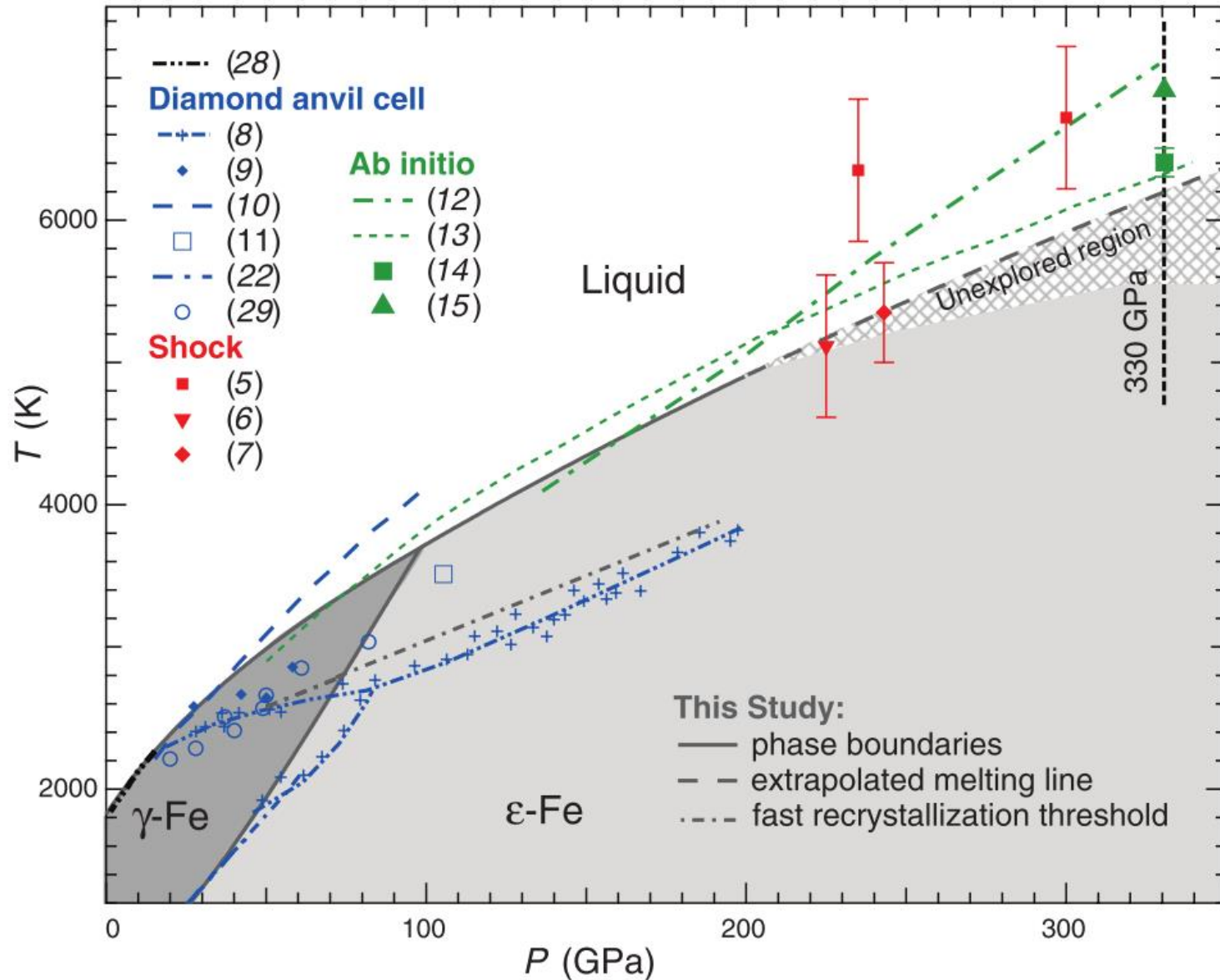
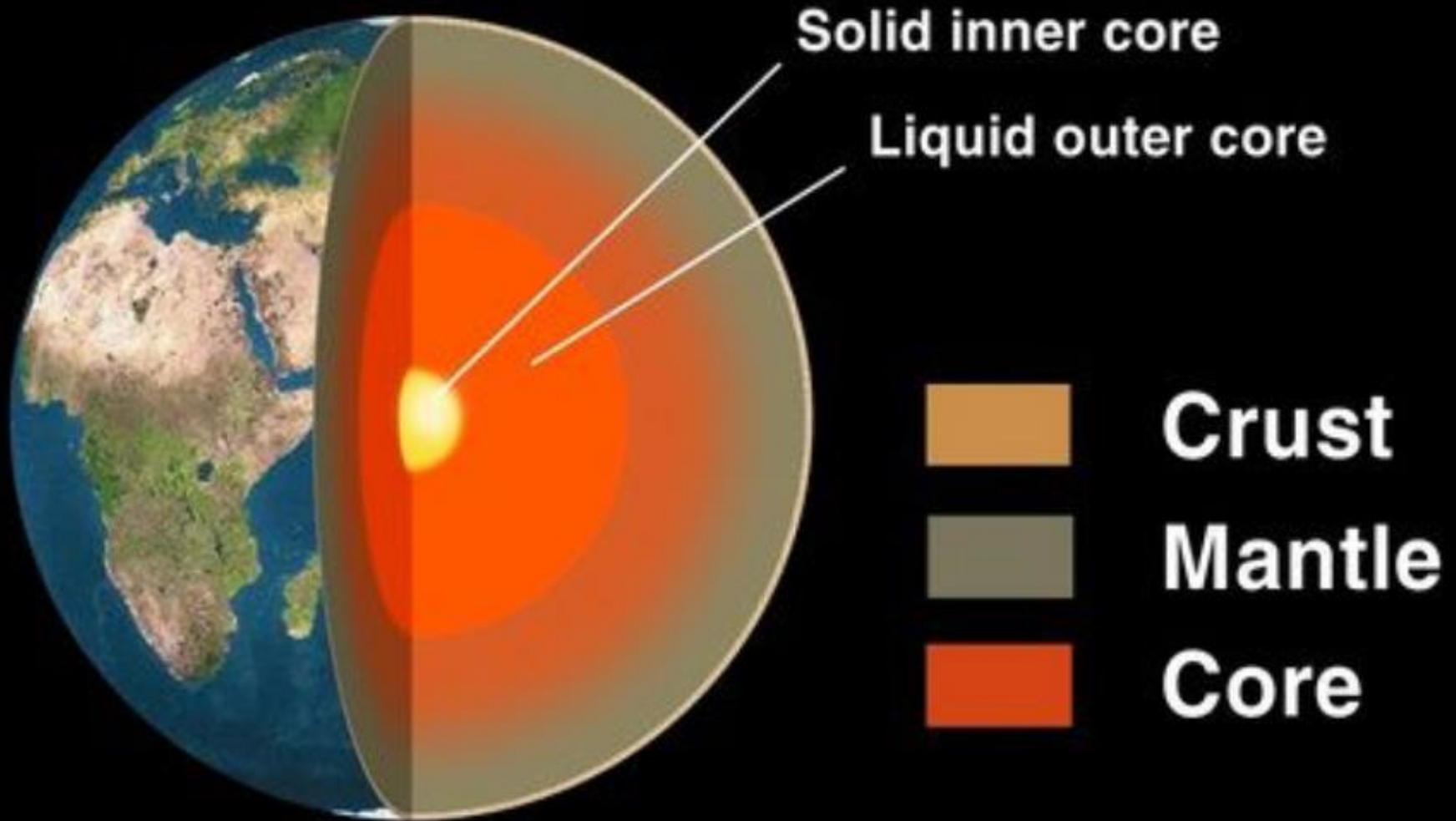


