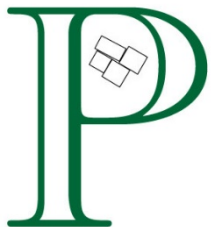


# An Introduction to ICDD and the Powder Diffraction File

James A. Kaduk

Poly Crystallography, Inc., Naperville IL 60540

[kaduk@polycrystallography.com](mailto:kaduk@polycrystallography.com)



NORTH CENTRAL  
COLLEGE 1861

ILLINOIS INSTITUTE  
OF TECHNOLOGY



# History

- Hanawalt, J. D. & Rinn, H. W. (1936). *Identification of crystalline materials. Ind. Eng. Chem. Anal. Ed. 8*, 244-247, reprinted in (1986) *Powder Diffraction 1*, 2-6.
- Hanawalt, J. D., Rinn, H. W. & Frevel, L. K. (1938). *Chemical analysis by X-ray diffraction. Ind. Eng. Chem. Anal. Ed. 10*, 457-512, reprinted in part in (1986) *Powder Diffraction 1*, 7-14.
- 1941 - Joint Committee for Chemical Analysis by Powder Diffraction Methods, supported by Committee E-4 of the American Society for Testing and Materials (ASTM)
- 1969 - independent non-profit Joint Committee on Powder Diffraction Standards (JCPDS).
- 1978 - name change to the International Centre for Diffraction Data

# Products and Services

- PDF-2 (5 year) – 316,820 entries (2021)
- PDF-4+ (annual) – 444,133 (2021)
- PDF-4 Organics – 547,295 (2021)
- WebPDF-4, PDF-4 Minerals (48,946), PDF-4/Axiom (97,789)
- Sieve/Sieve+
- *Powder Diffraction, Advances in X-ray Analysis*
- Denver X-ray Conference, Pharmaceutical Powder Diffraction Symposium
- Clinics and Workshops
- Grant-in-Aid

Initializing & Loading Main Form ...

License Acquired (Server: This PC)

C:\MyFiles\Clinic\2019\ramm136.raw

$\lambda = 1.54059 \text{ \AA}$

Cu

285 Days



Materials Data

JADE

Subfile ▾

Environment

- Ambient  
 Non-ambient  
 Temp.  
 Press.  
 Temp. & Press.

Status

- Primary  
 Alternate  
 Deleted

Quality Mark

- Star  
 Good  
 Indexed  
 Calculated  
 Prototyping  
 Minimal Acceptable  
 Blank  
 Low-Precision  
 Hypothetical

Database

- ICDD (00)  
 ICSD (01)  
 CSD (02)  
 NIST (03)  
 LPF (04)  
 ICDD Crystal Data (05)

- Custom PDF Set  
 Alkaloid  
 Amino Acid, Peptide & Complex  
 Battery Material  
 Bioactive
  - No Subclass
  - Depressant
  - Narcotic
  - Pesticide & Antimicrobial
  - Psychotropic
  - Stimulant

Periodic Table	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Formula/Name	1 <b>H</b> 1.008																	2 <b>He</b> 4.003
Classifications			<input checked="" type="radio"/> Boolean <input type="radio"/> Yes/No/Maybe <input type="radio"/> Composition Diagram List <input type="radio"/> And <input type="radio"/> Or <input type="checkbox"/> Not    Grouping: <input type="checkbox"/> Only <input type="radio"/> Just <input type="radio"/> And <input type="radio"/> Or															
Crystallography	3 <b>Li</b> 6.941	4 <b>Be</b> 9.012											5 <b>B</b> 10.811	6 <b>C</b> 12.01	7 <b>N</b> 14.007	8 <b>O</b> 15.999	9 <b>F</b> 18.998	10 <b>Ne</b> 20.180
Modulated													13 <b>Al</b> 26.982	14 <b>Si</b> 28.086	15 <b>P</b> 30.974	16 <b>S</b> 32.065	17 <b>Cl</b> 35.453	18 <b>Ar</b> 39.948
Diffraction	11 <b>Na</b> 22.990	12 <b>Mg</b> 24.305											31 <b>Ga</b> 69.723	32 <b>Ge</b> 72.64	33 <b>As</b> 74.922	34 <b>Se</b> 78.96	35 <b>Br</b> 79.904	36 <b>Kr</b> 83.798
Physical Properties	19 <b>K</b> 39.098	20 <b>Ca</b> 40.078	21 <b>Sc</b> 44.956	22 <b>Ti</b> 47.867	23 <b>V</b> 50.941	24 <b>Cr</b> 51.996	25 <b>Mn</b> 54.938	26 <b>Fe</b> 55.845	27 <b>Co</b> 58.993	28 <b>Ni</b> 58.693	29 <b>Cu</b> 63.546	30 <b>Zn</b> 65.409	31 <b>Ga</b> 69.723	32 <b>Ge</b> 72.64	33 <b>As</b> 74.922	34 <b>Se</b> 78.96	35 <b>Br</b> 79.904	36 <b>Kr</b> 83.798
Reference	37 <b>Rb</b> 85.468	38 <b>Sr</b> 87.62	39 <b>Y</b> 88.906	40 <b>Zr</b> 91.224	41 <b>Nb</b> 92.906	42 <b>Mo</b> 95.94	43 <b>Tc</b> [98]	44 <b>Ru</b> 101.07	45 <b>Rh</b> 102.906	46 <b>Pd</b> 106.42	47 <b>Ag</b> 107.868	48 <b>Cd</b> 112.41	49 <b>In</b> 114.818	50 <b>Sn</b> 118.71	51 <b>Sb</b> 121.76	52 <b>Te</b> 127.6	53 <b>I</b> 126.904	54 <b>Xe</b> 131.293
Comments	55 <b>Cs</b> 132.905	56 <b>Ba</b> 137.327		72 <b>Hf</b> 178.49	73 <b>Ta</b> 180.948	74 <b>W</b> 183.84	75 <b>Re</b> 186.207	76 <b>Os</b> 190.23	77 <b>Ir</b> 192.217	78 <b>Pt</b> 195.078	79 <b>Au</b> 196.967	80 <b>Hg</b> 200.59	81 <b>Tl</b> 204.383	82 <b>Pb</b> 207.2	83 <b>Bi</b> 208.98	84 <b>Po</b> [209]	85 <b>At</b> [210]	86 <b>Rn</b> [222]
	87 <b>Fr</b> [223]	88 <b>Ra</b> [226]		104 <b>Rf</b> [261]	105 <b>Db</b> [262]	106 <b>Sg</b> [266]	107 <b>Bh</b> [264]	108 <b>Hs</b> [277]	109 <b>Mt</b> [268]	110 <b>Ds</b> [271]	111 <b>Rg</b> [272]	112 <b>Cn</b> [285]	113 <b>Nh</b> [286]	114 <b>Fl</b> [289]	115 <b>Mc</b> [289]	116 <b>Lv</b> [293]	117 <b>Ts</b> [294]	118 <b>Og</b> [294]
La			57 <b>La</b> 138.906	58 <b>Ce</b> 140.116	59 <b>Pr</b> 140.908	60 <b>Nd</b> 144.242	61 <b>Pm</b> [145]	62 <b>Sm</b> 150.36	63 <b>Eu</b> 151.964	64 <b>Gd</b> 157.25	65 <b>Tb</b> 158.925	66 <b>Dy</b> 162.5	67 <b>Ho</b> 164.93	68 <b>Er</b> 167.259	69 <b>Tm</b> 168.934	70 <b>Yb</b> 173.04	71 <b>Lu</b> 174.967	
Ac			89 <b>Ac</b> [227]	90 <b>Th</b> 232.038	91 <b>Pa</b> 231.036	92 <b>U</b> 238.029	93 <b>Np</b> [237]	94 <b>Pu</b> [244]	95 <b>Am</b> [243]	96 <b>Cm</b> [247]	97 <b>Bk</b> [247]	98 <b>Cf</b> [251]	99 <b>Es</b> [252]	100 <b>Fm</b> [257]	101 <b>Md</b> [258]	102 <b>No</b> [259]	103 <b>Lr</b> [262]	

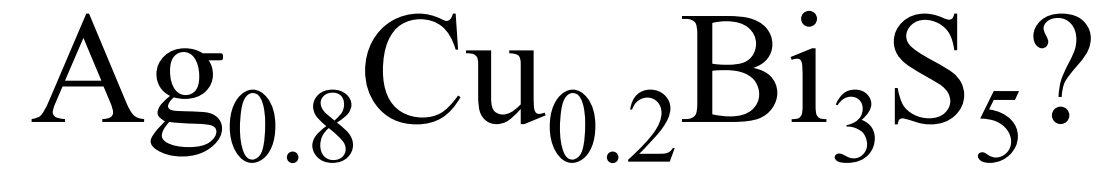
Search

Reset Tab

Reset All

Help

Range Input ▾ Global Operator ▾



Ag, Cu, Bi, S only

Subfile ▾

Environment









- Ambient
- Non-ambient
- Temp.
- Press.
- Temp. & Press.

- Atomic Coordinates 
- Raw Diffraction Data 

Status

- Primary
- Alternate
- Deleted

Quality Mark

-  Star
-  Good
-  Indexed
-  Calculated
-  Prototyping
-  Minimal Acceptable
-  Blank
-  Low-Precision
- Hypothetical


Database

- ICDD (00)
- ICSD (01)
- CSD (02)
- NIST (03)
- LPF (04)
- ICDD Crystal Data (05)

- Custom PDF Set
- Alkaloid
- Amino Acid, Peptide & Complex
- Battery Material
- Bioactive
  - No Subclass
  - Depressant
  - Narcotic
  - Pesticide & Antimicrobial
  - Psychotropic
  - Stimulant

Periodic Table

Formula ▾

Any Formula 

Formula/Name

Name ▾

Any Name 

Classifications

Crystallography

IMA No.

Modulated

CAS Number

Diffraction

Number of Elements

Physical Properties

Reference

Low

High

Comments

Composition ▾ 

Element

Weight %

± Error



Search

Reset Tab

Reset All

Help

Range Input ▾ Global Operator ▾

[Only (Cu And Ag And Bi And S)] And [Status (Primary, Alternate)]

Search

# Herbertsmithite

Name + Ambient + Coordinates



Search Subfile ▼

Custom PDF Set

- Alkaloid
- Amino Acid, Peptide & Complex
- Battery Material
- Bioactive
  - No Subclass
  - Depressant
  - Narcotic
  - Pesticide & Antimicrobial
  - Psychotropic
  - Stimulant

Environment


Ambient


Non-ambient

Temp.