Fig. 3. Phase stability domains for Fe obtained in the literature and in this study. The stability field for ϵ -Fe is based on the current study data and data from (19).



Earth's core is structured in a solid inner core, mainly composed of iron, and a liquid outer core.



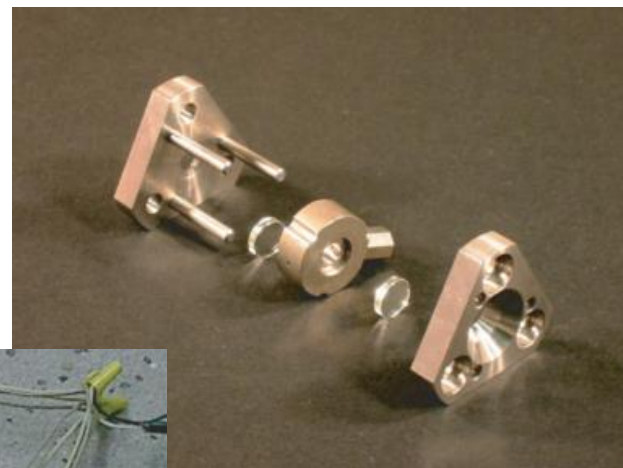
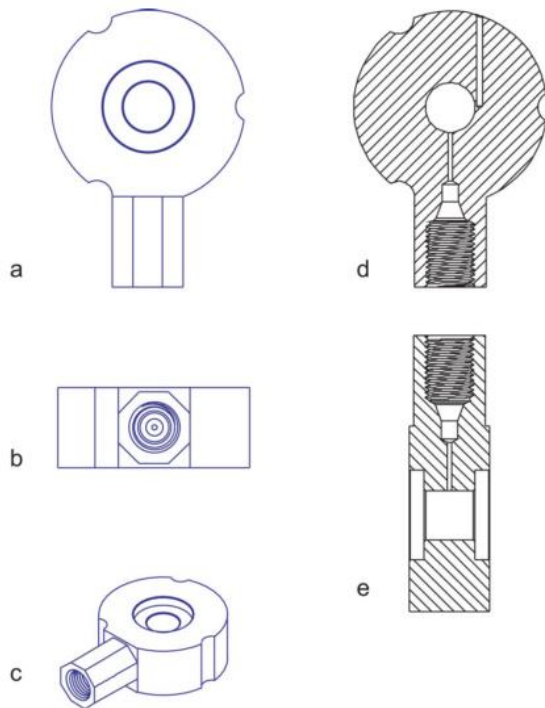
we conclude that the melting temperature of iron at the inner core boundary is 6230 ± 500 kelvin. This estimation favors a high heat flux at the core-mantle boundary with a possible partial melting of the mantle.