Press.

Temp. & Press.

Atomic Coordinates 

Raw Diffraction Data 

Status

Primary

Alternate

Deleted

Quality Mark

- Star
- Good
- Indexed
- Calculated
- Prototyping
- Minimal Acceptable
- Blank
- Low-Precision
- Hypothetical

Database

ICDD (00)

ICSD (01)

CSD (02)

NIST (03)

LPF (04)

ICDD Crystal Data (05)

Periodic Table

Formula/Name

Classifications

Crystallography

Modulated

Diffraction

Physical Properties

Reference

Comments

Mineral Classification ▼

- ⊕ AEN - Aenigmatite (group)
- ⊕ ALC - Allactite (group)
- ⊕ VRL - Alluaudite (group)
- ⊕ ALM - Alum (group)
- ⊕ ALN - Alunite (supergroup)
- ⊕ AMB - Amblygonite (group)
- ⊕ AMP - Amphibole (family)
- ⊕ ANC - Analcime (supergroup)
- ⊕ ANY - Ancyrite (supergroup)
- ⊕ ADA - Andalusite (group)
- ⊕ ANT - Antlerite (group)
- ⊕ APA - Apatite (group)
- ⊕ APH - Aphthitalite (supergroup)
- ⊕ APO - Apophyllite (supergroup)
- ⊕ ARA - Aragonite (group)
- ⊕ ARC - Arcanite (group)

Zeolite Classification ▼

- ABW - Li-A(BW)
- ACO - ACP-1
- AEI - AIPO4-18
- AEL - AIPO4-11
- AEN - AIPO-EN3
- AET - AIPO4-8
- AFG - Afghanite
- AFI - AIPO4-5
- AFN - AIPO-14
- AFO - AIPO4-41
- AFR - SAPO-40
- AFS - MAPSO-46
- AFT - AIPO4-52
- AFX - SAPO-56
- AFY - CoAPO-50
- AHT - AIPO4-H2

Organic Functional Group ▼

- >4\_Hetero\_atoms\_in\_ring(s)
- >5\_fused\_rings
- >9\_membered\_ring
- 1\_Hetero\_atom\_in\_ring(s)
- 1,2\_dione\_\_\_\_O=C-C=O
- 2\_fused\_rings
- 2\_Hetero\_atoms\_in\_ring(s)
- 3\_fused\_rings
- 3\_Hetero\_atoms\_in\_ring(s)
- 3\_membered\_ring
- 4\_fused\_rings
- 4\_Hetero\_atoms\_in\_ring(s)
- 4\_membered\_ring
- 5\_fused\_rings
- 5\_membered\_ring
- 6\_membered\_rin

Pearson Symbol ▼

With Hydrogen  ...

Prototype Structure ▼

Any Prototype Structure  ...

Formula Type (ANX) ▼

...

Search

Reset Tab

**Reset All**

Help

Range Input ▼ Global Operator ▼

Zeolite MFI

Subfile ▼

Environment

- Ambient  
 Non-ambient  
 Temp.  
 Press.  
 Temp. & Press.

Status

- Primary  
 Alternate  
 Deleted

Quality Mark

- Star  
 Good  
 Indexed  
 Calculated  
 Prototyping  
 Minimal Acceptable  
 Blank  
 Low-Precision  
 Hypothetical

Database

- ICDD (00)  
 ICSD (01)  
 CSD (02)  
 NIST (03)  
 LPF (04)  
 ICDD Crystal Data (05)

- Custom PDF Set
- Alkaloid
  - Amino Acid, Peptide & Complex
  - Battery Material
  - Bioactive
    - No Subclass
    - Depressant
    - Narcotic
    - Pesticide & Antimicrobial
    - Psychotropic
    - Stimulant

Periodic Table

Crystal System

- Triclinic (Anorthic)  Rhombohedral  
 Monoclinic  Hexagonal  
 Orthorhombic  Cubic

Formula/Name

Classifications

Crystallography

Tetragonal

Crystal (Symmetry Allowed) ▼

- Centrosymmetric  
 Non-centrosymmetric:  
 Enantiomorphic  Pyro / Piezo (p)  
 Optical Activity  Piezo (2nd Harm.)

Atomic Environment Type

Symbol Elements

- | Symbol | Elements |
|--------|----------|
| 1#a    | Ac       |
| 2#a    | Ag       |
| 2#b    | Al       |
| 3#a    | Am       |
| 3#b    | Ar       |

Modulated

Diffraction

Space Group ▼

Space Group Number ...

Physical Properties

Reference

Comments

Crystal Data

Axis (Å)

a: Value ± Error  
 b: Value ± Error  
 c: Value ± Error

Axial Ratio

α: Value ± Error  
 β: Value ± Error  
 γ: Value ± Error

Volume

Value ± Error

Search

Reset Tab

Reset All

Help

Range Input ▼ Global Operator ▼

Search

Author's Unit Cell

*P*-orthorhombic

$$a = 9.623, b = 10.565, c = 9.219$$

Change esd on volume to  $10 \text{ \AA}^3$

Subfile ▼

- Custom PDF Set
  - Alkaloid
  - Amino Acid, Peptide & Complex
  - Battery Material
  - Bioactive
    - No Subclass
    - Depressant
    - Narcotic
    - Pesticide & Antimicrobial
    - Psychotropic
    - Stimulant

Environment

- Ambient
- Non-ambient
- Temp.
- Press.
- Temp. & Press.

Status

- Primary
- Alternate
- Deleted

Quality Mark

- Star
- Good
- Indexed
- Calculated
- Prototyping
- Minimal Acceptable
- Blank
- Low-Precision
- Hypothetical

Database

- ICDD (00)
- ICSD (01)
- CSD (02)
- NIST (03)
- LPF (04)
- ICDD Crystal Data (05)

Periodic Table

Modulated Dimension

- 3 + 1d  3 + 2d  3 + 3d

Formula/Name

Subsystems

Classifications

- Modulated Structure (One Subsystem)  Composite Structure (Multiple Subsystems)

Crystallography

Modulated

Diffraction

Physical Properties

Reference

Comments

Search

Reset Tab

Reset All

Help

Range Input ▼ Global Operator ▼

Search

X-ray Diffraction
Wavelength: Custom ...  $\lambda$ : 0.709318
 Neutron Diffraction

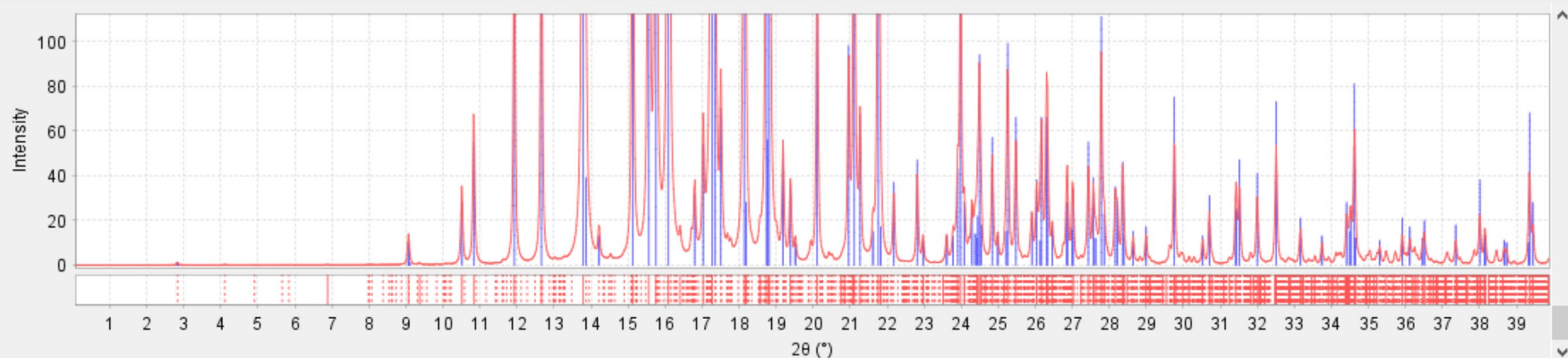
 Electron Diffraction

 Simulated Profile (Calc)

 Raw Diffraction Data

Fixed Slit Intensity

$2\theta$ (°)	$d$ (Å)	$I$	$h$	$k$	$l$	$m$	*
2.82640	14.38051	1	0	0	1		-2
9.06742	4.48677	10	1	1	0		0
9.08515	4.47803	1	1	1	0		-2
10.50935	3.87255	27	-2	0	1		0
10.83208	3.75750	55	-1	1	1		0
11.92629	3.41384	133m	0	0	3		-4
11.92629	3.41384	133m	1	1	1		0
12.65959	3.21683	164	2	0	1		0



PDF	Status:	Primary	Quality Mark:	Star	
Experimental	Environment:	Ambient	Temperature:	293.0 K	Pressure: -
Physical	Phase:	-			
Crystal	Chemical Formula:	$\text{Na}_{3.919}(\text{C O}_3)_2$			
Structure	Structural Formula:	-			
Classifications	Empirical Formula:	$\text{C}_2\text{Na}_{3.919}\text{O}_6$			
Cross-references	Weight %:	C11.43 Na42.88 O45.69			
References	Atomic %:	C16.78 Na32.88 O50.34			
Comments	Compound Name:	Sodium Carbonate			
	Mineral Name:	Natrite   IMA No: 1981-005			
	Alternate Name:	-			
	CAS Number:	-			
	Entry Date:	09/01/2014			
	Modification Date:	09/01/2015   Modifications: Dx			

Unit Cell:   

Maximum Combined Bonding Radius:

 Show Atom Labels Show Polyhedra

0%

25%

50%

75%

100%

125%

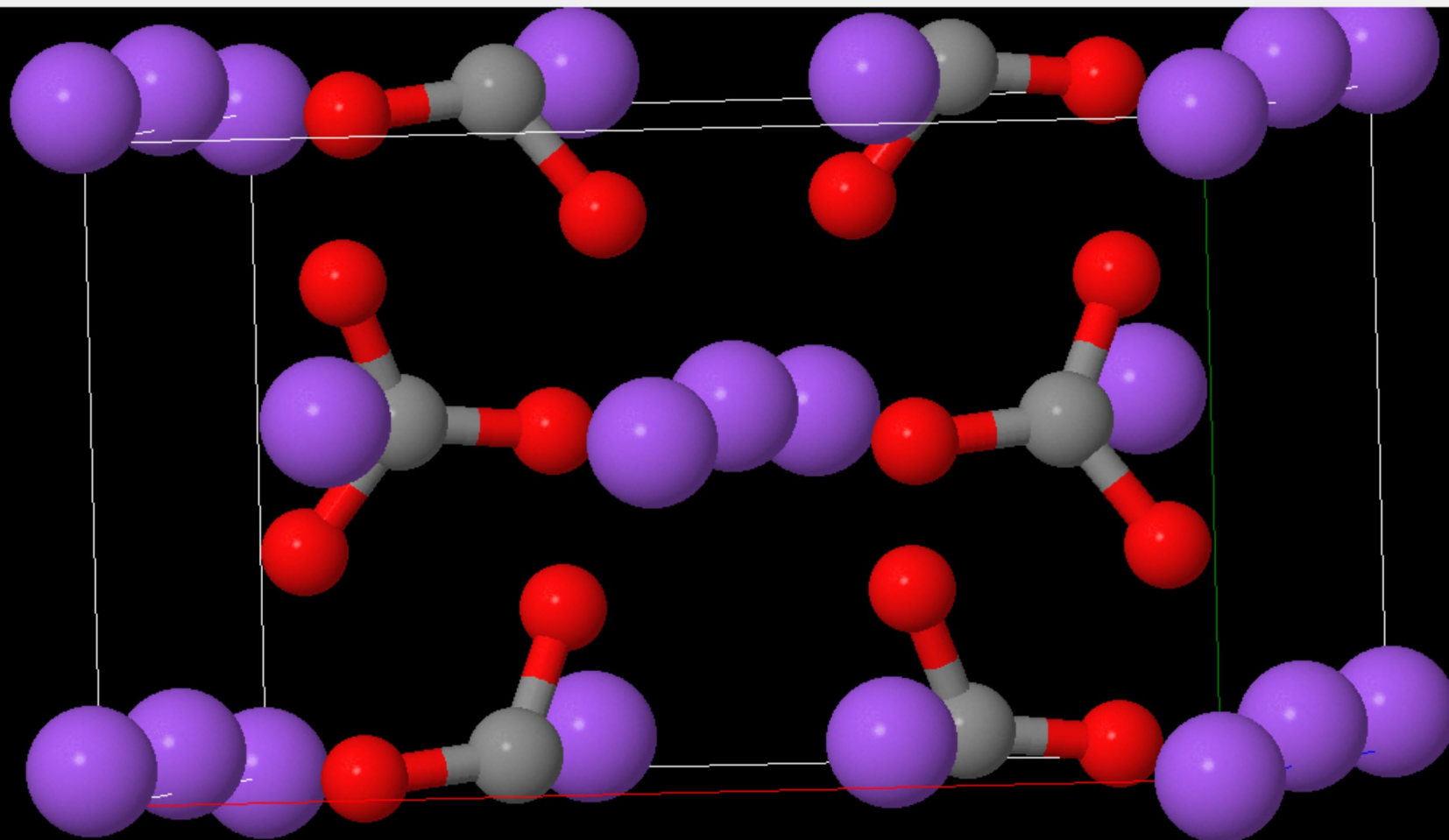
150%

Subsystem:

Toggle Modulation

Toggle Animation

SSG:  $C2/m(\alpha0\gamma)0s$   
 $q1=0.1837\ 0.0\ 0.31116$   
 $a=8.851\text{\AA}$   
 $b=5.240\text{\AA}$   
 $c=6.021\text{\AA}$   
 $\alpha=90.000^\circ$   
 $\beta=101.080^\circ$   
 $\gamma=90.000^\circ$



Subfile ▼

Environment

Status

Quality Mark

Database



- Custom PDF Set
  - Alkaloid
  - Amino Acid, Peptide & Complex
  - Battery Material
  - Bioactive
    - No Subclass
    - Depressant
    - Narcotic
    - Pesticide & Antimicrobial
    - Psychotropic
    - Stimulant

- Ambient
- Non-ambient
  - Temp.
  - Press.
  - Temp. & Press.

- Primary
- Alternate
- Deleted

- Star
- Good
- Indexed
- Calculated
- Prototyping
- Minimal Acceptable
- Blank
- Low-Precision
- Hypothetical

- ICDD (00)
- ICSD (01)
- CSD (02)
- NIST (03)
- LPF (04)
- ICDD Crystal Data (05)

- Atomic Coordinates 
- Raw Diffraction Data 

Periodic Table

Formula/Name

Strong Line (Å) ▼

Value ± Error

Classifications

 D1  D2  D3

Crystallography

Reported Intensity

 Integrated Intensities  Peak Intensities

Modulated

Diffraction

I/I-corundum ▼

Value ± Error

Physical Properties

Reference

R-factor ▼

Value ± Error

Comments

Smith-Snyder Figure of Merit ▼

Value ± Error

Temperature of Data Collection (K) ▼

Value ± Error

Radiation:  X-ray/Electron  CW Neutron

Long Line (Å) ▼

Value ± Error

 L1  L2  L3

Pressure of Data Collection (GPa) ▼

Value ± Error

Search

Reset Tab

Reset All

Help

Range Input ▼ Global Operator ▼

Search



Subfile ▾

Environment

- Ambient
- Non-ambient
- Temp.
- Press.
- Temp. & Press.

Status

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Periodic Table

Melting Point (K) ▾

Value  ± Error 

Formula/Name

Density (g/cm<sup>3</sup>) ▾Value  ± Error 

Classifications

Crystallography

 Measured Density
  Calculated Density
  Structural Density

Modulated

Color ▾

- Black
- Blue
- Brown
- Color Missing
- Colorless
- Gray
- Green
- Metallic
- Orange
- Pink
- Red
- Violet
- White

Diffraction

Physical Properties

Reference

Comments

 Property Sheet  Topology Data

Search

Reset Tab

Reset All

Help

Range Input ▾ Global Operator ▾

Subfile ▾

Environment

Status

Quality Mark

Database



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- LPF (04)
- ICDD Crystal Data (05)

- Atomic Coordinates 
- Raw Diffraction Data 

 Search All References  Search Primary Reference

Periodic Table

Formula/Name

DOI

Classifications

Title ▾

Crystallography

Modulated

Author ▾

Diffraction

Publication ▾

Physical Properties

Reference

Volume ▾

Comments

Year

Search

Reset Tab

Reset All

Help

Range Input ▾ Global Operator ▾

Search your own surname

Subfile ▾

- Custom PDF Set
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- ICDD Crystal Data (05)

Periodic Table

Database Comments ▾

Formula/Name

Classifications

Crystallography

Modulated

Diffraction

Physical Properties

Reference

Comments

Absolute Configuration

Additional Diffraction Lines

Additional Patterns

Analysis

ANX

Atomic Position

Bioactivity

Boiling Point

Calculated Pattern Original Remarks

Search

Reset Tab

Reset All

Help

Range Input ▾ Global Operator ▾

Search

ICSD Collection Code 88605

Element/Oxide Weight % +  
 Mg x 3 x x

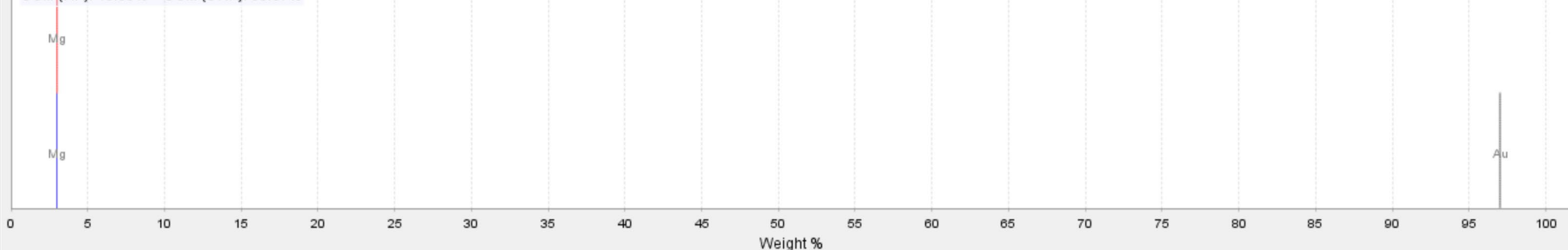
± Error (%): 5.0 f Set Filter

Ignore Elements: None [Status (Primary, Alternate)]

Matches (8,602 of 397,421)

GOM (WA) ↓	GOM (UWA)	PDF #	QM	Weight %	Compound Name	Mineral Name
49.83%	99.67%	00-019-0519	● B	Au97.01 Mg2.99	Gold Magnesium	
49.83%	99.67%	04-001-3003	● P	Au97.01 Mg2.99	Gold Magnesium	
49.83%	99.67%	04-004-4093	● P	Au97.01 Mg2.99	Gold Magnesium	
46.88%	93.75%	00-056-1208	● B	Mg3.20 W96.80	Magnesium Tungsten	
45.17%	90.33%	04-017-8497	● I	Ag97.29 Mg2.71	Magnesium Silver	
42.98%	85.96%	01-074-5964	● I	Mg3.49 Sn96.51	Magnesium Tin	
42.98%	85.96%	04-003-2216	● I	Mg3.49 Sn96.51	Magnesium Tin	
40.00%	80.00%	03-065-6701	● I	Au96.25 Mg3.75	Gold Magnesium	
39.79%	79.58%	00-020-0456	● I	Au96.23 Mg3.77	Gold Magnesium	
39.79%	79.58%	04-007-2004	● I	Au96.23 Mg3.77	Gold Magnesium	
39.17%	78.33%	01-071-6824	● I	Cd97.65 Mg2.35	Cadmium Magnesium	
38.33%	76.67%	04-001-0474	● I	In97.70 Mg2.30	Indium Magnesium	
37.97%	75.95%	00-019-0518	● I	Au96.05 Mg3.95	Gold Magnesium	
37.97%	75.95%	00-024-0461	● I	Au96.05 Mg3.95	Gold Magnesium	
37.97%	75.95%	01-072-5352	● I	Au96.05 Mg3.95	Gold Magnesium	
37.97%	75.95%	04-007-1482	● B	Au96.05 Mg3.95	Gold Magnesium	
37.97%	75.95%	04-007-1483	● B	Au96.05 Mg3.95	Gold Magnesium	
37.97%	75.95%	04-019-4954	● I	Au96.05 Mg3.95	Gold Magnesium	

GOM (WA): 49.83% • GOM (UWA): 99.67%



Search Reset Paste

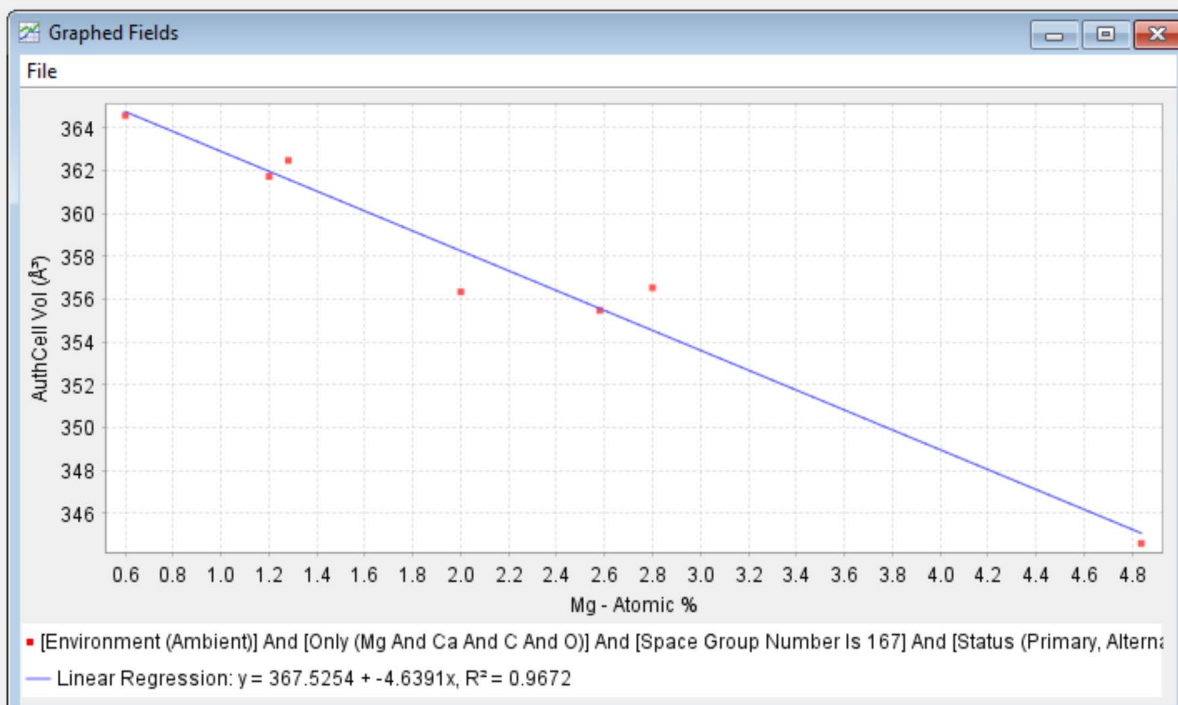
# Boolean Searches

Ca,Mg,C,O only and space group #167 and ambient

PDF #	QM	Chemical Formula	Compound Name	Mineral Name	D1 (Å)	D2 (Å)	D3 (Å)	SYS	Auth SPGR	XtCell a (Å)	XtCell c (Å)	AuthCell Vol (Å <sup>3</sup> )	L1 (Å)
★ 00-043-0697	S	(Ca, Mg)CO <sub>3</sub>	Calcium Magnesium Carbonate	Calcite, magnesian	3.004200	1.889150	2.262510	R	R-3c	4.943	16.852	356.53	3.8204
00-060-0473	I	(Ca, Mg)(CO <sub>3</sub> )	Calcium Magnesium Carbonate		3.008920	1.860030	2.267010	R	R-3c	4.955	16.898	359.24	3.8240
01-086-2336	S	(Mg <sub>0.129</sub> Ca <sub>0.871</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian	2.999470	2.259940	1.853450	R	R-3c	4.938	16.832	355.47	3.8126
01-089-1304	S	(Mg <sub>0.03</sub> Ca <sub>0.97</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian, syn	3.025470	1.869360	1.904940	R	R-3c	4.978	16.988	364.57	3.8442
01-089-1305	S	(Mg <sub>0.06</sub> Ca <sub>0.94</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian, syn	3.018190	1.864700	1.901030	R	R-3c	4.963	16.957	361.72	3.8336
04-008-8067	B	Ca <sub>0.9</sub> Mg <sub>0.1</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate	Calcite, magnesian	3.002300	1.890090	1.855100	R	R-3c	4.941	16.854	356.34	3.8153
04-012-6929	S	Ca <sub>0.845</sub> Mg <sub>0.155</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate		3.025480	1.869370	2.278600	R	R-3c	4.978	16.988	364.57	3.8442
04-012-6930	S	Ca <sub>0.715</sub> Mg <sub>0.285</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate		3.018190	2.272170	1.864700	R	R-3c	4.963	16.957	361.72	3.8336
04-013-2116	S	Ca <sub>0.936</sub> Mg <sub>0.064</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate	Calcite, magnesian	3.020030	1.865910	1.901900	R	R-3c	4.967	16.963	362.47	3.8365
04-019-9173	P	Ca <sub>0.758</sub> Mg <sub>0.242</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate		2.965820	2.238700	1.833360	R	R-3c	4.896	16.600	344.60	3.7759



PDF #	QM	Chemical Formula	Compound Name	Mineral Name	D1 (Å)	D2 (Å)	D3 (Å)	SYS	Auth SPGR	XtCell a (Å)	XtCell c (Å)	AuthCell Vol (Å <sup>3</sup> )	L1 (Å)
00-043-0697	S	(Ca, Mg)CO <sub>3</sub>	Calcium Magnesium Carbonate	Calcite, magnesian	3.004200	1.889150	2.262510	R	R-3c	4.943	16.852	356.53	3.8204
01-086-2336	S	(Mg <sub>0.129</sub> Ca <sub>0.871</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian	2.999470	2.259940	1.853450	R	R-3c	4.938	16.832	355.47	3.8126
01-089-1304	S	(Mg <sub>0.03</sub> Ca <sub>0.97</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian, syn	3.025470	1.869360	1.904940	R	R-3c	4.978	16.988	364.57	3.8442
01-089-1305	S	(Mg <sub>0.06</sub> Ca <sub>0.94</sub> )(CO <sub>3</sub> )	Magnesium Calcium Carbonate	Calcite, magnesian, syn	3.018190	1.864700	1.901030	R	R-3c	4.963	16.957	361.72	3.8336
04-008-8067	B	Ca <sub>0.9</sub> Mg <sub>0.1</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate	Calcite, magnesian	3.002300	1.890090	1.855100	R	R-3c	4.941	16.854	356.34	3.8153
04-013-2116	S	Ca <sub>0.936</sub> Mg <sub>0.064</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate	Calcite, magnesian	3.020030	1.865910	1.901900	R	R-3c	4.967	16.963	362.47	3.8365
04-019-9173	P	Ca <sub>0.758</sub> Mg <sub>0.242</sub> (CO <sub>3</sub> )	Calcium Magnesium Carbonate		2.965820	2.238700	1.833360	R	R-3c	4.896	16.600	344.60	3.7759



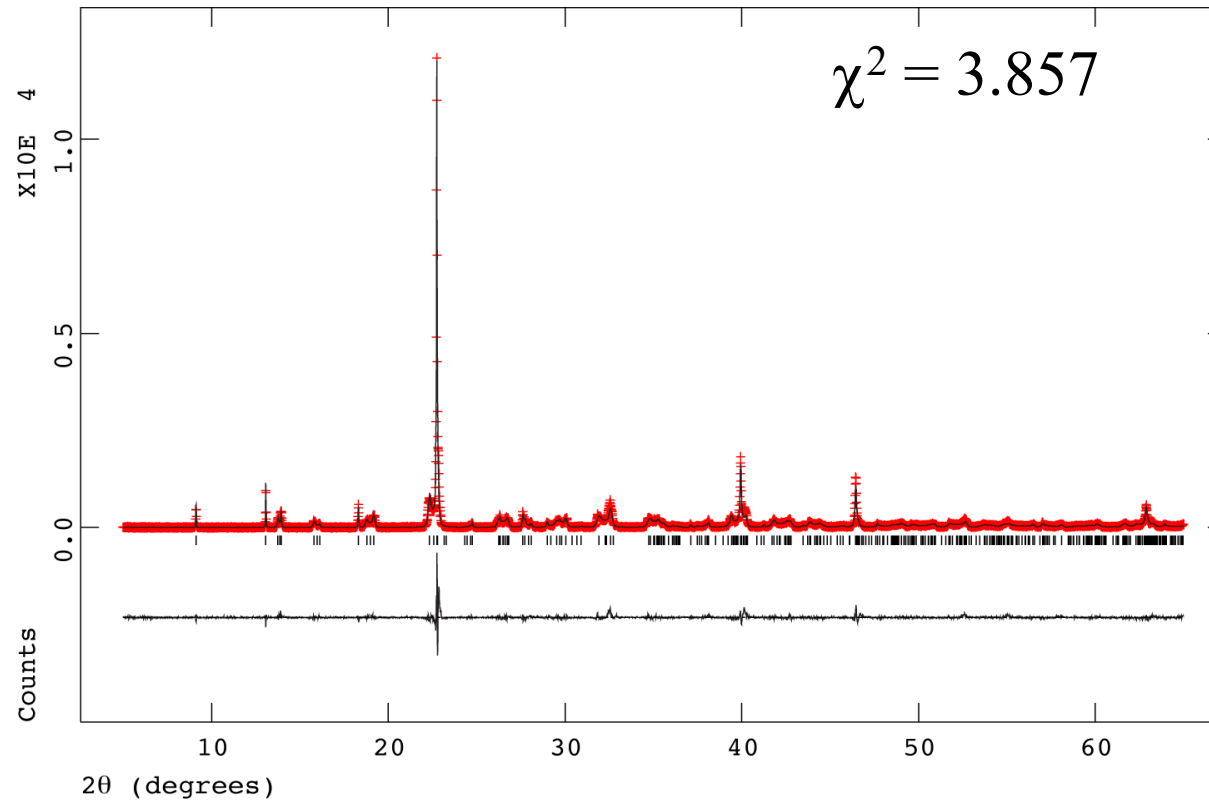
A Case Study:  
The 6H Perovskite  $\text{Ba}_3\text{CaSb}_2\text{O}_9$