1811 K
at 1 bar

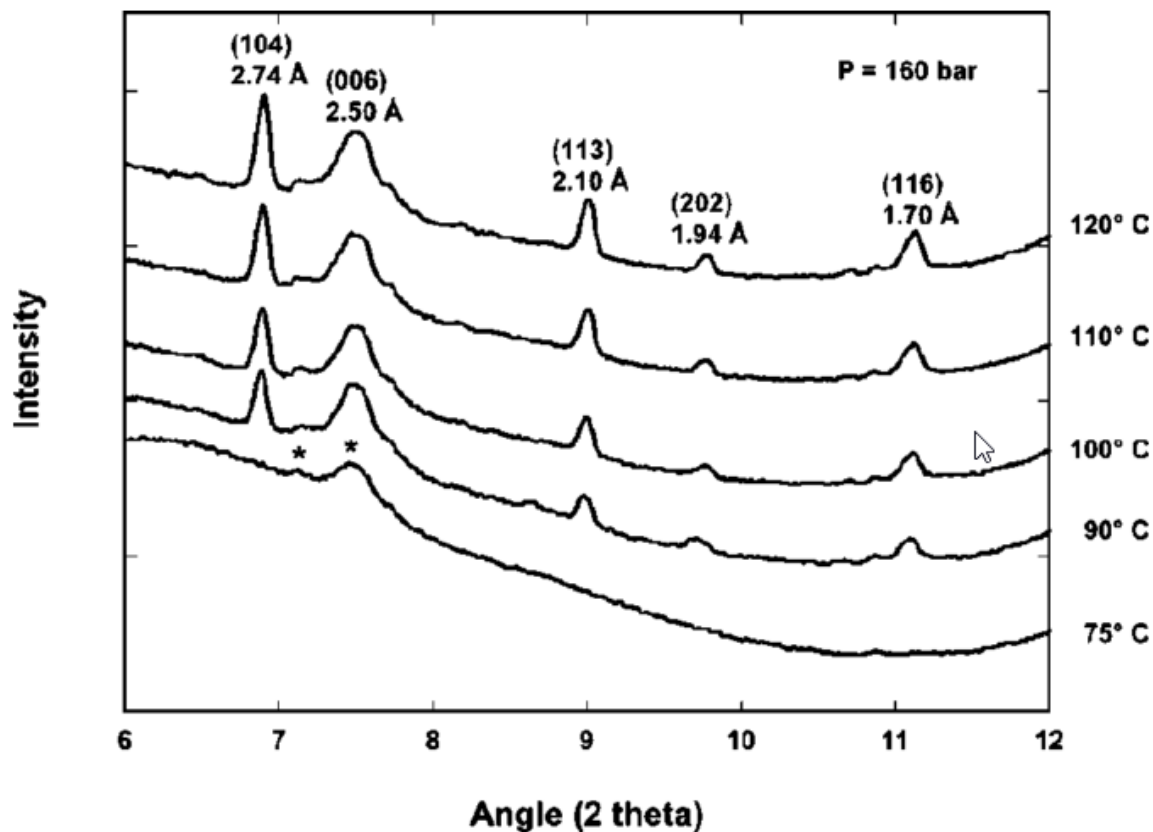
Externally controlled pressure and temperature microreactor for *in situ* x-ray diffraction, visual and spectroscopic reaction investigations under supercritical and subcritical conditions

Microreactor for pressure and temperature control

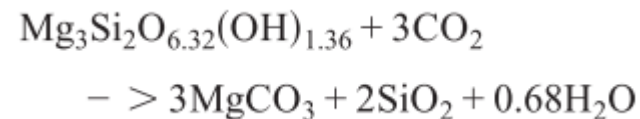
- In-Situ XRD and XAS experiments
- From ambient to up to 400 °C and 310 bar (external control)
- ✓ Structural studies
- ✓ *in situ* reaction processes