Chris Ling

The University of Sydney

# Data from 1-BM-C

$C2/c$ ,  $a = 5.99898(7)$ ,  $b = 10.37797(19)$ ,  $c = 14.8658(3)$  Å,  
 $\beta = 91.384(2)^\circ$ ,  $V = 925.23(3)$  Å<sup>3</sup>



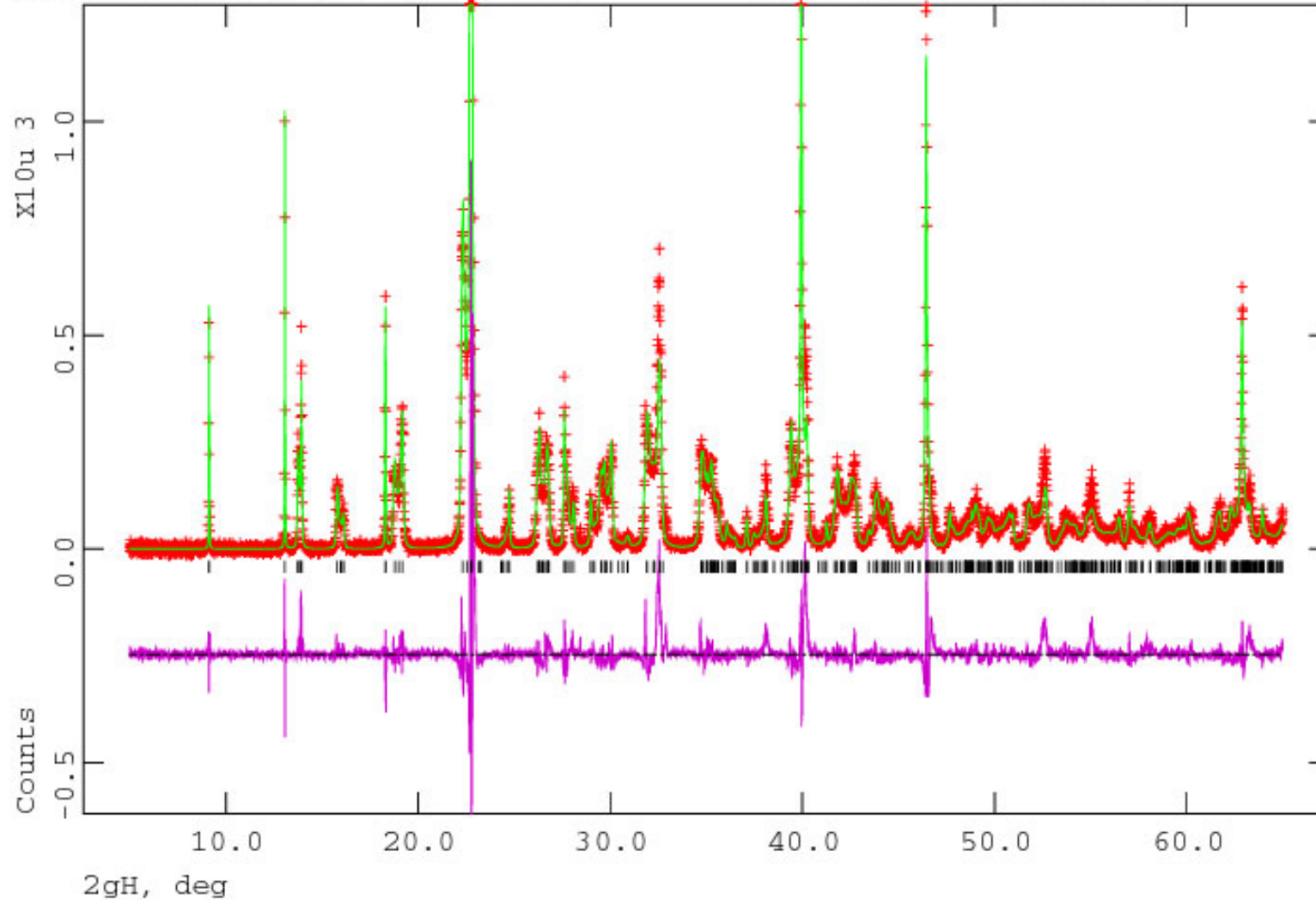
# Look more closely:

Ba3 Ca Sb2 O9 (KD5019\_1/Ling)

Hist 1

Lambda 1.1823 A, L-S cycle 122

Obsd. and Diff. Profiles



There are peaks unaccounted for:

$2\theta$	$d$ ( $\lambda = 1.182279 \text{ \AA}$ )	Intensity
22.87	2.9817	400
28.05	2.4392	70
28.44	2.4065	60
31.63	2.1619	150
32.51	2.1118	300
40.14	1.7226	260
46.70	1.4915	140
52.58	1.3346	110
55.04	1.2794	80

Import 1D Diffraction Pattern     Accept Phase    Search Method: Hanawalt     Weight d-Spacings     Set Filter  
 Import 2D Diffraction Pattern     Remove Last Phase    Search Window (°): 0.3     2nd Pass Filter    [Status (Primary, Alternate)]  
 New d-I List     Complete Phases    Match Window (°): 0.3    Min GOM: 2000  
 Insert PDF     Internal Standard  
 Create Graph     Intensity Ratios

GOM	PDF #	Peaks Matched	Chemical Formula	Compound Name	Mineral Name	Phase	Status	QM	Coords	I/Ic	D1 (Å)	D2 (Å)	D3 (Å)	D4 (Å)	D5 (Å)
7214	04-012-0403	8 of 43 (19%)	Ba <sub>2</sub> Cu <sub>0.10</sub> PrRu <sub>0.90</sub> O <sub>6</sub>	Barium Copper Praseodymium Ruthenium Oxide			P	I	✓	5.91	2.981660	2.108900	1.724360	2.113470	1.490830
7158	04-012-0402	8 of 50 (16%)	Ba <sub>2</sub> PrRuO <sub>6</sub>	Barium Praseodymium Ruthenium Oxide			A	I	✓	5.98	2.981660	2.989740	2.108930	1.724400	2.113490
7012	01-084-9262	8 of 29 (28%)	(Ba <sub>1.60</sub> Ca <sub>0.40</sub> )LaNbO <sub>6</sub>	Barium Calcium Lanthanum Niobium Oxide			P	I		5.38	2.980870	2.989250	2.108190	1.724000	2.113320
6870	04-022-2891	8 of 25 (32%)	Ba <sub>1.6</sub> Ca <sub>0.4</sub> LaEu <sub>x</sub> NbO <sub>6</sub>	Barium Calcium Europium Lanthanum Niobium Oxide			P	P	✓	5.15	2.980890	2.989270	2.108190	1.724360	2.113340
6768	04-002-4363	7 of 14 (50%)	Ba <sub>2</sub> Y Nb O <sub>6</sub>	Barium Yttrium Niobium Oxide			A	P	✓	15.39	2.983990	1.722810	2.110000	1.492000	1.334480
6768	04-002-6579	7 of 14 (50%)	Ba <sub>2</sub> Y Zr O <sub>6</sub>	Barium Yttrium Zirconium Oxide			P	P	✓	15.19	2.983990	1.722810	2.110000	1.492000	1.334480
6754	04-021-7855	7 of 14 (50%)	Ba <sub>2</sub> Ca <sub>0.79</sub> Y <sub>0.13</sub> Nb <sub>1.08</sub> O <sub>5.68</sub>	Barium Calcium Yttrium Niobium Oxide			P	I	✓	14.1	2.982860	1.722150	2.109200	1.491430	1.333980
6720	04-005-7520	7 of 14 (50%)	Ba <sub>2</sub> Y Nb O <sub>6</sub>	Barium Yttrium Niobium Oxide			A	P	✓	15.39	2.982580	1.721990	2.109000	1.491290	1.333850
6720	04-007-0162	7 of 14 (50%)	Ba <sub>2</sub> Y Nb O <sub>6</sub>	Barium Yttrium Niobium Oxide			A	P	✓	15.39	2.982580	1.721990	2.109000	1.491290	1.333850
6705	04-015-2511	7 of 9 (78%)	BaY <sub>0.2</sub> Zr <sub>0.8</sub> O <sub>2.9</sub>	Barium Yttrium Zirconium Oxide			P	I	✓	14.98	2.984400	1.723040	2.110290	1.492200	1.334660
6704	04-015-4802	7 of 9 (78%)	BaY <sub>0.2</sub> Zr <sub>0.8</sub> O <sub>2.9</sub>	Barium Yttrium Zirconium Oxide			A	I	✓	14.98	2.984400	1.723040	2.110290	1.492200	1.334660

Diffraction Patterns Phases Peaks (7 of 9)



Import 1D Diffraction Pattern     Accept Phase    Search Method:      Weight d-Spacings   

Import 2D Diffraction Pattern     Remove Last Phase    Search Window (°):      2nd Pass Filter    [Status (Primary, Alternate)]

New d-I List     Complete Phases    Match Window (°):     Min GOM:

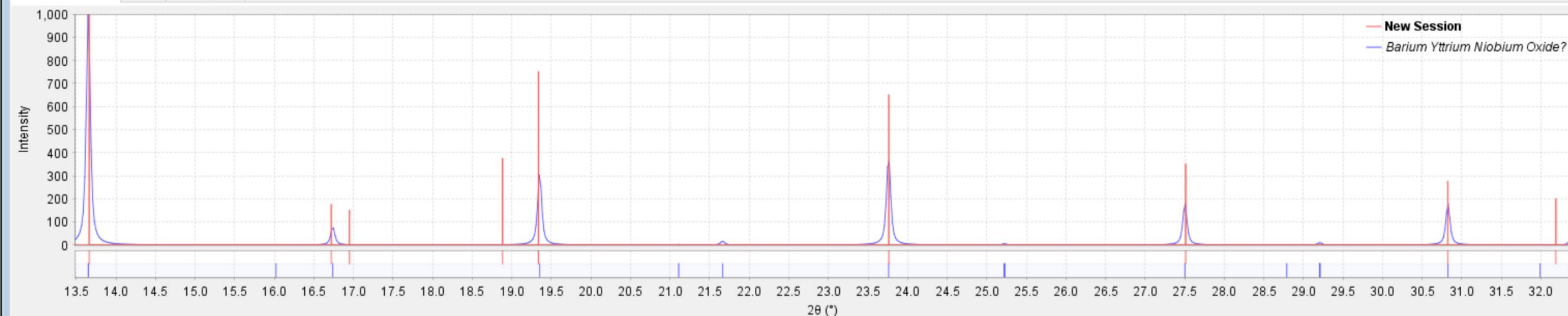
    Internal Standard

    Intensity Ratios ▼

File                      Action                      Matches (1,919 of 397,421)                      Table                      Corrections

GOM	PDF #	Peaks Matched	Chemical Formula	Compound Name	Mineral Name	Phase	Status	QM	Coords	I/Ic	LS1 (Å)	LS2 (Å)	LS3 (Å)	LS4 (Å)	LS5 (Å)
7318	04-012-0403	8 of 43 (19%)	Ba <sub>2</sub> Cu <sub>0.10</sub> PrRu <sub>0.90</sub> O <sub>6</sub>	Barium Copper Praseodymium Ruthenium Oxide			P	I	✓	5.91	2.981660	2.113470	2.108900	1.724360	1.494850
7184	04-018-0963	7 of 24 (29%)	Ba <sub>2</sub> YTaO <sub>6</sub>	Barium Yttrium Tantalum Oxide		Low tem...	A	B	✓	11.67	2.982230	2.112070	2.105450	1.722690	1.719990
6868	01-084-9262	8 of 29 (28%)	(Ba <sub>1.60</sub> Ca <sub>0.40</sub> )LaNbO <sub>6</sub>	Barium Calcium Lanthanum Niobium Oxide			P	I		5.38	2.989250	2.980870	2.113320	2.108190	1.724000
6854	04-022-2891	8 of 25 (32%)	Ba <sub>1.6</sub> Ca <sub>0.4</sub> LaEu <sub>x</sub> NbO <sub>6</sub>	Barium Calcium Europium Lanthanum Niobium Oxide			P	P	✓	5.15	2.989270	2.980890	2.113340	2.108190	1.724360
6845	04-012-0402	8 of 50 (16%)	Ba <sub>2</sub> PrRuO <sub>6</sub>	Barium Praseodymium Ruthenium Oxide			A	I	✓	5.98	2.989740	2.981660	2.113490	2.108930	1.724400
6796	04-009-4989	8 of 32 (25%)	BaTl <sub>0.6</sub> Sb <sub>0.4</sub> O <sub>2.9</sub>	Barium Thallium Antimony Oxide			P	P	✓	13.98	2.982660	2.970000	2.112750	2.105390	1.722940
6601	04-002-4363	7 of 14 (50%)	Ba <sub>2</sub> YNbO <sub>6</sub>	Barium Yttrium Niobium Oxide			A	P	✓	15.39	2.983990	2.436420	2.110000	1.722810	1.492000
6601	04-002-6579	7 of 14 (50%)	Ba <sub>2</sub> YZrO <sub>6</sub>	Barium Yttrium Zirconium Oxide			P	P	✓	15.19	2.983990	2.436420	2.110000	1.722810	1.492000
6601	04-006-0345	7 of 9 (78%)	BaY <sub>0.6</sub> Nb <sub>0.2</sub> W <sub>0.2</sub> O <sub>3</sub>	Barium Yttrium Niobium Tungsten Oxide			P	I	✓	15.96	2.983990	2.436420	2.110000	1.722810	1.492000
6595	04-014-0985	7 of 14 (50%)	Ba <sub>2</sub> YNbO <sub>6</sub>	Barium Yttrium Niobium Oxide			A	I	✓	15.31	2.984380	2.436740	2.110280	1.723030	1.492190
6595	04-015-2511	7 of 14 (50%)	Ba <sub>2</sub> YZrO <sub>6</sub>	Barium Yttrium Zirconium Oxide			P	I	✓	14.08	2.984400	2.436750	2.110300	1.723040	1.492200

Diffraction Patterns    Phases    Peaks (7 of 9)



A weak peak at  $28.44^\circ$  ( $2.4065 \text{ \AA}$ )  
is still unaccounted for:

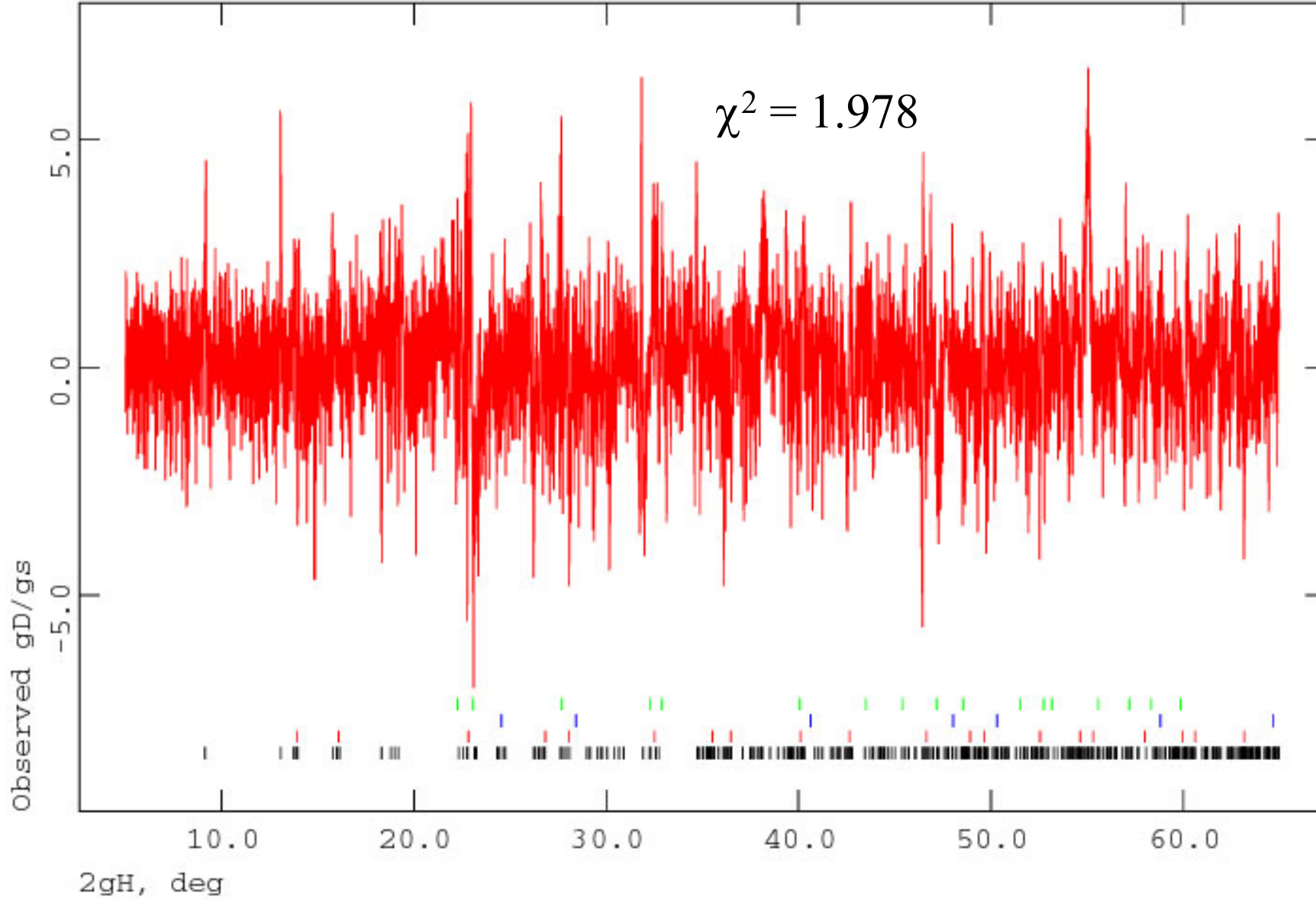


PDF #	QM	Chemical Formula	Compound Name	Mineral Name	D1 (Å)	D2 (Å)	D3 (Å)	SYS	Auth SPGR	XtCell a (Å)	XtCell c (Å)	AuthCell Vol (Å³)	L1 (Å)
00-004-0777	I	Ca O	Calcium Oxide	Lime, syn	2.405000	1.701000	2.778000	C	Fm-3m	4.811	4.811	111.32	2.7780
00-037-1497	S	Ca O	Calcium Oxide	Lime, syn	2.405870	1.700930	2.777370	C	Fm-3m	4.811	4.811	111.33	2.7773
00-048-1467	C	Ca O	Calcium Oxide	Lime, syn	2.405060	1.700810	2.777710	C	Fm-3m	4.811	4.811	111.33	2.7777
01-070-5490	S	Ca O	Calcium Oxide	Lime, syn	2.407600	1.702430	2.780060	C	Fm-3m	4.815	4.815	111.65	2.7800
01-077-2010	I	Ca O	Calcium Oxide	Lime, syn	2.406000	1.701300	2.778210	C	Fm-3m	4.812	4.812	111.42	2.7782
01-077-9574	S	Ca O	Calcium Oxide	Lime, syn	2.405360	1.700850	2.777470	C	Fm-3m	4.811	4.811	111.33	2.7774
01-080-7710	S	Ca O	Calcium Oxide		2.408550	1.703100	2.781150	C	Fm-3m	4.817	4.817	111.78	2.7811
01-082-1690	S	Ca O	Calcium Oxide	Lime, syn	2.402450	1.698790	2.774110	C	Fm-3m	4.805	4.805	110.93	2.7741
04-002-6758	I	Ca O	Calcium Oxide	Lime, syn	2.405000	1.700590	2.777050	C	Fm-3m	4.810	4.810	111.28	2.7770
04-003-7161	I	Ca O	Calcium Oxide	Lime, syn	2.404000	1.699880	2.775900	C	Fm-3m	4.808	4.808	111.15	2.7759
04-004-5528	I	Ca O	Calcium Oxide		2.401600	1.698190	2.773130	C	Fm-3m	4.803	4.803	110.81	2.7731
04-004-8549	I	Ba <sub>2</sub> Sb <sub>3</sub>	Antimony Barium		3.095000	2.414010	3.170170	M	P21/c	15.817	6.934	1475.44	10.2476
04-004-8985	I	Ca O	Calcium Oxide	Lime, syn	2.405600	1.701020	2.777750	C	Fm-3m	4.811	4.811	111.37	2.7777
04-005-4757	I	Ca O	Calcium Oxide	Lime, syn	2.401000	1.697760	2.772440	C	Fm-3m	4.802	4.802	110.73	2.7724
04-005-6351	I	Ca O	Calcium Oxide	Lime, syn	2.407500	1.702360	2.779940	C	Fm-3m	4.815	4.815	111.63	2.7799
04-005-9402	I	Ca O	Calcium Oxide	Lime, syn	2.407500	1.702360	2.779940	C	Fm-3m	4.815	4.815	111.63	2.7799
04-006-5940	H	Ca O	Calcium Oxide	Lime, syn	2.410000	1.704130	2.782830	C	Fm-3m	4.820	4.820	111.98	2.7828
04-006-5942	I	Ca O	Calcium Oxide	Lime, syn	2.406000	1.701300	2.778210	C	Fm-3m	4.812	4.812	111.42	2.7782
04-006-6519	I	Ca O	Calcium Oxide	Lime, syn	2.400000	1.697060	2.771280	C	Fm-3m	4.800	4.800	110.59	2.7712
04-006-9375	I	Ca O	Calcium Oxide	Lime, syn	2.405000	1.700590	2.777050	C	Fm-3m	4.810	4.810	111.28	2.7770
04-007-4743	I	Ca O	Calcium Oxide	Lime, syn	2.405290	1.700800	2.777400	C	Fm-3m	4.811	4.811	111.33	2.7774
04-007-8376	I	Ca O	Calcium Oxide	Lime, syn	2.405000	1.700590	2.777050	C	Fm-3m	4.810	4.810	111.28	2.7770
04-007-9734	I	Ca O	Calcium Oxide	Lime, syn	2.409500	1.703770	2.782250	C	Fm-3m	4.819	4.819	111.91	2.7822
04-010-5778	I	Ca O	Calcium Oxide	Lime, syn	2.407600	1.702430	2.780060	C	Fm-3m	4.815	4.815	111.65	2.7800
04-011-8430	I	Ca O	Calcium Oxide	Lime, syn	2.404000	1.699880	2.775900	C	Fm-3m	4.808	4.808	111.15	2.7759
04-011-9020	I	Ca O	Calcium Oxide	Lime, syn	2.404850	1.700490	2.776880	C	Fm-3m	4.810	4.810	111.26	2.7768

There's still a very weak peak  
at  $32.89^\circ/2.0881 \text{ \AA}$ :

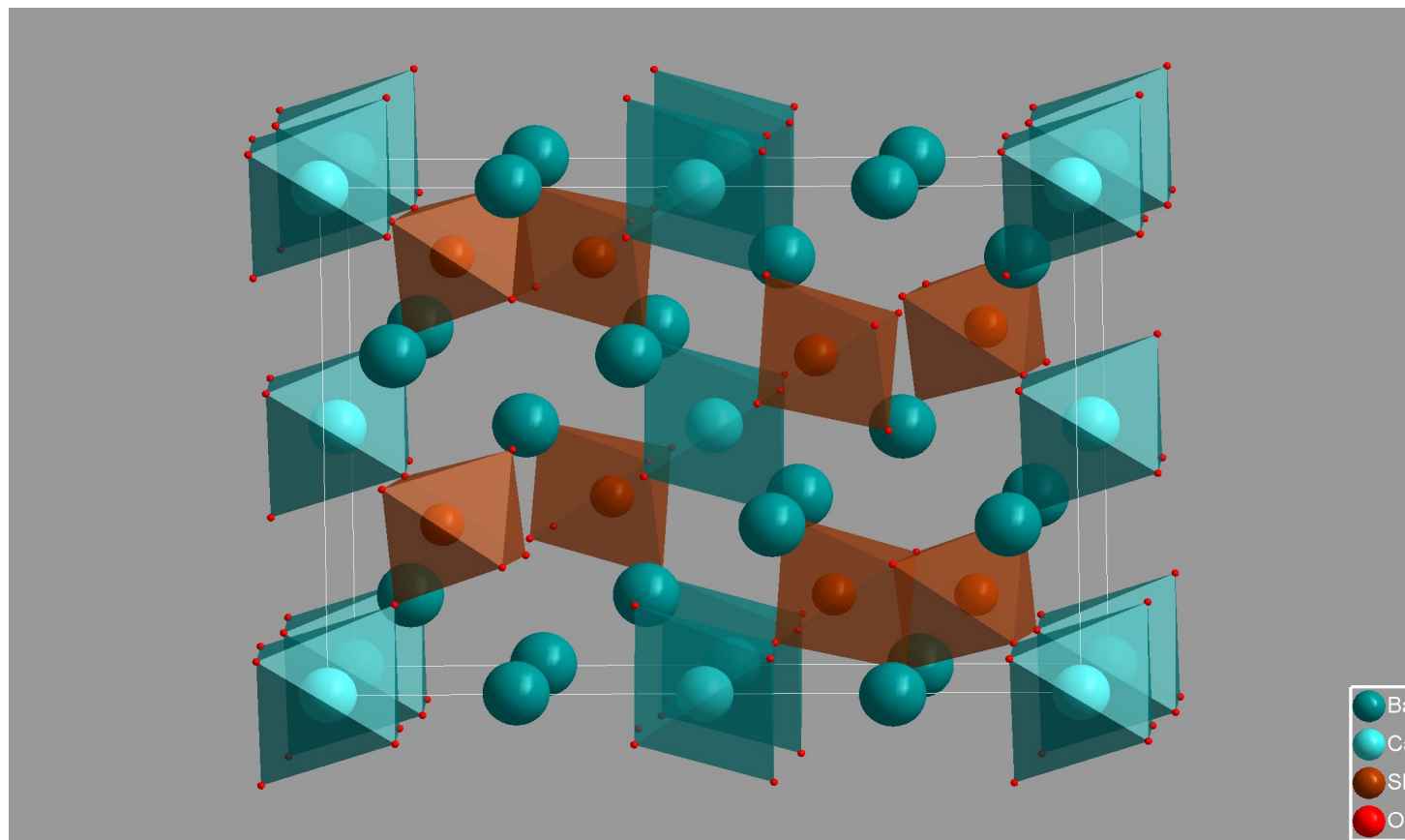
PDF #	QM	Chemical Formula	Compound Name	Mineral Name	D1 (Å)	D2 (Å)	D3 (Å)	SYS	Auth SPGR	XtCell a (Å)	XtCell c (Å)	AuthCell Vol (Å <sup>3</sup> )	L1 (Å)
00-017-0125	B	Sb	Antimony		2.940000	2.090000	1.710000	C	Pm-3m	2.960	2.960	25.93	2.9400
00-020-0209	O	Ca <sub>3</sub> Sb <sub>2</sub> O <sub>6</sub>	Calcium Antimony Oxide		2.970000	2.880000	2.080000	X					4.4200
01-083-8080	I	O <sub>8</sub>	Oxygen		2.158210	2.087170	3.271880	M	C2/m	6.947	3.656	139.90	4.3164
01-083-8081	S	O <sub>8</sub>	Oxygen		2.157840	2.085380	3.275620	M	C2/m	6.963	3.657	140.01	4.3156
04-002-2786	I	Ba O	Barium Oxide		2.084070	2.899140	2.050000	T	P4/nmm	4.100	2.998	50.40	2.9980
04-007-9029	S	Sb	Antimony		2.987500	2.099000	2.126230	R	R-3m	4.198	10.485	160.02	3.4950
04-007-9030	S	Sb	Antimony		2.983760	2.098000	2.121870	R	R-3m	4.196	10.455	159.41	3.4850
04-013-2320	I	(O <sub>2</sub> )	Oxygen		2.158210	2.087170	3.271880	M	C2/m	6.947	3.656	139.90	4.3164
04-013-2321	S	(O <sub>2</sub> )	Oxygen		2.157840	2.085380	3.275620	M	C2/m	6.963	3.657	140.01	4.3156
04-017-1162	P	Sb	Antimony		2.929190	2.060000	2.082690	R	R-3m	4.120	10.260	150.82	3.4200

[Just (Ca And Ba And Sb And O)] And [Strong Line = 2.09(0.01) Å] And [Status (Primary, Alternate)]



# Quantitative Phase Analysis

Phase	Concentration, wt%
$\text{Ba}_3\text{CaSb}_2\text{O}_9$	88.59(3)
$\text{Ba}_2\text{CaSbO}_6$	10.59(9)
CaO	0.53(3)
BaO?	0.28(2)
other	trace



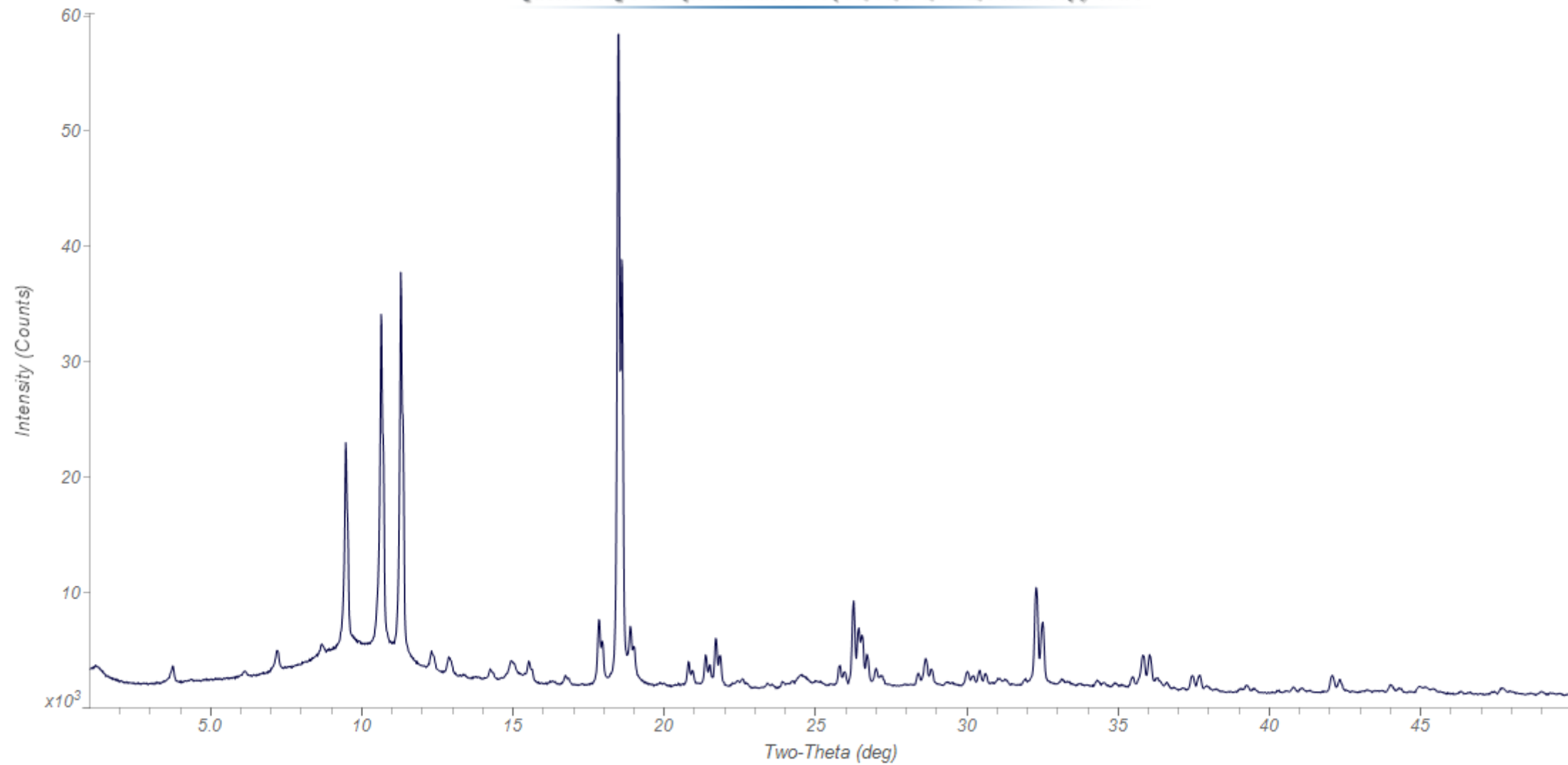
# Use of Sleve+ and JADE

“prelithiated Si”

ashu003.xrdml



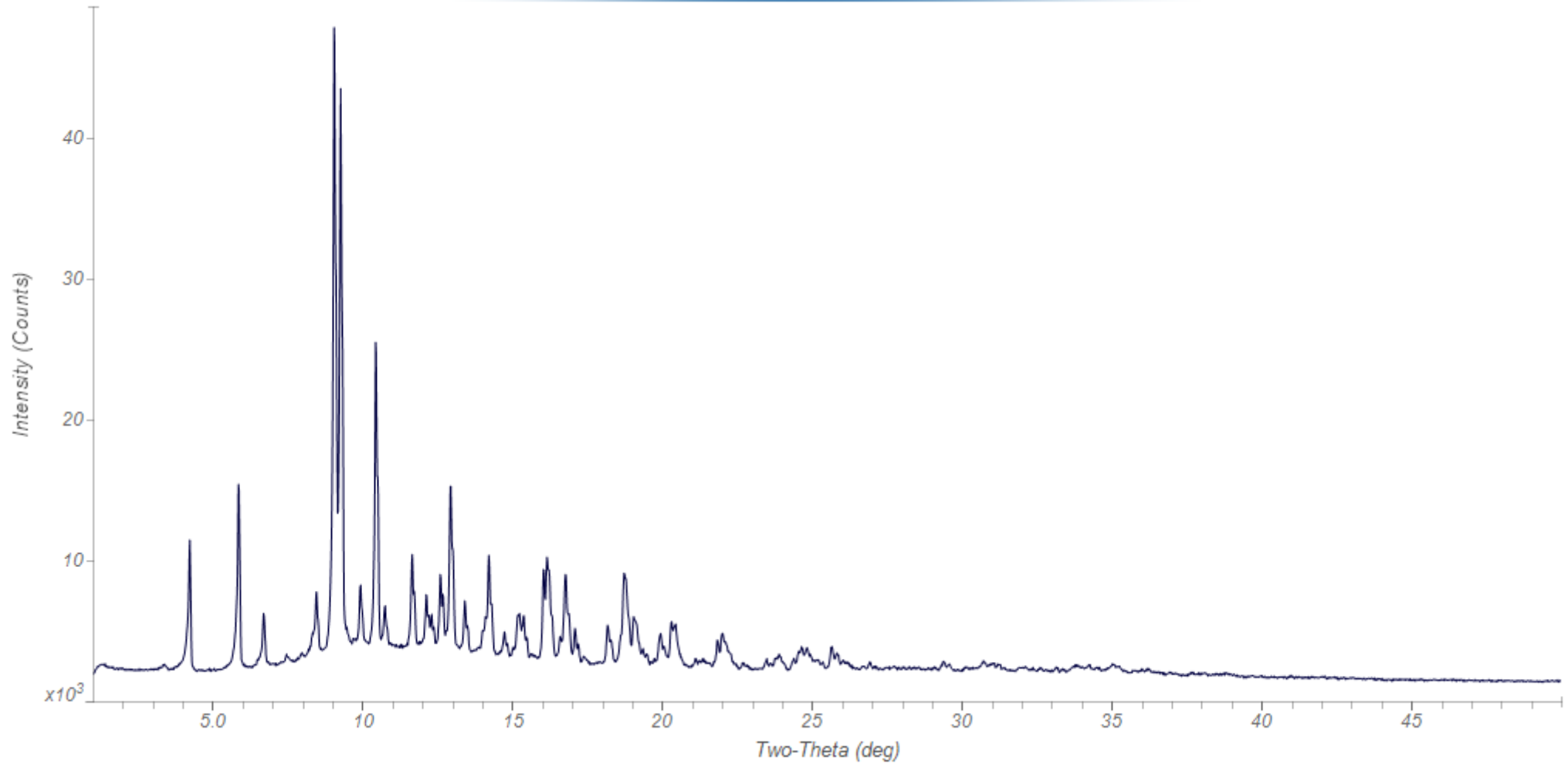
[ashu003] PSi - prelithiated Si (60,40,1/4,0.02,1 mm cap) JAK



Express flower food

kadu1852.xrdml

**[kadu1852] IExpress flower food (60,40,1/4,0.02,0.7 mm cap) JAK**



Supplied with PDF-4

“Sessions”

# Inorganic Phases

- **Exercise 7-1-1.** Believed to be a pure phase, but not sure that the label on the reagent bottle is correct.
- **Exercise 7-1-2.** A student tried to prepare  $\text{SrTiO}_3$  using reagent  $\text{SrO}$ , but the reaction did not go as planned. Check the  $\text{SrO}$ .
- **Exercise 7-1-3.** Check the product of a preparation of  $\text{CdTiO}_3$  to see if it is phase pure.

# Inorganic Phases

- **Exercise 7-1-4.** A mixture of three common oxides.
- **Exercise 7-1-5.** A smear mount of a natural specimen of witherite,  $\text{BaCO}_3$ .
- **Exercise 7-1-6.** A white paint.
- **Exercise 7-1-7.** A black filter deposit from a filter in a new solar hot water system. XRF indicates Fe, Ca, and minor Mg. An acid test indicates the presence of carbonate.

# Inorganic Phases

- **Exercise 7-1-8.** A sample of lime (prepared by heating calcite at  $1000^{\circ}\text{C}$ ) was left exposed to the air for several days. What is it now?
- **Exercise 7-1-9.** A synthetic red phosphor for color television tubes.
- **Exercise 7-1-10.** A fluorescent screen (which turns out to be a solid solution).

# Minerals

- **Exercise 7-2-1.** A natural specimen of a zinc mineral. Consider data quality of old PDF entries and associations.
- **Exercise 7-2-2.** A mixture of polymorphs.
- **Exercise 7-2-3.** A mixture prepared to simulate the pressure-induced changes which can be induced by severe grinding. The specimen was prepared without any special care to avoid preferred orientation.



# Minerals

- **Exercise 7-2-4.** An attempt to estimate the composition of a solid solution mineral.
- **Exercise 7-2-5.** A hydrothermal ore containing a trace of gold.
- **Exercise 7-2-6.** Typical raw material used to make dental ceramics (the best grade of whiteware porcelain). It is mined from a pegmatite deposit. We are asked to identify the feldspar(s) present.

# Minerals

- **Exercise 7-2-7.** A mixture of sulfide minerals.
- **Exercise 7-2-8.** A mineral with low-angle lines and an impurity.
- **Exercise 7-2-9.** The  $< 2\mu\text{m}$  fraction of a sedimentary rock (sandstone).

# Metals and Alloys

- **Exercise 7-3-1.** An artificial example to build confidence in identifying multi-phase unknowns
- **Exercise 7-3-2.** A steel
- **Exercise 7-3-3.** A corrosion product from a white metal bearing. XRF indicates the presence of Sn, with small amounts of Cu and Sb.

# Organic Phases

- **Exercise 7-4-1.** A tablet, ground to a fine powder and smeared on a zero-background holder.
- **Exercise 7-4-2.** A potassium-containing compound isolated from a winery.
- **Exercise 7-4-3.** The product of an organic synthesis which used both NaOH and HCl.
- **Exercise 7-4-4.** A complete unknown.

# Organic Phases

- **Exercise 7-4-5.** An over-the-counter medication.
- **Exercise 7-4-6.** A white powder found in an abandoned house.
- **Exercise 7-4-7.** An over-the-counter pharmaceutical.
- **Exercise 7-4-8.** A pharmaceutical which is known to contain sulfur.
- **Exercise 7-4-9.** A mixture which is known to contain CN, Cl, and COOH groups.
- **Exercise 7-4-10.** A prescription medication which is known to contain S by XRF.