Externally controlled pressure and temperature microreactor for *in situ* x-ray diffraction, visual and spectroscopic reaction investigations under supercritical and subcritical conditions



Conversion of a meta-serpentine sample to magnesite under high pressure and temperature



Carbonation of serpentine:

- 100,000 years for nature
- 1h with this high P/T setup



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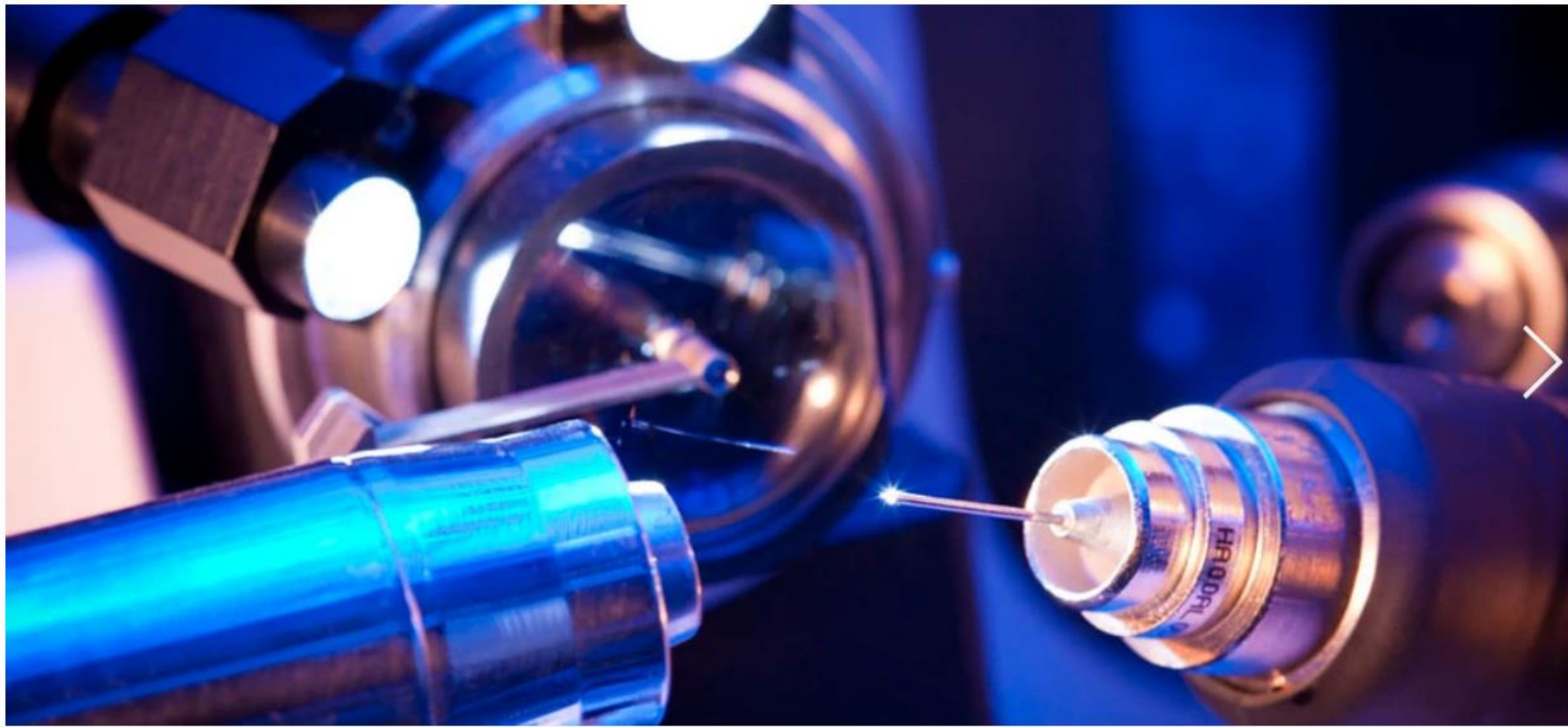
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Conclusions

- In-situ experiments allow to study the materials and components under working conditions
- They yield very important information about the processes, facilitating the improvement of the materials and devices
- Many options exist both for in-house diffractometers and synchrotrons
- Synchrotrons are very flexible and well suited for in-situ experiments.



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High energy diffraction for in-situ studies

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Small/wide angle X-ray scattering (SAXS/WAXS)

High pressure crystallography

X-ray reflectivity

Grazing incidence diffraction (GID)

Anomalous diffraction and magnetic diffraction

All 3 beamlines are now part of the general user program!



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Acknowledgments

